Comparison of essential oils composition and antioxidative property of leaf extracts of Ocimum sanctum (p) and Ocimumbasilicum.

1.Dr.M.Rama, 2. Prof.B.SyamaSundar

 Dept. of Chemistry, Ch.S.D.St Theresa's College for women(A), Eluru,W.G AP mobile-9912107611 Mailid:manne_rama2001@yahoo.co.in
Former VC, Yogi Vemana University, Former Principal, Former HOD of Chemistry, Acharya Nagarjuna University, Guntur-522510.

Abstract

The present study deals with the total essential oil composition and antioxidant activity of two plants of lamiaceace family. Extraction of total essential oils and volatile essential oils from *Ocimum sanctum* (p) and *Ocimumbasilicum*in four different seasons of 2010 calendar year. Extraction of total essential oil content of plant materials was carried out by Soxhulet extraction and that of volatile oils by steam distillation using Clevenger type apparatus. The percentage composition of eugenol in four seasons was found from GC analysis. The antioxidant activity was studied by DPPH radical scavenging method. Total essential oil and volatile oil contents are more in winter(present study in the month of November) where as very less in summer(in the month of May). Eugenol is the major constituent present in these plants. Methanolic extract of leaves of *OcimumSanctum*(p) and*ocimumbasilicum*showed a very good DPPH radical scavenging activity and the percent of inhibition was 46.91% and 43.1% respectively. This study revealed that extract of leaves of Eugenol.

Key words: *Ocimumsanctum(p), ocimumbasilicum- total essential oil- eugenol-total phenols-flavonoids-antioxidants-DPPH assay*

1. INTRODUCTION

Ocimum sanctum also known as Ocimumtenuiflorumbelongs to Lamiaceae family is commonly known as Tulsior holybasil. The plant grows wild in India but is also widely cultivated in home and temple gardens and is used for household remediation [1]. Holy basil has a strong anise like, slightly musky and lemony taste with a camphoraceous aroma. The dominant aroma component in holy basil isEugenol. The more exuberantly flavoured red holy basil (redor purple variety) has dark green leaves withreddish purple stems and a purplish cast on theyounger leaves known as Krishna Tulsi.

*Ocimumbasilicum*L.is commonly known as *Sweet basil* is a pleasant smelling perennial shrub which grows in several regions all over the world [2,3]. *Sweet basil* is a popular culinary herb and a source of essential oils extracted by steam distillation from the leaves and the flowering tops which are used to flavour foods, in dental and oral products, and in fragrances.[4-9]The stem and leaves of holy basilcontain a variety of constituents that may havebiological activity, including saponins, flavonoids, triterpenoids, and tannins[10]. Theleaf volatile oil[11] contains eugenol, euginal (alsocalled eugenic acid), urosolic acid[12], carvacrol,linalool, limatrol, caryophyllene, methylcarvicol, uteolin, vitexin, isovitexin, orientin, isoorientin, aesculin, chlorogenic acid, aesculetin, caffeic acid[13], beta-carotene[14],Rosmarinic acid, apigenin, cirsimaritin, isothymusin and isothymonin. Two

watersoluble flavonoids:[15] Orientin and Vicenin, ascorbic acid and carotene are also reported[16].

Different parts of *Tulsi*plant such as leaves,flowers, stem, root, seeds etc. are knownto possess therapeutic potentials and have beenused by traditional medical practitioners, asexpectorant, analgesic, anti-cancer, antiasthmatic, antiemetic, diaphoretic, antidiabetic, anti-fertility,hepatoprotective, hypotensive, antistress, analgesic, antihyperlipidemic, antioxidantpotentials in experimental animals[17-22].

