

# Recent Approaches to the Synthesis of Medicinally Potent Pyrimidine Derivatives using Microwave: A Review

Sinderpal Kabalsingh Tank<sup>ab</sup>

<sup>a</sup>Department of Chemistry, Sant Baba Bhag Singh University, Jalandhar, India

<sup>b</sup>Department of Chemistry, Kanya Maha Vidyalaya, Jalandhar, India

## ABSTRACT:

*Syntheses of medicinally potent pyrimidine derivatives are the most significant tasks in N-heterocyclic chemistry because these compounds have proved to be very attractive and useful for the design of new molecular frameworks of potential drugs with varying pharmacological activities. Among the heterocyclic compounds, pyrimidines occupy a central position due to their presence in genetic material of cells. This article aims to review the work reported on the synthesis of pyrimidine derivatives using microwave, the chemistry and the biological activities of pyrimidines (2009 to 2017).*

**Keywords:** *Pyrimidines, synthesis, microwave, biological activity.*

## Introduction

Heterocyclic chemistry is an important branch of organic chemistry accounting for nearly one-third of modern publications [1]. Heterocyclic compounds have vital role in our biological system. They are an integral part of many pharmacologically active molecules, natural products and nucleic acids. The base pair of DNA & RNA (guanine, cytosine, adenine and thymine) are also made up of heterocyclic compounds like purine, pyrimidine etc. Pyrimidine (Fig. 1) is the parent ring system of a variety of substances that play vital roles in biological processes. This structural design also has a subset of a large number of pharmaceutical agents and naturally occurring substances such as vitamins, coenzymes, purines, pterins, nucleotides and nucleic acids. The properties of pyrimidines are governed to a large extent by the electron-attracting properties of the two nitrogen atoms; each of these reinforces the electronic effect of the other in the 2-, 4- and 6-positions. Therefore, electrophilic attack will be strongly hindered at these positions, and nucleophilic attack is strongly facilitated. The 5-position is affected only by the inductive effect of the two nitrogen atoms, thereby making this position susceptible to electrophilic attack. Pyrimidine derivatives have various therapeutic applications in medicinal chemistry. One anticipated reason for their activity is presence of a pyrimidine base in thymine, cytosine and uracil, which are essential building blocks of nucleic acids, DNA & RNA [2]. Number of chemical compounds consisting of pyrimidine as core nucleus are reported to be synthesized and evaluated for antihypertensive [3], anticancer [4], antimicrobial [5], antihyperglycemic [6], antiarrhythmic, anti-inflammatory [7], analgesic [8], antibacterial [9], anti-HIV [10] and antitubercular activity [11].

A major challenge of modern synthetic chemistry is to search the ways of developing and applying more efficiently and environmentally benign strategies for future sustainable growth [12]. In this context Green Chemistry has become a research field of great interest due to replacement with greener reagents and solvents instead of hazardous and toxic materials [13].

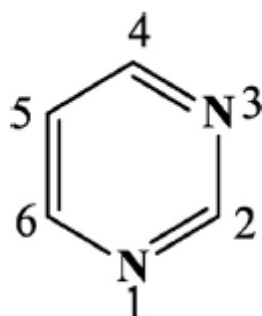
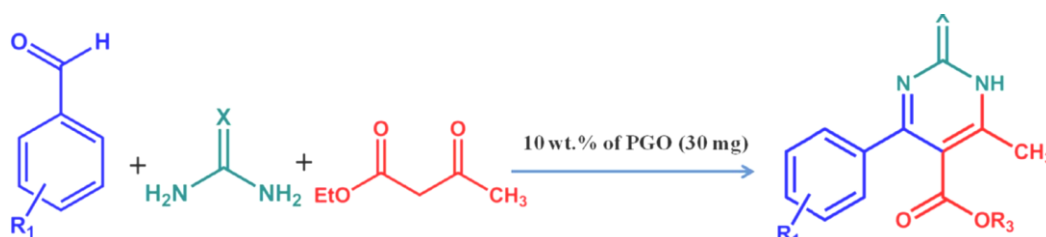


Fig. 1. Pyrimidine.

Now, green chemistry is a new and valuable strategy of looking at organic synthesis and the designing of drug molecules, offering considerable environmental and economic benefits over traditional synthetic processes[14]. The accomplishment of reactions in the presence of green mediums (use of recyclable ionic liquids, greener and safer alternative solvents) or under solvent-free conditions, microwave and ultrasonic technologies are some of types of green synthetic strategies. In the last few years microwave assisted organic synthesis is gaining popularity as a non-conventional technique for rapid organic synthesis. Some interesting features of this method are the rapid reaction rates, simplicity, solvent-free conditions, the ease of work-up after the reaction, and better selectivity[15–17]. It has increased the safety of organic reactions. There is a rate acceleration of approximately 385 times under microwave conditions. It is said to be an eco- friendly method. Since there is decrease in heating time it effectively reduces the cost of fuel and minimizes environmental pollution [18, 19]. Microwave energy, however, heats the entire samples at once, eliminating hot spots and reducing reaction times, all of which results in larger, purer yields.

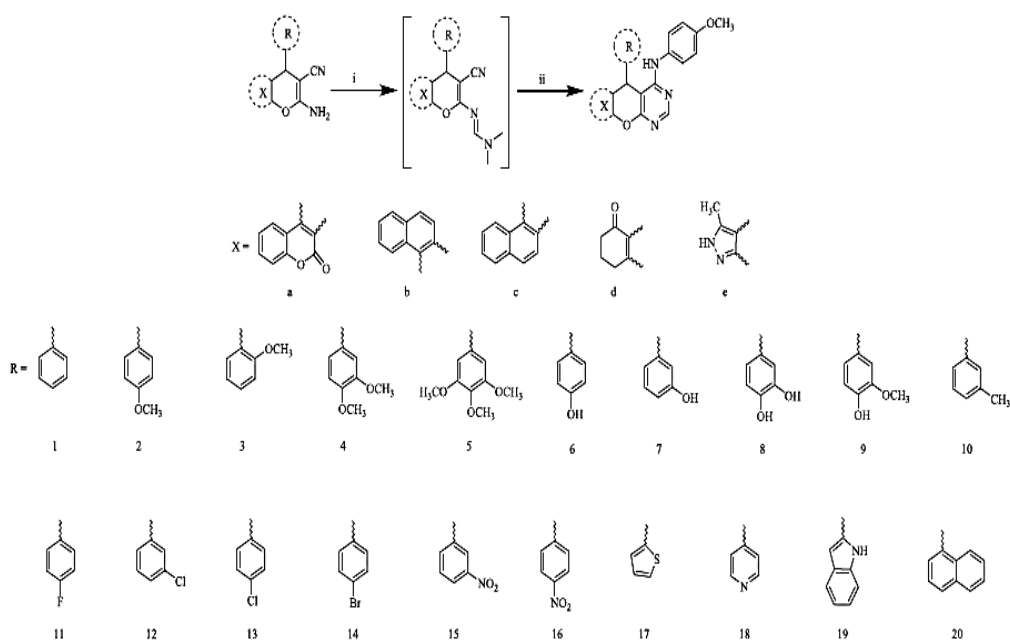
This review paper summarizes the synthesis of pyrimidine derivatives using non conventional microwave techniques, the chemistry and the biological activities of pyrimidines (from 2009 to 2017).

Acharya L., et al.[20] reported the microwave assisted synthesis of DHPM and DAPM (3,4-dihydropyrimidin-2(1H)-ones (DHPMs) and 4,6-diarylpyrimidones)(DAPM) molecules using phosphate functionalized GO (PGO) (phosphonated graphene oxide) as an efficient heterogeneous catalyst. The Pyrimidones synthesis was obtained via ternary condensation of urea, aldehydes, and 1, 3-dicarbonyl in the presence of PGO nanocatalyst (Scheme 1)



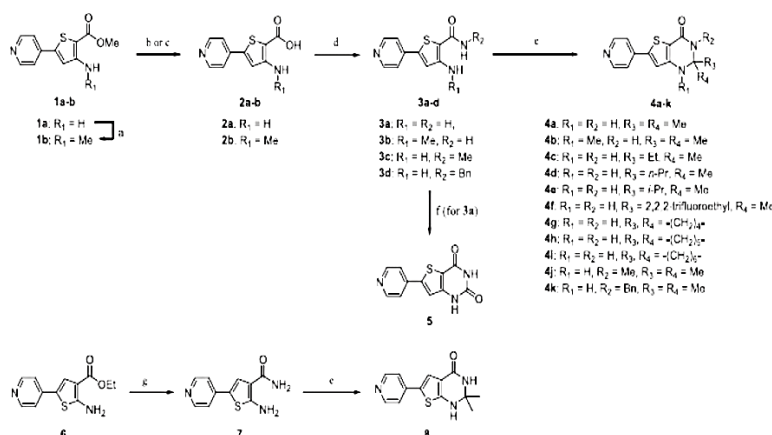
Scheme 1. A model reaction for the synthesis of 3,4-dihydropyrimidin-2(1H)-one.

Kaur Manroopraj et al.[21] reported synthesis of pyrano[3,2-d]pyrimidines from malononitrile, aryl/heteroaryl aldehyde and CeH activated acidic compound in the presence of catalytic amounts of DMAP under microwave irradiation (Scheme 2). The pyrimidine derivatives were screened for effective xanthine oxidase (XO) enzyme inhibitors. Docking study were performed to investigate the recognition pattern between xanthine oxidase and the most potent XO inhibitor.



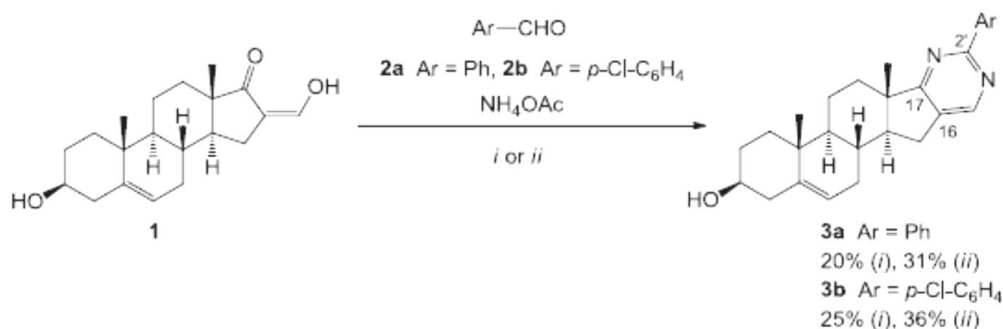
Scheme 2. Synthesis of fused pyrano[3,2-d]pyrimidine derivatives. Reagent and reaction condition: (i) DMF-DMA, stir, rt; (ii) p-anisidine, glacial acetic acid, reflux

Kurasawa Osamu et.al [22] reported synthesis of Dihydrothienopyrimidine derivatives possessing 4-pyridyl group as a hinge binder (4a–k, 5, 8) according to the procedure in Scheme3. After N-methylation of methyl 3- amino-5-(4-pyridyl)thiophene-2-carboxylate 1a, the esters 1a, b were then hydrolyzed under basic conditions to give the carboxylic acids 2a, b. Compounds 2a, b were condensed with appropriate amines to provide the corresponding amides 3a–d. Similarly, ethyl 2-amino-5-(4-pyridyl)thiophene-3-carboxylate 6 was converted to amide 7 in 38% yield by saponification and subsequent amidation. The aminothiophenecarboxamides 3a–d, 7 were reacted with various ketones in the presence of catalytic amount of p-toluenesulfonic acid to afford cyclized products 4a–k, 8 in 11–74% yields. Moreover, the amide 3a was subjected to ring-closure reaction using urea to yield thienopyrimidin-2,4-dione 5. Further structure-activity relationships(SARs), and biological effects of this chemical series was studied. Furthermore, difference in binding modes of a selected compound with Cdc7 and ROCK in docking models is discussed to elucidate high selectivity.

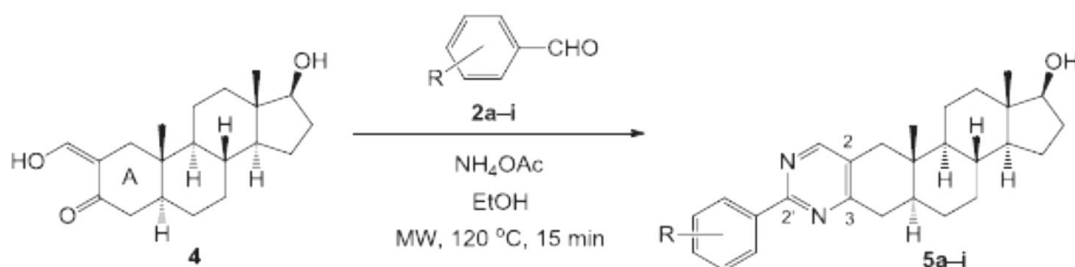


Scheme 3. Reagents and conditions: (a) NaH, MeI, DMF, 0 °C to rt, 62%; (b) NaOH, MeOH–water, 70°C, 92%; (c) LiOH, MeOH–water, 80°C, quant.; (d) NH<sub>4</sub>Cl or R<sub>2</sub>NH<sub>2</sub>, EDCI, HOBT, Et<sub>3</sub>N, DMF, rt, 50–84%; (e) R<sub>3</sub>COR<sub>4</sub>, p-TsOH, H<sub>2</sub>O, toluene or AcOH, 70–110°C, 11–74%; (f) urea, 180°C, 34%; (g) (1) LiOH, EtOH–water, 80°C; (2) NH<sub>4</sub>Cl, EDCI, HOBT, Et<sub>3</sub>N, DMF, rt, 38%.

BajiÁdám, et.al [23] reported synthesis of novel ring D- and A-fused pyrimidines in the androstane series efficiently within 10-15 min in polar protic solvents under microwave irradiation via two kinds of multicomponent heterocyclization reactions followed by spontaneous or promoted oxidation (Scheme 4a and 4b). The synthesized compounds were tested in vitro on human cancer cell lines as well as on non-cancerous fibroblast cells by the MTT assay in order to investigate their biological effects. As a result of the pharmacological screen, a remarkable structure-function relationship was observed as the acetylated Biginelli products exhibited higher toxicity compared to the deacetylated version of each compound.

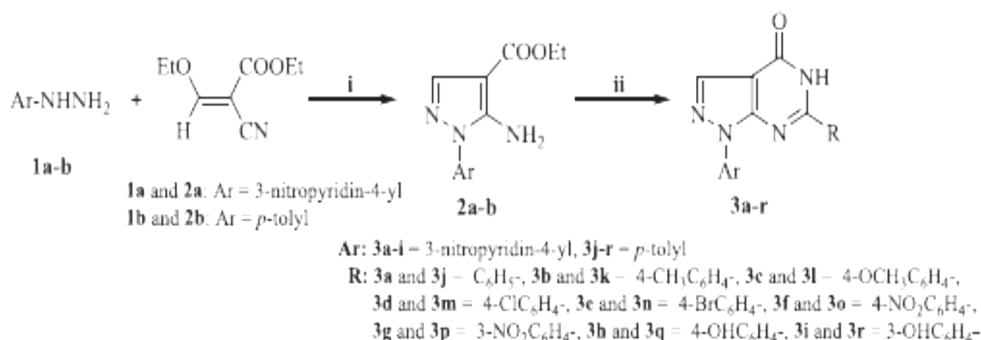


Scheme 4a: MW-assisted synthesis of ring D-fused arylpyrimidines in the androstane series. Reagents and conditions: 1:2:NH<sub>4</sub>OAc = 1:2:2 (*i*) silica gel, MW, 120 °C, 6 min (solvent-free); (*ii*) EtOH, MW, 120 °C, 15 min.



Scheme 4b: MW-assisted synthesis of ring A-fused arylpyrimidines in the androstane series

Sureja Dipen K. et.al [24] reported an efficient and solvent-free synthesis of some novel pyrazolo[3,4-d]pyrimidin-4(5H)-one derivatives under microwave irradiation by the reaction of ortho amino ester of 1-substituted pyrazole with different aromatic nitriles in the presence of potassium tert-butoxide. Furthermore, all the newly synthesized compounds have been screened in-vitro for their antimicrobial activities against *S. aureus*, *B. subtilis*, *E. coli*, *P. aeruginosa*, *A. niger* and *C. albicans*. All the compounds showed weak to good activity against all tested microorganisms.

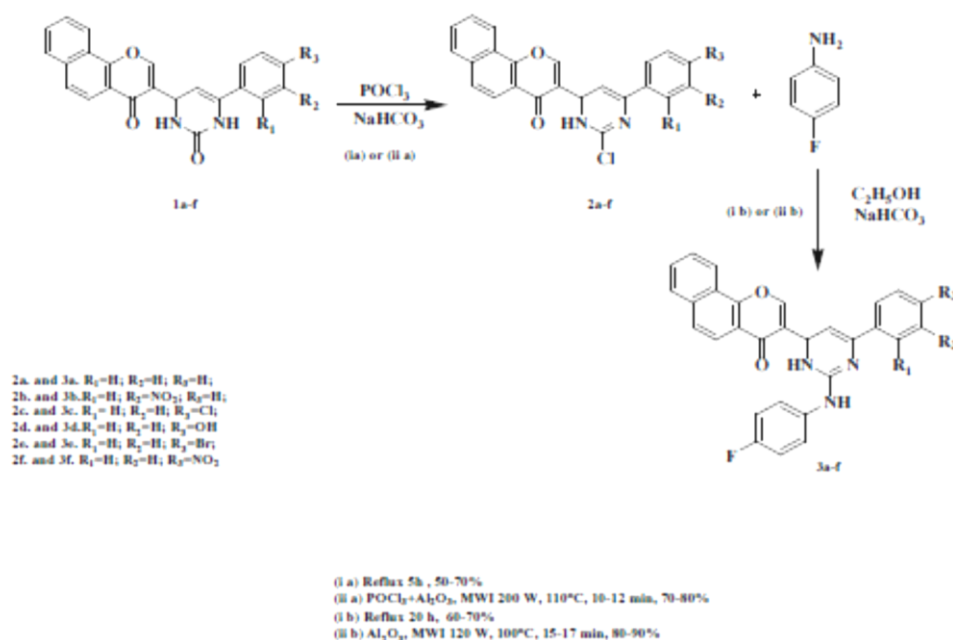


Reaction conditions: *i* = Dioxane, Piperidine, stirring, 2-3 h.

*ii* = R-CN, *t*-BuOK, MW irradiation, 1.5-3 min, (75-92%)

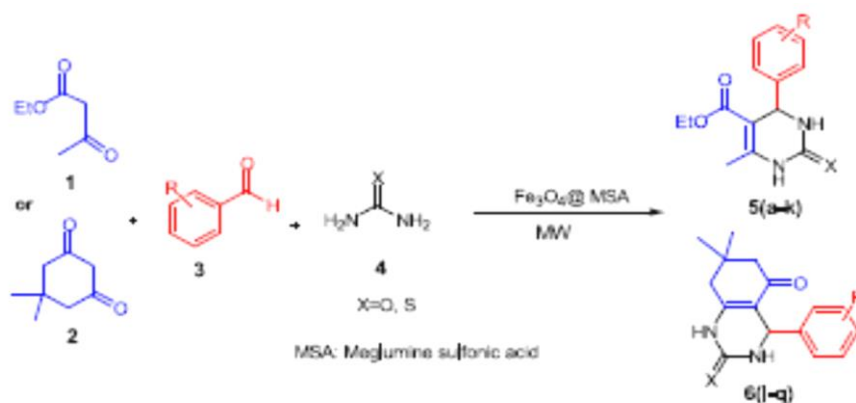
## Scheme 5: Synthetic route of pyrazolo[3,4-d]pyrimidin-4(5H)-one derivatives

LalaKundan, et.al [25] reported the microwave assisted synthesis of some new pyrimidine derivatives with benzochromone pyrimidine nucleus along with substituted phenyl ring. The synthesis of 3-(2-chloro-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 2a-f and 3-(2-(4-fluorophenylamino)-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 3a-f is discussed. In first step, synthesis of 3-(2-chloro-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 2a-f have been carried out from 4-(4-oxo-4H-benzo[h]chromen-3-yl)-6-substituted phenyl-3,4-dihydropyrimidin-2(1H)-one 1a-f by mixing with phosphoryl chloride ( $\text{POCl}_3$ ) and irradiating under microwave radiation at 175 W at 110 °C for 10-12 min. In step 2, a mixture of 3-(2-chloro-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 2a-f (1mmol) and alumina and p-fluoro aniline (1 mmol) were irradiated under microwave irradiation at 175 W at 110 °C for 10-12 min (Scheme 6). These new compounds have been evaluated for their biological activities against various species of gram positive and gram negative bacteria and the compounds possessed promising antibacterial properties.