The characteristic compounds of *basil* essential oil are linalool, methyl chavicol, eugenol ,estragol ,thymol and *p*-cymen were found[7,23-26]. The phenolic compounds and flavonoids such as cinnamic acid, caffeic acid, sinapic acid, ferulic acid, rosemarinic acid, p-coumaric acid ,rutin ,kaempferol and Quercitin in *O. basilicum*L. have also been reported[27]. Two minor components of the essential oil of *sweet basil*, juvocimene I and II, have been reported as potent juvenile hormone analogs[28]. The leaves and

flowering tops of the plant are perceived as carminative, galactogogue and antispasmodic in folk medicine [29]Chicoric acid was recently identified and quantified in *basil* leaves[30]*Sweet basil* oil possess anticancer, antibacterial , toxic [24,31-39],insecticidal [38,40-42], repellent [43-46],nematicidal[47], fungistatic[35], or antimicrobial [48]activities. The allylphenolic derivative eugenol has a sharp spicy odour reminiscent ofcloves, and is used as a dental analgesic and disinfectant [49].

The objective of this research was to examine and compare the % composition of essential oil,% composition of eugenol, total

phenols, flavonoids, antioxidants and DPPH assay of purple variety of *O.sanctum* and *Ocimumbasilicum.*% composition of essential oil and % composition of eugenol of purple*Ocimum sanctum* and *Ocimumbasilicum*were compared in four seasons of 2010 calendar year

2. MATERIALS AND METHODS

2.1 Collection and identification of plantmaterials

All the selected plants were grown in the same geographical conditions in the spacious domestic home garden at Gavaravaram

village, Elurumandal in West Godavari district of Andhra Pradesh is situated between 16.7°North 81.1° East, elevation 22mts 72 feet. Plant material was collected and vouchers of specimens were deposited at the Botany department. Each specimen was labelled, numbered, annotated with the date of collection. In the present study the specimen numbers areM2,N2, M5 and N5. Each specimen was subjected for identification at plant systematic laboratory, Kakatiya University, Warangal, Andhra Pradesh,India.

One kilogram of leaves of each sample was collected, dried under shade, finely powdered in an electric blender (80 mesh) and stored in air tight containers at room temperature in the dark until used.

2.1.1 Preparation of plant extract

One gram of dried leaf powder was grinded with 20ml of 50% methanol and filtered. The filtrate was made up to the volume 50ml

with 50% methanol .This extract was used to analyze the total phenol content, the flavonoids content and the antioxidant capacity.

2.1.2 Soxhulet extraction

Extraction of total essential oil content of plant materials was carried out by soxhuletextraction[50]method. 5gm of dry powder was subjected tosoxhulet extraction with 250ml methanol as solvent, extraction was carried out for 3hrs, 10 cycles and temperature was

maintained at 65°C. This extract was used to analyze DPPH assay and the antimicrobial activity.

2.1.3 Steam distillation

Extraction of volatile oils from the plant materials was carried out by steam distillation using Clevenger type apparatus [51]. 100g powdered sample was water distilled by using a Clevenger oil arm fitted with condensers through which cooled water was circulated to

prevent low volatiles from escaping. The temperature was maintained at 60° C. The volatile oil was collected and dried over anhydrous Sodium Sulphate and stored at -4°C. 1mg of volatile extract was dissolved in 1ml ofmethanol; from that solution 10µl was taken and

made up to 100 μ l with methanol. This solution was used for GC analysis. (Same procedure was followed for the preparation of standard eugenol.)

2.2 Gas chromatograph analysis

The essential oils were analyzed using a Shimadzu gas chromatograph model 17 A Japan(2014)at Chandra laboratory, Hyderabad, equipped with flame ionization detector (FID) and DB-Wax capillary column (30mm x 0.32mm, film thickness 0. 5 µm). Injector and detector temperatures were set at 240 and 250° C, respectively. Column oven temperature was programmed from 40° C to 220° C at the rate of 8° C min-1; initial and final temperatures were held at 3 minutes and 10 minutes, respectively. Helium was used as a carrier gas with a flow of

1.5 mL/min. A sample of $0.1 \ \mu\text{L}$ was injected using slit mode (split ratio, 1:20). Quantification was completed by built-in data-handling

software supplied by the manufacturer (spin chrome CFR) of the gas chromatograph. Theresults (composition) were reported as a relative

percentage of the total peak area.