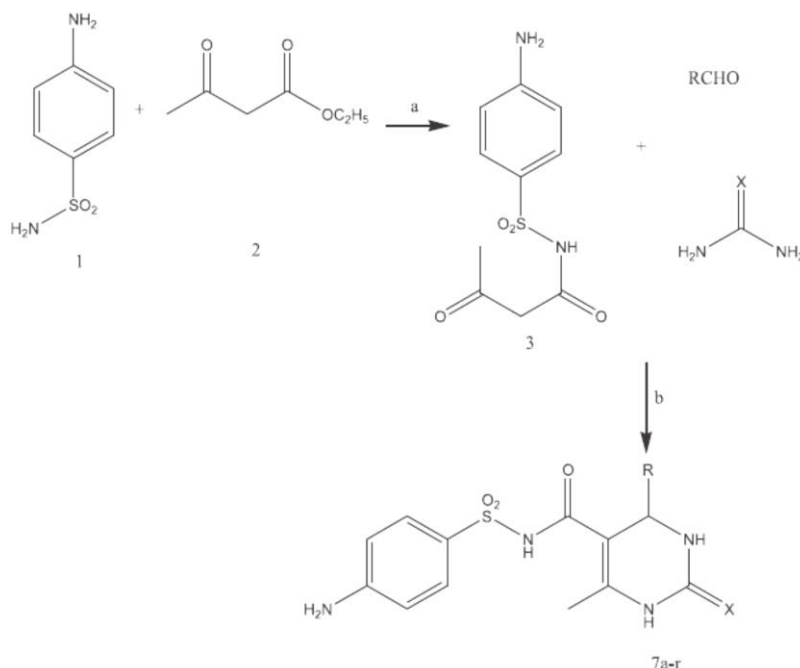


Scheme 6: Synthetic route of 3-(2-chloro-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 2a-f and 3-(2-(4-fluorophenylamino)-6-substituted phenyl-3,4-dihydropyrimidin-4-yl)-4H-benzo[h]chromen-4-one 3a-f.

Moradi Leila et.al [26] reported the synthesis of 3,4-dihydropyrimidinone/thiones by reacting 0.025 g of  $\text{Fe}_3\text{O}_4@MSA$  ( $\text{Fe}_3\text{O}_4@$  meglumine sulfonic acid) with a homogeneous mixture of aldehyde (1 mmol), b-dicarbonyl (1 mmol) urea or thiourea (1.2 mmol) in a 5 ml of water/ethanol (1:1) under microwave irradiation (400 w, 80°C) (Scheme 7).

Scheme 7: Synthesis of 3,4-DHPs using  $\text{Fe}_3\text{O}_4\text{@MSA}$ . L.

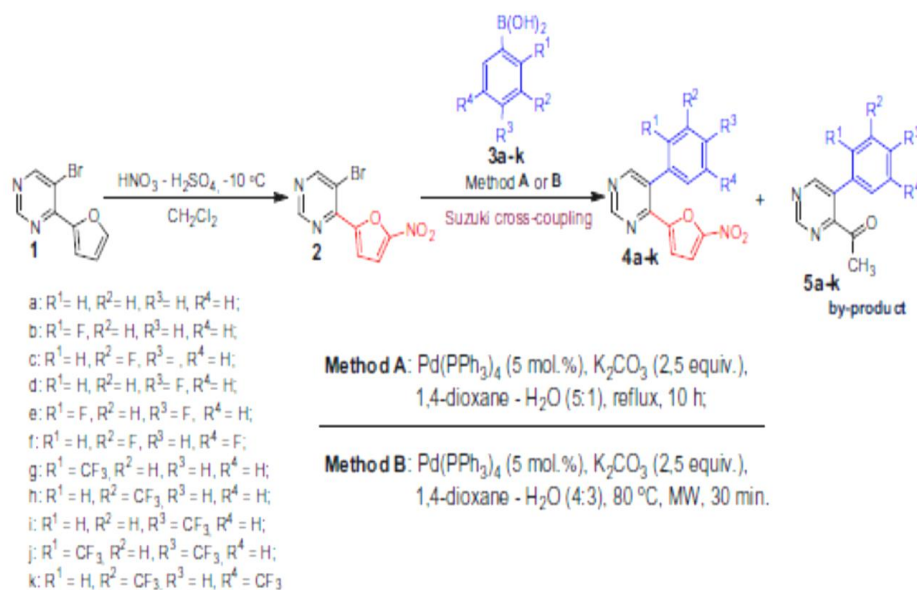
Elumalai Karthikeyan et.al [27] reported synthesis of series of novel sulphanilamide condensed 1,2,3,4-tetrahydropyrimidines by reacting N-[(4-aminophenyl) sulphonyl]-3-oxobutanamide with urea/thiourea and aryl aldehyde in the presence of a catalytic amount of laboratory prepared chlorosulphonic acid as an efficient catalyst by reacting mixture of N-[(4-aminophenyl) sulphonyl]-3-oxobutanamide (0.005 M), urea/thiourea (0.0075 M), and aryl aldehyde (0.005 M) with a catalytic amount of chlorosulphonic acid in 10 ml of ethanol and subjecting to microwave irradiation (300 W) for 14 min with 10 s intervals (Scheme 8). The synthesized compounds were evaluated for antimicrobial activity against the Gram-positive bacteria *Bacillus subtilis* and the Gram-negative bacteria *Escherichia coli* and for cytotoxicity against Vero cells. These compounds exhibited weak, moderate, or high antimicrobial activities and cytotoxicity.



Scheme 8: Synthesis of compounds (7sa-r). Reagents and conditions: (a) reflux 2.5 h, catalytic amount of anhydrous  $\text{K}_2\text{CO}_3$  and 10 ml of ethanol (95%); (b)  $\text{C}_2\text{H}_5\text{OH}$ , chlorosulphonic acid, and microwave irradiation (300 W) for 14 min.

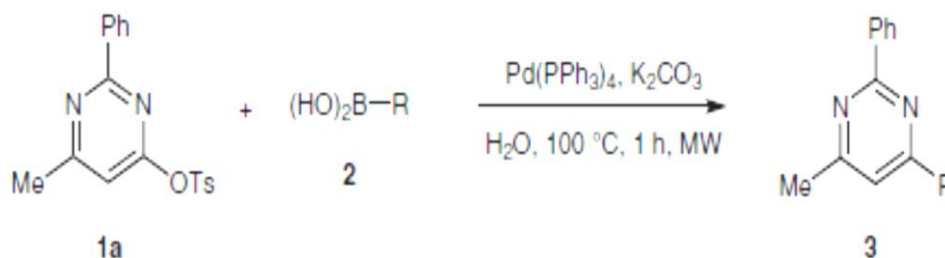
Verbitskiy Egor V. et.al [28] reported synthesis of 5-aryl-4-(5-nitrofur-2-yl)-pyrimidines by reaction of 5-bromo-4-(5-nitrofur-2-yl)-pyrimidine (2) with the corresponding boronic acid (3a-k) under microwave irradiation in 1,4-dioxane- $\text{H}_2\text{O}$  (4:3) in the presence of  $\text{K}_2\text{CO}_3$  and  $\text{Pd}(\text{PPh}_3)_4$ , as catalyst (Scheme 9, method B). All synthesized compounds were screened in vitro for their antibacterial activities against twelve various bacterial

strains. It was demonstrated that some of these compounds exhibited significant antibacterial activities against strains *Neisseria gonorrhoeae* and *Staphylococcus aureus*, comparable and even higher with that commercial drug Spectinomycin.