2.2.1 Estimation of total phenolics

Total phenolic content was determined by using Folin-Ciocalteu reagent as previously described [52]. Each plant extract (250 μ L) was

mixed with 5ml of Folin-Ciocalteu reagent and 4 ml of (20%) sodium carbonate, and they were vortexed for 50sec and they were let to stand for 30mins in water bath at 40° C. The optical density was measured by using systronics(C1117) colorimeter at 680nm. The total phenol

content of the extracts was obtained by using the standard curve. The total phenol content was expressed as tannic acid (0.1 mg/ml) equivalent in % w/w of the extracts. Total phenolic content =

optical density of sample×concentration of tannic acid 2.2.2 Total flavonoids content

The total flavonoid content was determined using the Dowd method [53]. Each plant extract solution (250 μ l) was mixed with 0.1ml of 2% aluminium chloride and 1ml of 0.1M potassium acetate, mixed well and allowed to stand for 30min. at room temperature. The optical density was measured by using systemics (C1117) colorimeter at 420nm. Totalflavonoid contents were calculated as quercetin

(0.1mg/ml) equivalent from a calibration curve.

Total flavonoids content =

optical density of sample × concentration of quercetin

optical density of standard

2.2.3 Reducing Power Assay

The reducing power of the extracts was measured by using ascorbic acid as previously described[54].Each plant extract solution (250 µl)

_

was mixed 2.5 ml of phosphate buffer (pH 6.6)and 2.5ml of 1%potassium ferricyanide and were incubated in water bath at 50°C for 20min. Then the test tube was centrifuged for 10min. at 10,000 rpm . 2.5ml of supernatant were diluted with 2.5 ml water and shaken with 0.5ml of freshly prepared 0.1 % Ferric chloride. The optical density was measured by using systronics (C1117) colorimeter at 680nm. Total antioxidants contentwas calculated as ascorbic acid (0.1mg/ml) equivalent from a calibration curve. Total

antioxidants content optical density of sample × concentration of ascorbic acid optical density of standard

2.2.4 DPPH radical scavenging assay (antioxidant assay)

The following assay procedure was modified from those described by Blois [55] and Govindaragan [56]. Sample stock solutions (1.0 mg/ml) were diluted to final concentrations of 250, 125,50, 25, 10 and 5 μ g/ml in methanol. A portion of sample solution (500 μ l) was mixed with an equal volume of 6x10-5 M DPPH (1,1-diphenyl-2 picryl-hydrazyl;in methanol) and allowed to stand at room temperature for 30 min. In each experiment methanol (1ml) plus plant extract

solution was used as blank while the DPPH solution alone in methanol was used as control. The absorbance (A) of sample solution was measured by using UV-VIS spectrophotometer 2.2. (double-beam) (SL191 series) at 520 nm, compared with that of control solution (maximum absorbance). The scavenging activity of samples corresponded to the intensity of quenching DPPH. The results were expressed aspercentage inhibition.

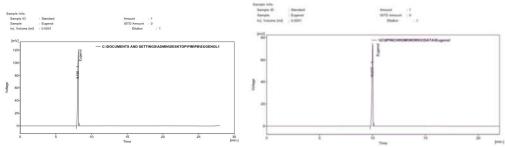
% inhibition = $\frac{[(A \text{ control} - A \text{ sample})]}{A \text{ control}} X 100$

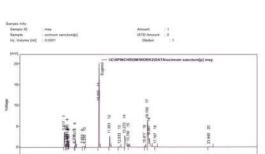
3. Result and discussion:

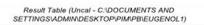
and O. basilicum								
Month & Year	Total essential oil content%		Percentage composition Eugenol					
	O.Sanctum(P)	O.Basilicum	O.Sanctum(P)	O.Basilicum				
February 2010	1.85%	1.53%	49.00%	NA				
May 2010	1.52%	1.28%	37.969%	6.553%				
August 2010	1.61%	1.41%	42.448%	7.747%				
November	2.0%	1.7%	61.538%	9.229%				
2010								

TABLE-1: Total essential content and Percentage composition of eugenol in *O. sanctum(p)* and O basilicum

Standard Eugenol chromatogram







Area [mV.s]

798.431

798.431

Height

[mV]

120.739

120.739

Area [%]

100.000

100.000

Reten. Time

[min]

Total

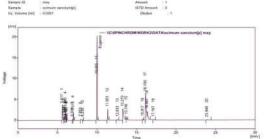
8.130

1

Result Table (Uncal -\\C\SPINCHROM\WORK2\DATA\Eugenol)

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%] 100.000	
1	10.017	464.988	74.656		
	Total	464.988	74.656	100.000	

1572 Amount B

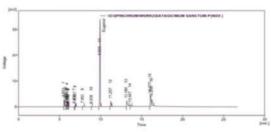




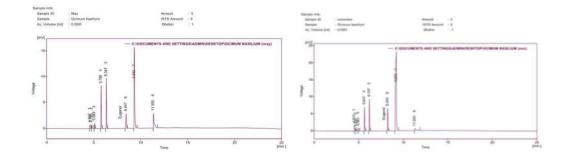
CONTRACTOR OF ICT.IN P

Result Table (Uncal -IICISPINCHROM/WORK2/DATAIOCIMUM SANCTUM-P(MAY)

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]
1	5.617	12.217	3.550	6.077
2	5.847	2.998	0.983	1.492
3	5.960	0.673	0.329	0.335
4	6.040	6.938	2.527	3.451
5	6.133	0.235	0.141	0.117
6	6.187	0.410	0.181	0.204
7	6.930	1.005	0.352	0.500
8	7.013	5.527	1.863	2.750
9	7.953	2.446	0.319	1.217
10	8.230	1.029	0.306	0.51
11	10.003	76.329	20.028	37.969
12	11.353	10.035	2.632	4.992
13	12.633	1.120	0.222	0.55
14	13.270	19.112	2.697	9.507
15	13.740	2.529	0.596	1.258
16	15.817	1.196	0.332	0.595
17	16.150	39.076	6.329	19.438
18	16.367	16.200	1,992	8.059
19	17.167	0.607	0.105	0.302
20	23.940	1.350	0.124	0.671
	Total	201.032	45.609	100.000



	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]
1	5.553	8.377	2.704	3.489
2	5.777	2.342	0.774	0.975
3	5.893	0.418	0.212	0.174
4	5.967	5.769	2.193	2.402
5	6.060	0.154	0.093	0.064
6	6.110	0.250	0.111	0.104
7	6.850	0.727	0.274	0.303
8	6.927	4.012	1.391	1.67
9	7.853	1.360	0.286	0.56
10	8.930	0.298	0.094	0.124
11	9.933	147.773	33.853	61.538
12	11.257	7.366	2.043	3.067
13	13.180	14.971	2.203	6.23
14.	13.647	2.109	0.485	0.87
15	16.077	26.262	4.194	10.937
16	16.250	17.946	2.494	7.473
	Total	240.135	53.404	100.000



Result	Table	(Uncal -	C: DOCUMEN	ITS	AND	
SETTINGS	ADMI	NDESK	TOP\OCIMUM	BA	SILIUM	(may

	Reten. Time [min]	Area [mV.s]	Height [mV]	Area [%]			le (Uncal - C:\DO MINDESKTOPIO	
1	4.607	1.971	0.734	1.325		Reten. Time	Area	Þ
2	4.793	1.911	0.717	1.284		[min]	[e.Vm]	- 1
3	5.083	2.088	1.057	1.403	1	4.513	4.210	
12	NE(6.7.12)	Contraction of the	11.5555.0	A CONTRACTOR OF A CONTRACTOR O	2	4.697	1.501	
-4	5.780	18.652	8.180	12.533	3	4.983	1.599	
5	6.347	23.834	9.629	16.015	4	5.647	14.834	
-6	8.447	9.753	2.816	6.553	5	6.197	22.571	
7	9.330	72.310	15.667	48.589	6	8.243	21.948	
8	11.350	18.302	2.817	12 298	7	9.203	165.952	
-	Total	148.822	41.617	100.000	8	11.293	5.195	
	Tural	140.022	41.017	100.000		Total	237.810	

6.238 7.039 9.49 6.590 9 225 23.204 69.783 40.011 100.000

3.1 Total essential oil and percentage composition of eugenol

Table 1 shows the yield of total essential oil and percentage composition of eugenol in Ocimumsanctum(p) and ocimumbasilicumin four different seasons of 2010 calendar year. It shows that the yield was affected by seasonal changes. The highest amount of the oil in these plants was found in winter i.e in November and very low in summer i.e in May. In Ocimumsanctum(p)the percentage composition of essential oil is high in November (2.0%) and low in May (1.52%), where as in Ocimumbasilicum the percentage composition of essential oil is high in November (1.7%) and low in May (1.28%). In India the temperature is very high in summer ranging 35-42°c. Low essential oil yield in summer might be attributed to the high temperature and partial evaporation of some constituents of oil can be expected.