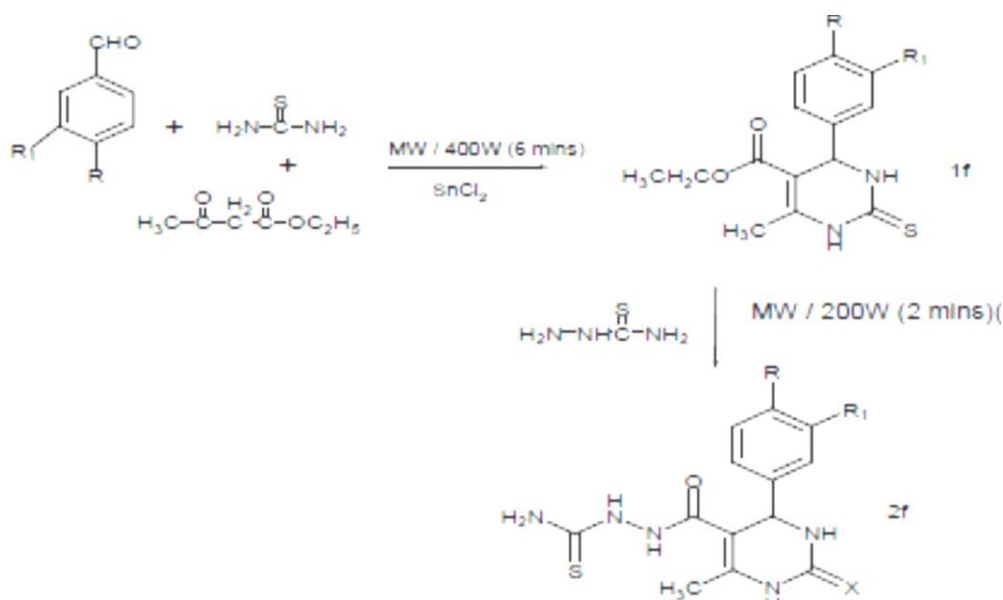
Scheme 9 : Synthesis of 5-aryl-4-(5-nitrofuran-2-yl)pyrimidines (4a–k)

AguilarRodríguez, et.al [29] reported the palladium catalysed Suzuki–Miyaura coupling reaction of 4-pyrimidyl tosylates with aryl, heteroaryl and alkyl boronic acids. The reaction provided 4-substituted pyrimidines in good-to-excellent yields after one-hour microwave irradiation in water at 100 °C. (Scheme 10)



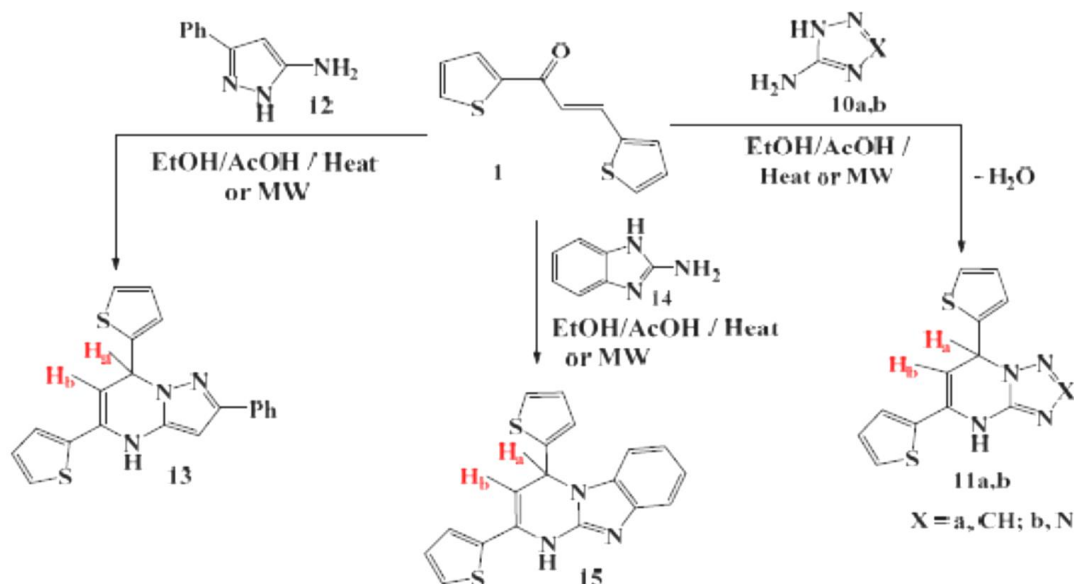
Scheme 10: Suzuki–Miyaura coupling of 4-pyrimidyl tosylate 1 with various aryl and heteroaryl boronic acids

Karthic R, et.al [30] reported synthesis of 5-(5-amino-1,3,4-thiadiazol-2-yl)-3,4-dihydro-6-methyl-4-phenylpyrimidin-2(1H)-thione (3f-j) (Scheme 11). Compound 1 was converted to carbothiamide 2 by reacting compound 1 with thiosemicarbazide in catalytic amount of acetone is irradiated with help of domestic microwave oven (200W) for 2 minutes. The compound 2 was converted to corresponding thiadiazole 3 by treatment with conc. H<sub>2</sub>SO<sub>4</sub> and NH<sub>3</sub>. Few of these Pyrimidine derivatives had been evaluated for their possible antibacterial activity. Most of the tested compounds showed significant antibacterial activity



Scheme 11: Synthesis of 5-(5-amino-1,3,4-thiadiazole-2-yl)-3,4-dihydro-6-methyl-4-Phenyl pyrimidin-2(1H)-thione (3f-j)

GomhaSobhi M., et.al [31] reported synthesis of Azolopyrimidines 11a,b, 13, and 15 via the reaction of chalcone 1 with heterocyclic amines 10a,b, 12, and 14 in ethanol in the presence of catalytic amount of AcOH using both thermal amines 10a,b, 12, and 14 in ethanol in the presence of catalytic amount of AcOH using both heating and microwave irradiation for comparison (Scheme 12). Their pharmacological activity toward bacteria and fungi inhibition was screened and compared to the references Chloramphenicol and Trimethoprim/sulphamethoxazole. The antimicrobial results of the investigated compounds revealed promising results and some derivatives have activities similar to the references used.

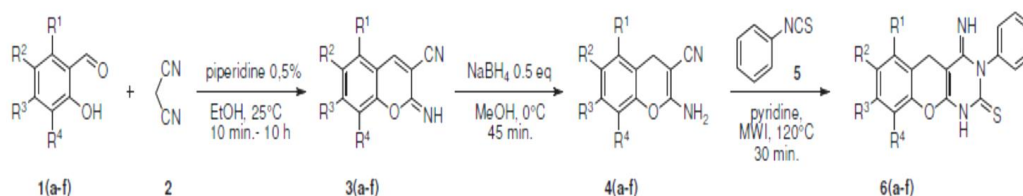


Scheme 12: Synthesis of azolopyrimidine derivatives 11a,b, 13 and 15.

Bouattour Ali et.al [32] reported synthesis of new 4-imino-3,4-dihydro-1H-chromeno[2,3-d]pyrimidine-2(5H)-thiones 6(a-f) without substituent in C-5 position by carrying out reaction between 2-amino-4H[1]chromene-3-carbonitrile 4 (2.5 mmol.), commercial phenylisothiocyanate 5 (338 mg, 2.5 mmol.) and dry pyridine (5 mL) and irradiating reaction mixture at 120°C for 30 min. under vigorous magnetic stirring. (Scheme 13). The

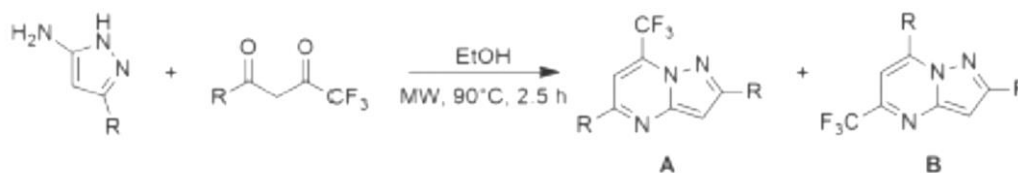


biological activities of the compounds were explored and found to have impact on protein kinase activity and antiproliferative activity with representative tumoral cell lines.



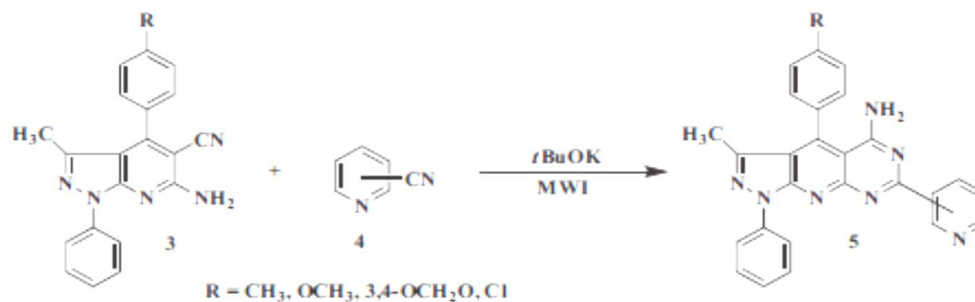
Scheme 13: Synthesis of 2-amino-4H-1-chromene-3-carbonitrile 4(a-f) and 4-imino-3-phenyl-3,4-dihydro-1H-chromeno[2,3-d]pyrimidine-2(5H)-thione 6(a-f) under microwave irradiation

Haucka Stefanie et.al [33] reported synthesis of 7-(trifluoromethyl) pyrazolo[1,5a]pyrimidine by reaction of 1,3-diketone (1 equiv) and the 3-(5)-aminopyrazole (1 equiv) in 3ml ethanol and heating to 90°C via microwave irradiation for 2.5 h. (Scheme 14). Pyrazolo[1,5a]pyrimidine-based FUBP1 inhibitor derived from medium throughput screening, was found to interfere with the binding of FUBP1 to its single stranded target DNA FUSE.



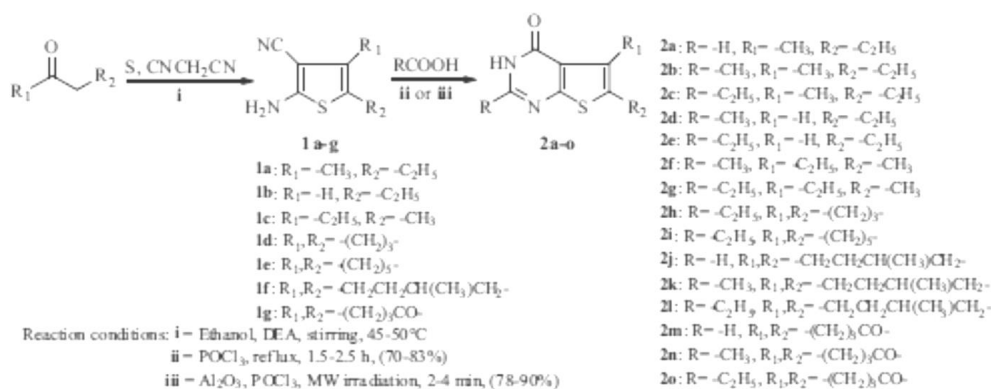
Scheme 14: Synthesis of pyrazolo[1,5a]pyrimidine derivatives

Acosta Paola et.al [34] reported synthesis of novel fused pyrazolo[4,3-b:5,6-d]pyrido[2,3-d]pyrimidines by reaction of mixture of ortho-aminonitrile 3 (0.3 mmol), cyanopyridine 4 (0.4 mmol) and tBuOK (10 mol%), and exposing to microwave irradiation from 1 to 6 min at 100 °C, a power of 250W and 30 PSI of pressure. (Scheme 15). All compounds were also tested for antifungal properties against two clinically important fungi; *Candida albicans* and *Cryptococcus neoformans*. Several compounds showed moderate activity against both fungi.