From Table 1 it was found that the percentage composition of Eugenol in Ocimumsanctum(p) and ocimumbasilicum was high in the month of November (61.538%), (9.229%) and very low in the month of May(37.969%),(6.553%) respectively. Asha and coworkers[57] reported that the oil of ocimum sanctum posses

Eugenol(53.10%) as the main compound .Sonia verma[58] reported that in O. basilicumthe majorconstituents of the oil was linalool (64.35%) and with very low composition of eugenol(3.21%).

Name of the	%	Total Phenols	Total	Antioxidants	DPPH
plant	compositi	mg TA	flavonoids	mg of ascorbic	%inhibiti
	on of	equivalent/gm	mg of	acid equivalent	on
	Eugenol	(dw)	quercetin	/gm(dw)	
			equivalent/gm(
			dw)		
O.Sanctum(61.538%	127.3±1.8	0.733 ± 0.01	0.921±0.03	46.91%
<i>p</i>)					
O.Basilicum	9.229%	123±1.4	0.604 ± 0.02	1.047 ± 0.01	43.1%

Table 2: Percentage composition of eugenol, total phenols, flavonoids, antioxidants and **DPPHscavenging assay.**

Values are mean \pm standard deviation of three samples. dw: dry weight of the powdered sample

3.2 Total phenols:

The phenols contain hydroxyls that are responsible for antioxidant activity and the radical scavenging effect mainly due to redox properties [59-61]. In the present study O.Sanctum(p) and O.Basilicumshow thevalue of total phenols127.3±1.8 and123±1.4TAequivalent/gram.

Due to the higher phenol content in the leaf, the plant possesses high antioxidant activity and other pharmacological

activities [62].

3.3 Total flavonoids:

The compounds such as flavonoids, which contain hydroxyls, are responsible for the radical scavenging effect in the plants[63,64].

The flavonoid contents of the extracts in the present study in terms of

quercetin equivalent for O.Sanctum(p) and O.Basilicum were 0.733 ± 0.01 mg/g and 0.604 ± 0.02 mg/g respectively.Flavonoidswhich were isolated from the aqueous extract of *Tulsi*have been shown significant anti oxidantactivity, both *in vivo* and *in vitro*[15].

3.4 Antioxidants:

In the present study the values of antioxidants are given as 0.921 ± 0.03 mg/g for *Ocimumsanctum(p)*, and 1.047 ± 0.01 mg/g

ascorbic acid equivalent/gm for *Ocimumbasilicum* respectively. According to the recent reports, a highly positive relationship between total phenols and antioxidant activity was found in many plant species[65]. The phenolic compounds may contribute directly to the antioxidativeaction[66]. A number of phenolic compounds with strong antioxidant activity have been identified in *Ocimumbasilicum* plant extracts[67]

3.4 DPPH Scavenging assay

The DPPH molecule contains stable free radical which has been widely used to evaluate the radical scavenging ability of antioxidants.

DPPH stable free radical method is a sensitive way to determine the antioxidant activity of plant extracts [68,69] and measured the

inhibition of reactive species (free radicals)generated in the reaction mixture and these results depend on the type of reactive species

used [70]. In general essential oils exhibited weak to strong free DPPH scavenging activity rang can be shown as(weak: <25%; moderate: 25-50%; strong: >50%) at the tested concentration of 0.1% (v/v). The extracts of *Ocimum sanctum(p)*(46.91%) and *OcimumBasilicum*(43.1%) showed the moderate DPPH radical scavenging activity. The high inhibition of DPPH activity of *Ocimum sanctum* purple in this recentstudy(46.91%%) was lower than the value (71%) reported by LukmanulHakkimet al [71].