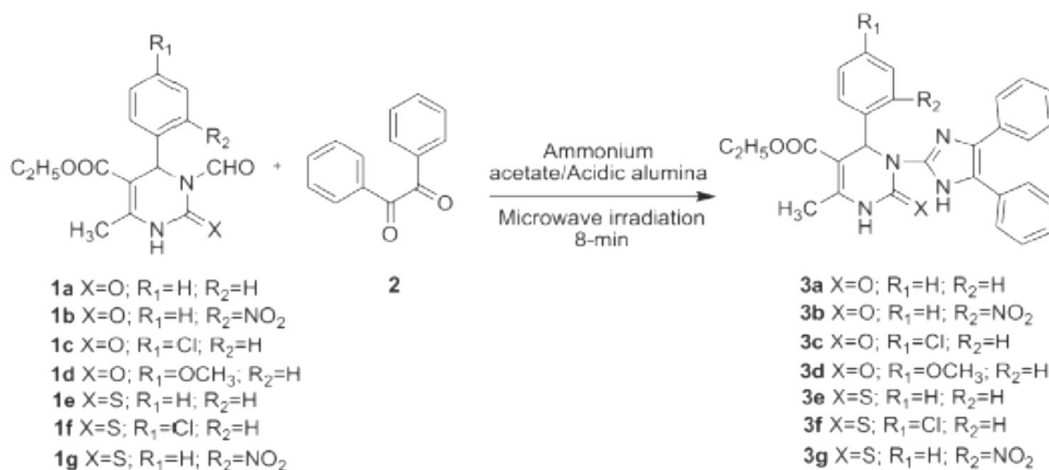
Scheme 15: Synthesis of pyrazolo[4,3-b:5,6-d]pyrido[2,3-d]pyrimidine derivatives 5

Sureja Dipen K. et.al [35] reported synthesis of thieno[2,3-d] pyrimidin-4(3H)-one derivatives (2a-o) by finely crushing mixture of 2-amino-4,5-substituted thiophene-3-carbonitrile (1a-g) (1 mM), appropriate aliphatic acid (2 mL) and alumina (0.5 g) and then adding phosphorus oxychloride (0.2 mL). The glass vial was then capped and microwaves were irradiated in a microwave oven at the power of 960W for 2-4 min. (Scheme 16). The compounds were screened in vitro to study their antimicrobial activity, which showed weak to moderate activity against all tested microorganisms.



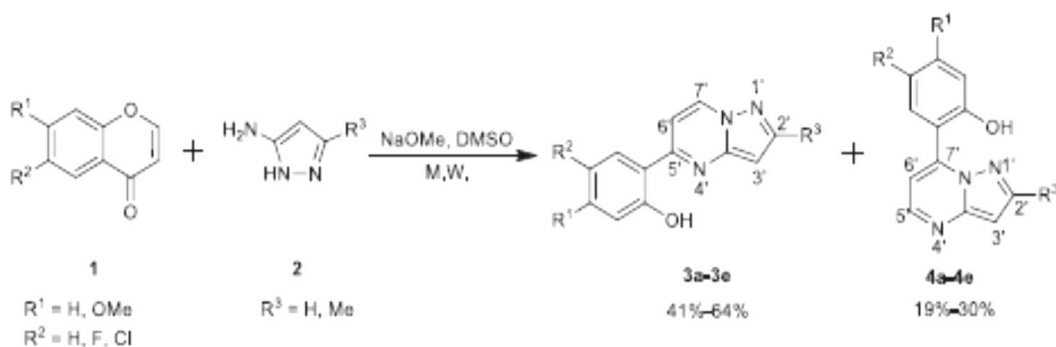
Scheme 16: Synthetic route of thieno[2,3-d] pyrimidin-4(3H)-one derivatives (2a–o)

Pathan Naziyanaz B. et.al [36] reported synthesis of substituted ethyl 1,2,3,6- tetrahydro-4-methyl-2-oxo/thioxo-6-phenyl-1-(4,5-diphenyl-1- H-imidazol-2-yl) pyrimidine-5-carboxylates 3a–g by mixing thoroughly substituted ethyl 1-formyl-1,2,3,6-tetrahydro-4-methyl-6-phenyl-2-oxo/thioxopyrimidine-5-carboxylates 1a–g (25 mmol), Benzil 2 (25 mmol, 5.25 g), ammonium acetate (10 g) and acidic alumina (1.0 g) in an agate mortar followed by addition of 2-drops of glacial acetic acid. The beaker was kept in a microwave oven at maximum power level for 8 min with intermittent interval. (Scheme 17). Synthesized compounds were screened for their in vitro antibacterial activity against *Staphylococcus aureus*, *Salmonella typhi*, *Pseudomonas aurogenosa* and *Klebsiella pneumoniae* and also antifungal activity against the opportunistic pathogens *Candida albicans*.



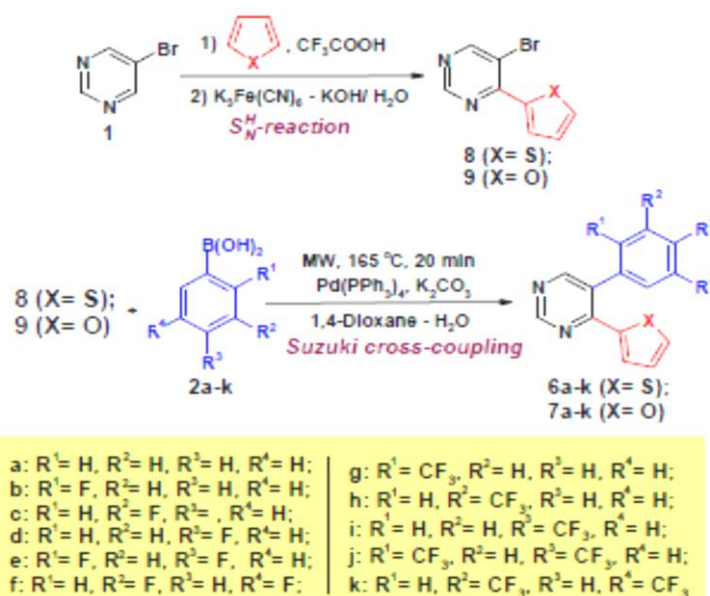
Scheme 17: Synthetic route of ethyl 1,2,3,6- tetrahydro-4-methyl-2-oxo/thioxo-6-phenyl-1-(4,5-diphenyl-1- H-imidazol-2-yl) pyrimidine-5-carboxylates 3a–g

Zhang Jin et al. [37] reported synthesis of 2-(pyrazolo[1,5-a] pyrimidin-5-yl)phenols (3) and 2-(pyrazolo[1,5-a]pyrimidin-5-yl) phenols (4) by dissolving chromone (1; 1 mmol), 3-aminopyrazole (2a; 1.5 mmol) or 3- amino-5-methyl pyrazole (2b; 2 mmol), and CH<sub>3</sub>ONa (135 mg, 2.5 mmol) in dried DMSO (15 mL). The resulting mixture was heated under microwave irradiation for 10 min at 100°C. (Scheme 18). The antifungal abilities of the obtained products 3 and 4 were evaluated against five phytopathogenic fungi (*Cytospora* sp., *Colletotrichum gloeosporioides*, *Botrytis cinerea*, *Alternaria solani* and *Fusarium solani*) and exhibited good antifungal abilities against *Colletotrichum gloeosporioides*.



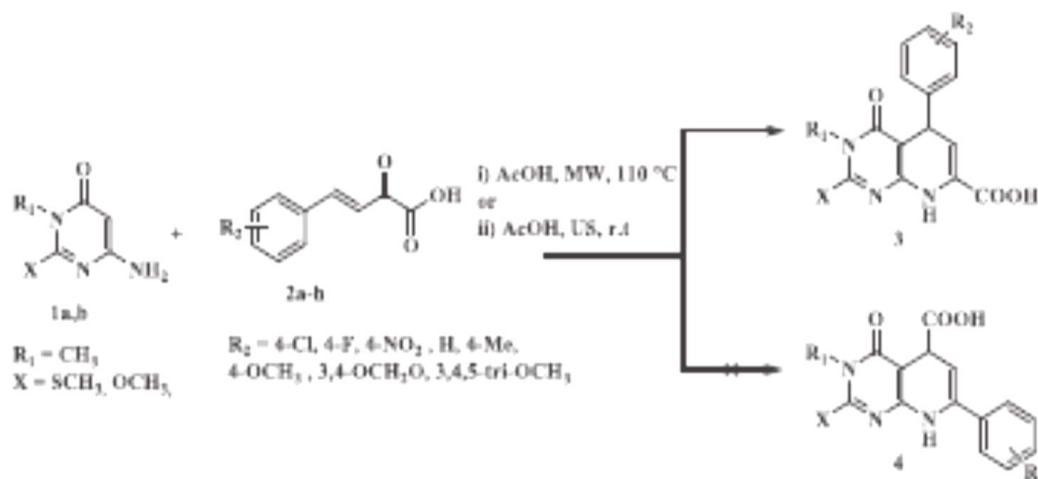
Scheme 18: General synthetic route for compounds 2-(pyrazolo[1,5-a] pyrimidin-5-yl)phenols (3) and 2-(pyrazolo[1,5-a]pyrimidin-5-yl) phenols (4)

VerbitskiyEgor V., et.al [38]reported synthesis of 4-(hetero)aryl-5-phenylpyrimidines 6a and 7a 5- (fluoroaryl)-4-(hetero)arylpymidines 6b-k and 7b-k by reaction of a solution of K<sub>2</sub>CO<sub>3</sub> (2.5 mmol) in 3 mL H<sub>2</sub>O and mixture of 5-bromo-4-(thien-2-yl)pyrimidine (8) [or 5- bromo-4-(furan-2-yl)pyrimidine(9)] (0.5 mmol),the corresponding arylboronic acid (2a-k) (0.6 mmol) and Pd(PPh<sub>3</sub>)<sub>4</sub> (29 mg, 5 mol %) in 1,4-dioxane (4 mL) and irradiating in a microwave apparatus at 165 °C (250 W) for 20 min. (Scheme 19). The antitubercular activity against Mycobacterium tuberculosis H37Rv has been explored. The outcome of the study disclose that, some of the compounds have showed promising activity in micromolar concentration against Mycobacterium tuberculosis H37Rv, Mycobacterium avium, Mycobacterium terrae, and multidrug-resistant strains isolated from tuberculosis patients in Ural region (Russia).