4. CONCLUSIONS

Eugenol, (l-hydroxy-2-methoxy-4- allylbenzene), is a phenolic compound, common and major volatile constituent of essential oils

extracted from leaves of the *Ocimum Sanctum*(p), but in *Basilicum* it is minor compound *.OcimumSanctum*(p) and OcimumBasilicum possessed significant antioxidant activity. in *Ocimum Sanctum*(p) it was mainlydue to the presence of eugenol, where as in *Ocimumbasilicum* it is due other phenolic compound like Linalool. These plants are easily available in our surroundings and the process of cultivation is also very easy. So we can make use of these plants to extract the costly componentslike eugenol& Linalool. These plants show different medicinal properties. So, still more clinical trials should be conducted to support their medicinal therapeutic uses.

ACKNOWLEDGEMENTS

One of the authors, M. Rama is thankful to the management of Ch.S.D.St.Theresa's (A)College for Women, Eluru, Andhra Pradesh, for their permission to conduct this work and to the University Grants Commission, New Delhi, India, for awarding FDP scholarship during

2009-11.

REFERENCES

1. Mukherjee, P.K., Maiti, K., Mukherjee, K.and Houghton, P.J., J. Ethnopharmacology,**106**: 1 (2006).

2. Akgül A. : Spice Science and Technology. Turkish Association Food TechnologistsPubl. No. 15, Ankara, Turkey. (in Turkish)(1993)

- 3. Baritaux O., Richard H., Touche J., Derbesy M., Flavour Fragr.J., 7: 267–(1992).
- 4. Gunther, E. The Essential Oils, 5: 160 (1952).
- 5. Zegorka G and Glowniak K. (2001) J. Pharm. Biomed. Anal. 26: 179(2001).
- 6. Heath H.B.: Source Book of Flavour. Avi Publ., Westport(1981).
- 7. Akgül A. (1989): Nahrung, 33: 87(1989).
- 8. Lachowicz K.J., Jones G.P., Briggs D.R., BienvenuF.E., PalmerM.V., TingS.S.T., Hunter M,
- J. Agr. Food Chem., 44: 877(1996).

9. Machale K.W., Niranjan K., Pangarkar V.G. J. Chem. Tech.Biotechnol., 69:362(1997).

10. Jaggi RK, Madaan R, Singh B. Indian J ExpBiol; 14;1329 (2003).

- 11. Kelm MA, Nair MG, Strasburg GM, DeWitt DL. Phytomedicine;7:7 (2000).
- 12. Shishodia S, Majumdar S, Banerjee S, Aggarwal BB. Cancer Res; 63:4375 (2003).
- 13. Malik.M,Shafiq;Rafique.M,Pak.j.sci.ind.res.,**30**(5),369 (1987).
- 14. Skaltsa H, Couladi M, Philianos S, Singh M.Fitoterapia; 58: 286 (1987).

15. Uma Devi P, Ganasoundari A, Vrinda B, Srinivasan KK, Unnikrishnan MK. RadiatRes; **154**:455 (2000).

16. Anonymous. Wealth of India, *Publicationand Information Directorate*, *CSIR*, *NewDelhi*.**7**: 79 (1991).

17. Rajeshwari. S, *Ocimum sanctum*. The Indianhome remedy. In: Current Medical Scene,(Cipla Ltd., Bombay Central, Bombay)(1992).

18. Khanna S, Gupta SR, Grover SK. Indian JExperimental Biology; 24: 302 (1986).

19. Bhargava KP, Singh N. Indian J MedicalResearch; 73: 443 (1981).

20. Ray A. Recent trends in stress research:Focus on adaptogenesis. Proc. XXXVIIIthConference of Indian PharmacologicalSociety held at Punjabi University, Patiala,Nov. 23–26, p.68 (1995).