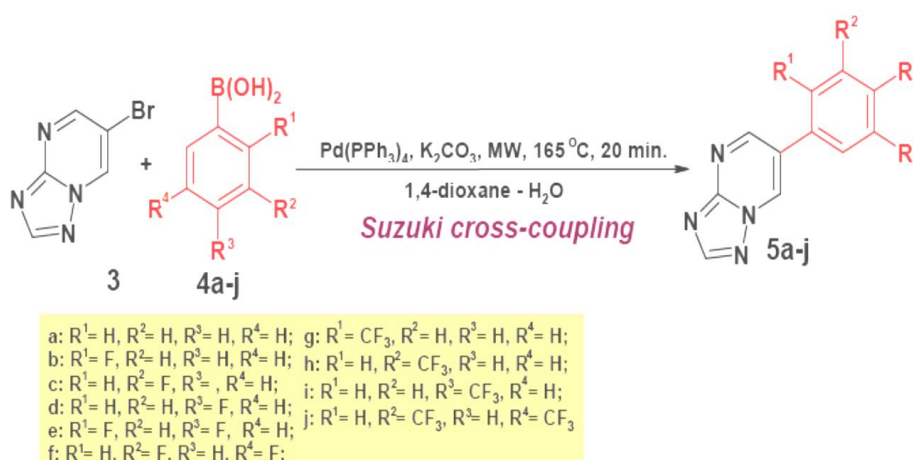
Scheme 19: Synthesis of 4-(hetero)aryl-5-phenylpyrimidines 6a and 7a 5- (fluoroaryl)-4-(hetero)aryl substituted pyrimidines 6b-k and 7b-k through the sequence of SNH and the Suzuki cross-coupling reactions

Quiroga Jairo et.al [39]reported synthesis of 5-aryl-4-oxo-3,4,5,8-tetrahydropyrido[2,3-d]pyrimidine-7-carboxylic acids 3 by reacting the mixture of 6-aminopyrimidines 1 (1 mmol),arylidene pyruvic acid 2 (1 mmol) and AcOH (1 mL) in microwave for 4 min at 110°C, power 100 W and pressure 30 psi (Scheme 20).The antioxidant properties, DPPH free radical scavenging, ORAC, and anodic potential oxidation of the new pyridopyrimidines were studied.



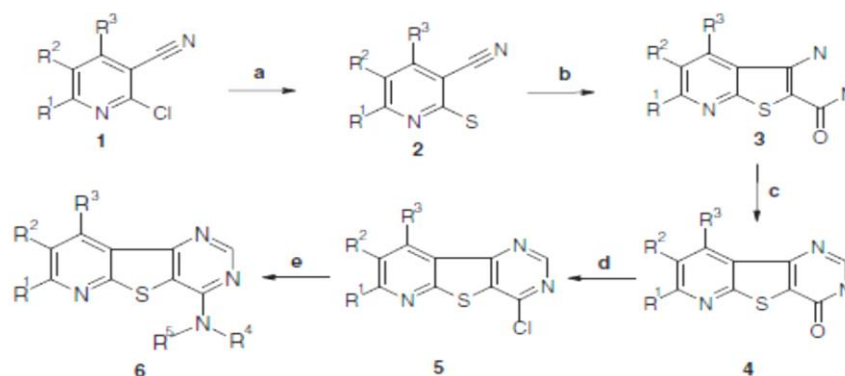
Scheme 20: Synthesis of 5-aryl-4-oxo-3,4,5,8-tetrahydropyrido[2,3-d]pyrimidine-7-carboxylic acids 3.

Verbitskiy Egor V. et.al [40] reported synthesis of novel 6-fluoroaryl-[1,2,4]triazolo[1,5-a]pyrimidines by adding  $\text{K}_2\text{CO}_3$  (346 mg, 2.5 mmol) in  $\text{H}_2\text{O}$  (3 mL) to a mixture of a 6-bromo-[1,2,4]triazolo[1,5-a]pyrimidine (3) (1.0 mmol), arylboronic (4a-j) acid (1.2 mmol) and  $\text{Pd}(\text{PPh}_3)_4$  (58 mg, 5 mol %) in 1,4-dioxane (4 mL). The resulting mixture was irradiated in a microwave apparatus at 165 °C (250 W) for 20 min. to afford the desired cross-coupling products (5a-j) (Scheme 21). The antimicrobial activity of new compounds was evaluated in vitro against Mycobacterium tuberculosis H37Rv and gram-negative (Neisseria gonorrhoeae ATCC 49226) bacteria.



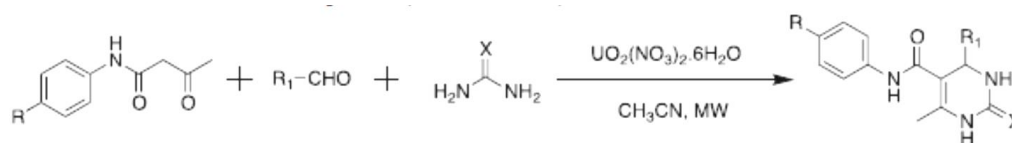
Scheme 21: Synthesis of 6-phenyl- and 6-(fluorophenyl)-[1,2,4]triazolo[1,5-a]pyrimidines (5a-j)

Zhao Chunlin et.al [41] reported synthesis of pyrido-thieno-pyrimidines by treatment of 2-chloro-nicotinonitrile 1 with sodium sulfide generated 2-mercapto-nicotinonitrile 2, which was reacted with 2-chloro-acetamide to give 3-amino-thieno-[2,3-b]-pyridine-2-carboxylic acid amide 3. Compound 4 was made by reacting 3 with trimethyl orthoformate with the catalysis of acid. Chlorination of 4 with phosphorus oxychloride followed by substitution of chlorine with amine  $\text{NHR}^4\text{R}^5$  gave the desired product 6 (Scheme 22).



Scheme 22: Synthesis of pyrido-thieno-pyrimidines. Reagents and conditions: (a)  $\text{Na}_2\text{S}$ , t-Butanol, microwave,  $150^\circ\text{C}$ , 20 min; (b) chloroacetamide,  $\text{K}_2\text{CO}_3/\text{EtOH}$ , microwave,  $120^\circ\text{C}$ , 10 min; (c)  $\text{HC}(\text{OCH}_3)_3, \text{CH}_3\text{COOH}$ , microwave,  $150^\circ\text{C}$ , 10 min; (d)  $\text{POCl}_3$ , refluxing; (e)  $\text{HNR}^4\text{R}^5$  microwave,  $120^\circ\text{C}$ , 20 min (when  $\text{R}^4$  and  $\text{R}^5$  are H, the reaction was run in a sealed tube at  $80^\circ\text{C}$  overnight).

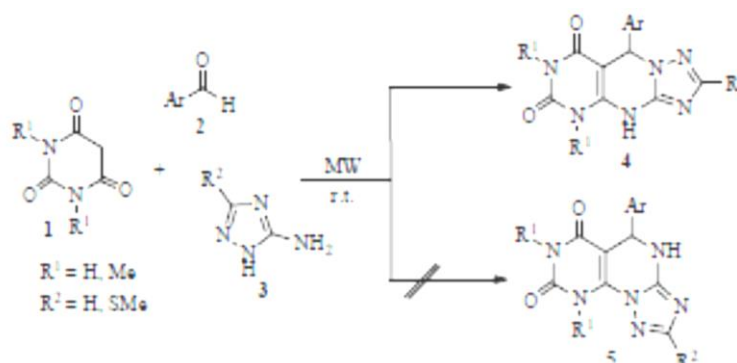
Venkatesan K. et.al [42] reported synthesis of 6-methyl-1,2,3,4-tetrahydro-N-aryl-2-thio/oxo-4-arylpyrimidine-5-carboxamide (4a-o) by reaction of mixture of aldehyde (1 mmol), acetoacetanilide (1 mmol), urea or thiourea (1.2 mmol),  $\text{UO}_2(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  (5 mol%) and acetonitrile (5 mL) and subjecting to microwave irradiation at an interval of 3 min at 160 W for about 15-18 min; varying time periods. (Scheme 23). The synthesized compounds 4a-o were screened for their in vitro antioxidant activity using DPPH free radical scavenging method and showed good antioxidant activity.



$\text{R} = \text{H, Cl, X} = \text{O, S; R}_1 = 2,4\text{-dichlorophenyl, 4-ethoxyphenyl, 3-ethoxy-4-hydroxyphenyl, 3-methoxy-4-hydroxyphenyl, 2-thiophene}$

Scheme 23: Facile one pot synthesis of 6-methyl-1,2,3,4-tetrahydro-N-aryl-2-oxo-4-arylpyrimidine-5-carboxamide derivatives (4a-o) catalyzed by  $\text{UO}_2(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$

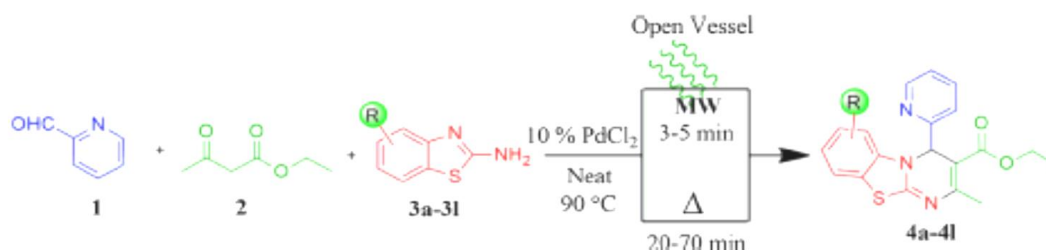
Farahi Mahnaz et.al [43] reported synthesis of 9-aryl-5,9-dihydropyrimido[4,5-d][1,2,4]triazolo[1,5-a]pyrimidine-6,8(4H,7H)-dione 4 by reaction of mixture of barbituric acid derivatives (1 mmol), arylaldehyde (1 mmol) and 3-amino-1H-1,2,4-triazoles (1 mmol) and irradiating at 300 W in a domestic MW oven for 10-15 min (Scheme 24).



Scheme 24: MW-assisted regioselective synthesis of 9-aryl-5,9-dihydropyrimido[4,5-d][1,2,4]triazolo[1,5-a]pyrimidine-6,8(4H,7H)-diones

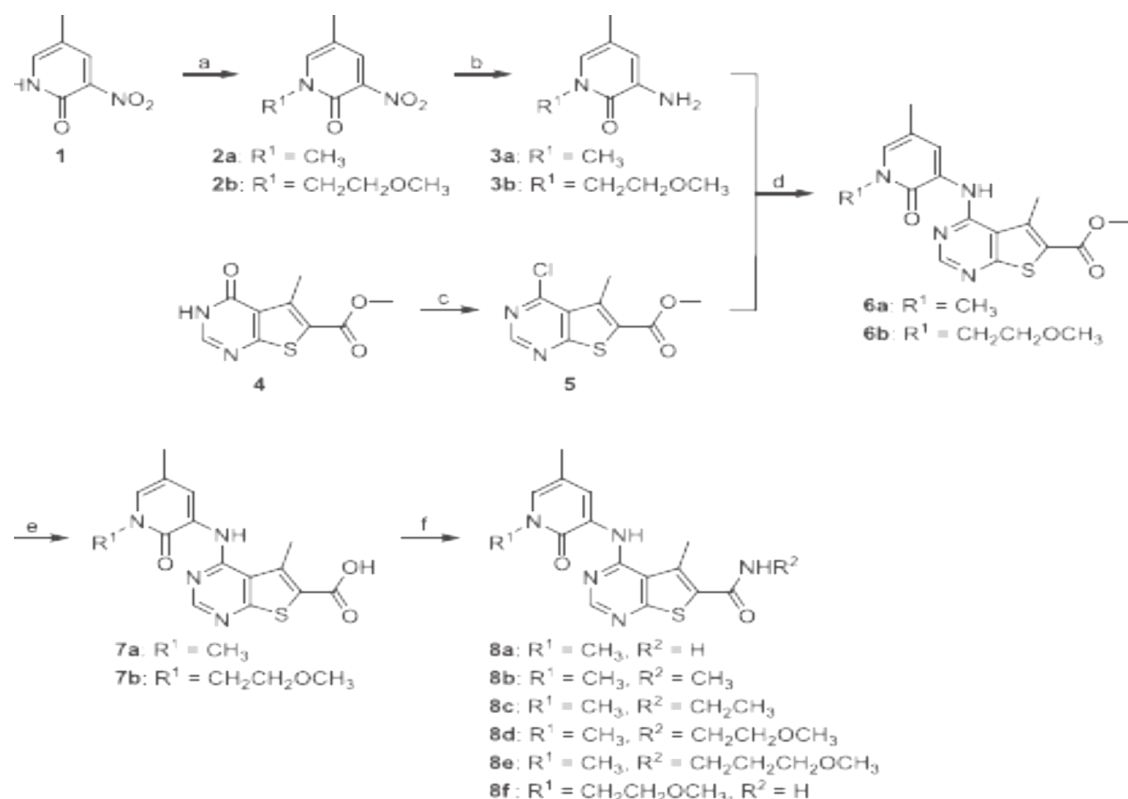
Bhoi Manoj N. et.al [44] reported synthesis of ethyl-(substituted)-2-methyl-4-(pyridin-2-yl)-4H-benzo[4,5]thiazolo[3,2-a]pyrimidine-3-carboxylate (4a-4l) by reacting mixture of

pyridine 2-aldehydes **1** (4.668 mmol),  $\beta$ -diketone like Ethyl acetoacetate **2** (4.668 mmol) and various derivatives of 2-amino benzothiazole **3a–3l** (4.668 mmol) in open vessel glass tube in the cavity of a microwave oven (CEM Discover microwave) and irradiating for 2–5 min at 90°C under solvent-free conditions by using PdCl<sub>2</sub> (10 mol%) (Scheme 25). The synthesized compounds were evaluated *in vitro* antibacterial and antioxidant activities. The antibacterial data revealed that the all synthesized compounds proved to be active against the test organism, two gram negative and two gram positive reference strains compared to standard drugs. Moreover, all the synthesized compounds were evaluated as antioxidants according to a DPPH radical scavenging activity, superoxide anion scavenging assay and ABTS+ radical scavenging activity of our synthesized compound. All compounds showed moderate to good antioxidant activity. The compounds were also screened for antitubercular activity against Mycobacterium tuberculosis H37RV strain and compounds showed good antitubercular activity.



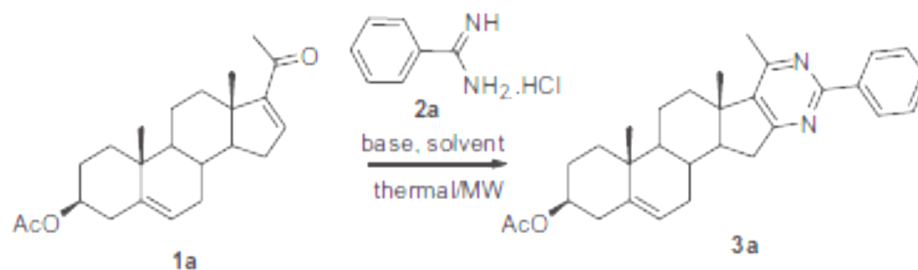
Scheme 25: Synthetic route for ethyl-(substituted)-2-methyl-4-(pyridin-2-yl)-4H-benzo[4,5]thiazolo[3,2-a]pyrimidine-3-carboxylate (**4a–4l**) under solvent-free condition reaction

Yu Mingfeng et.al [45] reported the synthesis of methyl 4-((1-alkyl-2-oxo-5-methyl-1,2-dihydropyridin-3-yl)amino)-5-methylthieno [2,3-d]pyrimidine-6-carboxylates (**6a** and **6b**) and their derivatives (**7a**, **7b** and **8a-f**). *N*-Alkylation of 5-methyl-3-nitropyridin-2(1H)-one **1** with iodomethane or 1-bromo-2-methoxyethane in the presence of potassium carbonate gave 1,5-dimethyl-3-nitropyridin-2(1H)-one **2a** or 1-(2-methoxyethyl)-5-methyl-3-nitropyridin-2(1H)-one **2b** respectively in good yields (**2a**: 95% and **2b**: 60%), which underwent palladium-catalysed hydrogenation to afford the corresponding 3-amino-1,5-dimethylpyridin-2(1H)-one **3a** or 3-amino-1-(2-methoxyethyl)-5-methylpyridin-2(1H)-one **3b** in excellent yields (**3a**: 94% and **3b**: 100%). Chlorination of methyl 5-methyl-4-oxo-3,4-dihydrothieno[2,3-d]pyrimidine-6-carboxylate **4** with thionyl chloride in the presence of *N,N*-dimethylformamide (DMF) yielded methyl 4-chloro-5-methylthieno[2,3-d]pyrimidine-6-carboxylate **5** in a yield of 89%. Each of amines **3a** and **3b** was individually coupled with chloride **5** in the presence of *p*-toluenesulfonic acid under microwave irradiation to give the corresponding methyl 4-((1-alkyl-2-oxo-5-methyl-1,2-dihydropyridin-3-yl)amino)-5-methylthieno[2,3-d]pyrimidine-6-carboxylates **6a** and **6b** in good yields (**6a**: 86% and **6b**: 54%), which were subsequently subjected to saponification in an alkaline methanolic solution to yield the carboxylic acids **7a** and **7b** respectively in excellent yields (**7a**: 100% and **7b**: 84%). Amination of carboxylic acids **7a** and **7b** with appropriate amines was effected in anhydrous DMF in the presence of the coupling reagent 1-[bis(dimethylamino)methylene]-1H-1,2,3-triazolo[4,5-b]pyridinium 3-oxid hexafluorophosphate (HATU) and *N,N*-diisopropylethylamine (DIPEA) to give the corresponding amides **8a-f** in moderate to good yields (36–80%) (Scheme 26). These derivatives are found to be as potent Mnk inhibitors. Phosphorylation of the eukaryotic initiation factor 4E (eIF4E) by mitogen-activated protein kinase (MAPK)-interacting kinases (Mnks) is essential for oncogenesis but unnecessary for normal development. Thus, pharmacological inhibition of Mnks may offer an effective and non-toxic anti-cancer therapeutic strategy.