21. Sarkar A, Pandey DN, Pant MC. Indian JPhysiology Pharmacology; 38(4): 311(1994).

22. Sethi J, Sood S, Seth S, Thakur A. Indian JPhysiol Pharmacology; 47(1): 115 (2003).

23. L.O. Orafidiya, S.K. Adesina Jr, O.A. Igbeneghu, E.O. Akinkunmi, G.E. Adetogun, A.O. Salau. International Journal of Aromatherapy, **16**(2):57(2006).

24. Martins AP , Salgueiro LR , Vila R. Planta Med . ; 65 (2): 187(1999).

25. Khatri L.M., Nasir M.K.A., Saleem R., Noor F. Pakistan J. Sci. Ind. Res., 38:281(1995).

26. Pino J.A., Rosado A., Fuentes V. J. Essent. Oil Res., 8: 139(1996).

27. Grayer, R. J., S. E. Bryan, N. C. Veitch, F. J. Goldstone, A. Paton and E.Wollenweber. Phytochemistry43:1041(1996).

28. Nishida, R., W.S. Bowers, and P.H. Evans. J. Chem. EcoL10:1435(1984)

29. Sajjadi, S. E. Daru, 14(3): 128(2006).

30. Lee, J., &Scagel, C. F. Food Chemistry, 115, 650(2009).

31. Lahariya A.K., Rao J.T. Ind. Drugs, 16: 150(1979).

32. Dube S., upadhyay P.D., Tripathi S.C. Can. J. Bot., 67: 2085(1989).

33. Özcan M. Z. Leb.-Mitt. Unters. u- Forsch. A, 207: 253(1998).

34. Reuveni, R., A. Fleisher, and E. Putievsky. Phytopathology Z. 110: 20(1984).

35. R.A. Werner, Environ. Entomol. 24 : 372(1995).

36. Chavan, S. R., and S. T. Nikam. Indian J. Med Res. 75:220(1982).

37. Lawrence, B. M. *Labiatae*oils-mother nature. Chemical factory, essential oils, allured, carol stream, IL, pp188(1993)

38. Deshpande, R. S., and H. P. Tipnis. Pesticides 11: 11(1977).

39. Duke JA (2001). Handbook of Phytochemical Constituents of GRAS Herbs and OtherEconomic Plants. Boca Raton, London, New York, Washington, D.C: CRC Press 40. Chogo, J. B., and G. Crank. J. Nat. Prod. **44**: 308(1981).

41. Keita, S. M., C. Vincent, J. P. Schmit, J. T. Arnason, and A. Belanger. J. StoredProducts Res. **37**: 339(2001).

42. Salvatore, A., S. Borkosky, E. Willink, and A. Bardon. J. Chem. Ecol. 30: 323(2004).

43.Okigbo, R.N., Okeke, J.J. and Madu, N.C. Journal of Agricultural Technology ;6(4):703(2010).

44. Maganga, M. E., G. Gries, and R. Gries.. Environ. Entomol. 25: 1182(1996)

45. Paula, J. P., P. V. Farago, LEM. Checchia, K. M. Hirose, and JLC Ribas. ActaFarmaceuticaBonaerense; 23: 376(2004).

46. Popovic, Z., M. Kostic, S. Popovic, and S. Skoric. Biotechol. Biotechnological Equip.20: 36(2006)

47. Chatterje, A., N. C. Sukul, S. Laskal, and S. Ghoshmajumdar. J. Nematol. 14:118(1982)

48.Nitezurubanza, L., J.J.C. Scheffer, A. Looman, and A. BaerhiemSvendsen. Planta Med.385(1984).

49. G. Samuelsson, Drugs of natural origin, in: A Textbook of Pharmacognosy, SwedishPharmaceutical Press, Stockholm, p. 320(1991).

50. Mustafa Z. Ozel, HilalKaymaz, Superheated water extraction, steamdistillation and Soxhlet extraction of of *Origanumonites*. Analytical and Bioanalytical Chemistry, Volume 379, Issue 7-8, pp 1127-1133(2004).

51. Clevenger, J. F. Apparatus for the determination of volatile oil. Journal of the American Pharmacists Association, **17**, 346 (1928).

52. McDonald S, Prenzler PD, Autolovich M, Robards K. Phenolic content and antioxidant activity of olive extracts. Food Chemistry,

73:73-84 (2001).

53. Meda, A., Lamien, C. E., Romito, M., Millogo, J. &Nacoulma, O. G. Determination of the total phenolic, flavonoid and proline contents in Burkina fasanhoney, as well as their radical scavenging activity. Food Chemistry, **91**, 571–577 (2005).

54. Oyaizu, M. Studies on products of browning reaction–antioxidant activities of products of browning reaction prepared from glucoamine. *Jap. J. Nutr.* **44**, 307–315 (1986).

55. Blois MS. Antioxidant determinations by the use of a stable free radical. Nat. **26**: 1199-1200 (1958).

56. Govindarajan R., M. Vijaya Kumar, A.K.S. Rawat, S. Mehrotra. Free radical scavenging potential of *Picrrorhizakurroa*Royle ex

Benth. Indian J. Exptl. Biol., 41: 875 (2003).

57. Asha MK, PrasathD,MuraliB,Padmaja R and Amit A. Anthelmintic activity ofessentialofocimum sanctum and eugenol. Fitoterapia (2001).

58. Sonia Verma, PreetiKothiyal .pharmacological activities of different species of tulsi*International Journal of Biopharm& Phytochemical Research* Vol. 1(1), Jan 2012; 21-39 59. Arumugam P, Ramamurthy P, SanthiyaSTand Ramesh A. Antioxidant activitymeasured in different solvent fractions

obtained from Menthaspicata Linn.: Ananalysis by ABTS.+ decolorizationassay.Asia Pac. J. Clin. Nutr. 119-124 (2006).

60. Rahman K. Studies on free radicals, antioxidants, and cofactors. ClinIntervAging. 2(2): 219–236 (2007).

61. Rice-Evans CA, Miller NJ, PagangaG.Antioxidant properties of phenoliccompounds. Trend. Plant Sci. 4:152-159(1997).

62. Nikhil Kumar1, PragyaMisra, AnuradhaDube, Shailja Bhattacharya, MadhuDikshitand ShirishRanade. *Piper betel* Linn., a

maligned Pan-Asiatic plant with an array of pharmacological activities and prospects fordrug discovery; current science, vol. 99, no.7, 10 october 2010.

63. Das NP, Pereira TA . Effects of flavonoids on thermal autooxidation of Palm oil:structure- activity relationship. J. American Oil Chemists Society, 67: 255- 258(1990).

64. Younes M. Inhibitory action of some flavonoids on enhanced spontaneous lipidperoxidation following glutathione depletion. Planta Medica, 43: 240- 245(1981).

65. Gulcin I, Oktay M, Kufrevioglu OI, Aslan A . Determination of antioxidant activity ofLichen Cetrariaislandica (L). Ach. J. Ethnopharmacol., 79(3): 325-329(2002b)

66. Duh PD, Tu YY, Yen GC . Antioxidant activity of water extract of HarngJyur(Chrysanthemum morifolium Ramat). Lebensmittel- Wissenschaft und Technologie, 32: 269-277(1999).

67. Nakatani, N.N. Antioxidants from species and herbs. In: *Natural antioxidants: Chemistry health effects and application*. (Ed.): F. Shahidi. pp. 64-75. Champaign, IL:AOCS Press(1997).

68. Koleva II, Van Beek TA, Linseen JPH, deGroot A, Evstatieva LN. Screening of plantextracts for antioxidant activity: a

comparative study on three testing methods. Phytochem. Anal. 13:8-17 (2002).

69. Suresh PK, Sucheta S, Sudarshana VD, Selvamani P, Latha S. Antioxidant activityin some selected Indian medicinal plants.

Afr. J. Biotechnol. 7: 1826-1828 (2008).

70. Cao G, Sofic E, Prior RL. Antioxidantcapacity of tea and common vegetables. J.Agric. Food Chem. 4: 3426-3431 (1996).

71. LukmanulHakkim F, Gowri Shankar C, Girija S. Chemical composition and antioxidant property of holy basil (Ocimum sanctum L.) leaves, stems and inflorescence and their invitro callus cultures. Journal of Agricultural and Foodchemistry. 55: 9109-9117(2007).