Scheme 26: Synthesis of methyl 4-((1-alkyl-2-oxo-5-methyl-1,2-dihydropyridin-3-yl)amino)-5-methylthieno[2,3-d]pyrimidine-6-carboxylates (6a-b) and their derivatives (7a-b and 8a-f). Reagents and conditions: (a) appropriate alkyl halide,  $\text{K}_2\text{CO}_3$ , DMF,  $0^\circ\text{C}$  to rt for  $\text{CH}_3\text{I}$ ,  $0^\circ\text{C}$  to  $70^\circ\text{C}$  for  $\text{BrCH}_2\text{CH}_2\text{OCH}_3$ , o/n, 2a: 95%, 2b: 60%; (b)  $\text{H}_2$ , 10% Pd/C,  $\text{CH}_3\text{OH}$ , rt, o/n, 3a: 94%, 3b: 100%; (c)  $\text{SOCl}_2$ , DMF, reflux, 2 h, 89%; (d) TsOH,  $\text{H}_2\text{O}$ , 1,4-dioxane, microwave 200-300W,  $150^\circ\text{C}$ , 30 min, 6a: 86%, 6b: 54%; (e) 2 M NaOH,  $\text{CH}_3\text{OH}$ , reflux, 2 h, 7a: 100%, 7b: 84%; (f) appropriate amine, HATU, DIPEA, DMF,  $0^\circ\text{C}$  to rt, o/n, 8a: 80%, 8b: 36%; 8c: 66%, 8d: 58%; 8e: 37%, 8f: 66%.

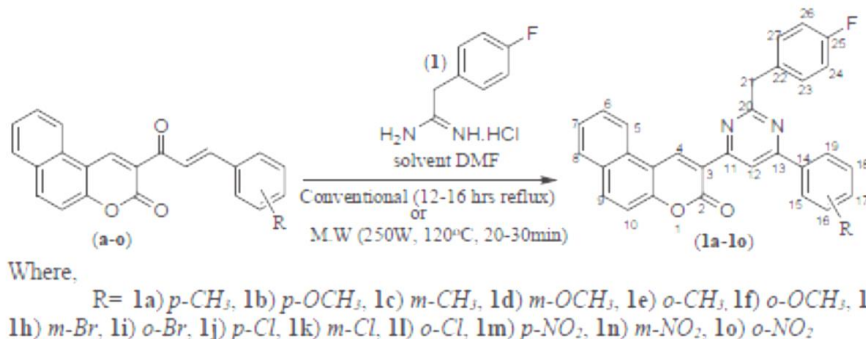
SaikiaPallabi, et.al [46] reported the synthesis of steroidal and nonsteroidal pyrimidines by refluxing 16-DPA (1a, 16-dehydropregnenolone acetate, 1.0 mmol) and benzamidine hydrochloride (2a, 1.0 mmol) in isopropanol in the presence of base NaOMe (two equivalents, 2.0 mmol) under microwave irradiation at  $110^\circ\text{C}/700\text{ W}$  for 10 min to furnish pyrimidine 3a in 26% yield. (Scheme 27).



Scheme 27: Synthetic route for steroidal and nonsteroidal pyrimidines

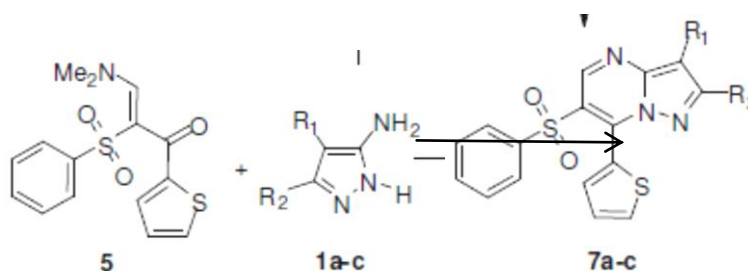
Reddy Dinesh S. et.al [47] reported the synthesis of 2-(2-(4-fluorobenzyl)-6-(substituted phenyl) pyrimidin-4-yl)-3H-benzo[f]chromen-3-one derivatives (1a-1o) by reaction of mixture of substituted chalconated benzocoumarin (0.01 mol) with 2-(4-fluorophenyl) acetamide hydrochloride (0.012 mol) and irradiating in a microwave reactor under 250W power at  $120^\circ\text{C}$  for 20-30 min in 5mL DMF (Scheme 28). Compounds (1a-1o) were evaluated for their in-vitro antitubercular activity while the most active compounds were further subjected for their cytotoxicity and DNA cleavage study. Results revealed that most

of the tested compounds displayed potent antitubercular activity with MIC in the range 0.05-2.81 mg/mL. Among them, compound (1b) possessed excellent activity (MIC 0.05 mg/mL) against M.tb H37Rv strain and exhibited low level of cytotoxicity against Vero cells, which suggested compound (1b) is found to be a promising lead for subsequent investigation in search of new antitubercular agents. DNA cleavage by gel electrophoresis method revealed that compounds (1b, 1g, 1k and 1n) were found to cleave the DNA completely.



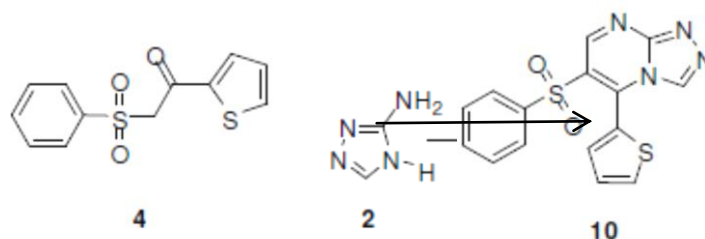
Scheme 28: Synthetic route for the preparation of benzocoumarin-pyrimidine hybrids (1a-1o)

Salema Mostafa E. et.al [48] reported the synthesis of pyrazolo[1,5-*a*]pyrimidine derivatives (7a-c) and triazolo[1,5-*a*]pyrimidine derivative (10). Pyrazolo[1,5-*a*]pyrimidine derivatives (7a-c) was synthesized by reaction of a mixture of the enaminosulfone 5 (3.21 g, 10 mmol) and the appropriate aminopyrazole derivative 1a-c (10 mmol) in pyridine (10 ml) and irradiating with microwaves under pressurized conditions (17.2 bar, 130°C) for 20 min. in 80-87% yield. (Scheme 29a).



Scheme 29a: Synthetic route for pyrazolo[1,5-*a*]pyrimidine derivatives (7a-c)

Synthesis of triazolo[1,5-*a*]pyrimidine derivative (10) was done by reaction of a mixture of the sulfone 4 (3.21g, 10 mmol) and 3-amino-1,2,4-triazole (2) and triethylorthoformate (10 mL) and irradiating with microwaves under pressurized conditions (17.2 bar, 100 °C) for 20 min. (Scheme 29b).

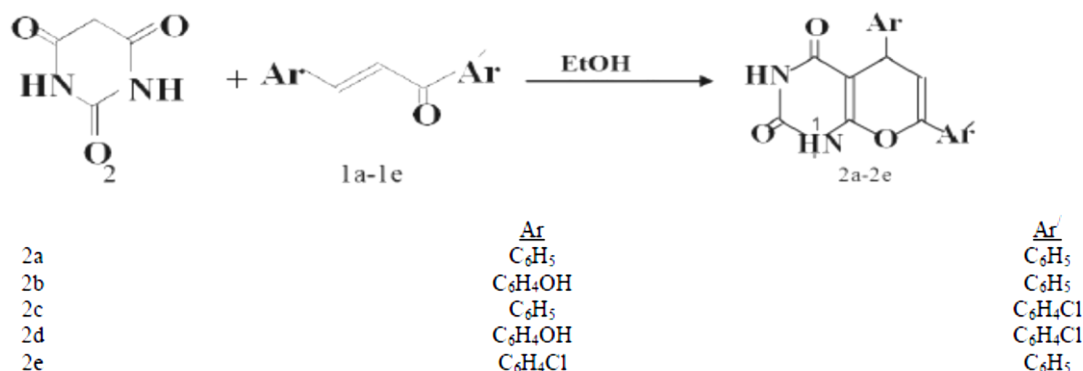


Scheme 29b: Synthetic route for triazolo[1,5-*a*]pyrimidine derivative (10).

Sattar M. A. et.al [49] reported the synthesis of Barbituric acid derivatives (2a-2e) by reaction of an equimolar mixture of barbituric acid (2) (0.005 mol) and



arylideneacetphenone (1a-1e) (0.005 mol) in 25 ml rectified spirit and 25 ml water. The mixture was irradiated with microwave at different power level for several minutes. (Scheme 30). The antimicrobial and cytotoxic activities of the synthesized compounds were also investigated. *Staphylococcus aureus*, *Bacillus megaterium*, *Escherichia coli* and *Pseudomonas aeruginosa* revealed the zone of inhibition were 6-12 mm where sample concentration was 100 µg/disc. However, cytotoxic analysis, the mortality 47-95% were appeared when sample concentration were 0.78-25 (µg/ml) and more than 50 (µg/ml) concentration showed 100% mortality. The presence of a reactive and unsaturated ketone function in synthesized compounds was found to be responsible for their potential antimicrobial and cytotoxic activity.



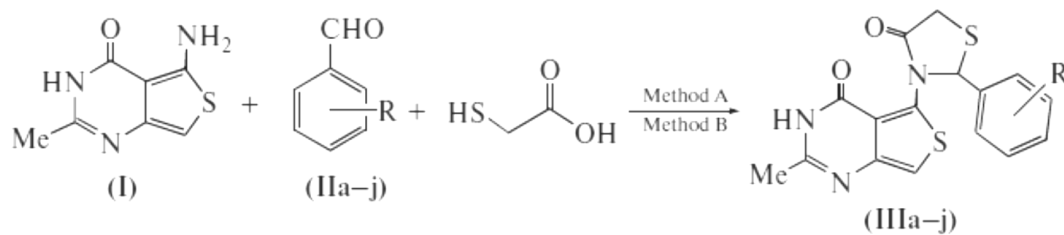
Scheme. 30: Synthetic route for barbituric acid derivatives (2a-2e).

Azabai. H. El et.al [50] reported the synthesis of new 2-methyl-5-(4-oxo-2-(substituted phenyl)thiazolidin-3-yl)thieno[3,4-d]-pyrimidin-4-one, 5-(2,7-diphenyl-5-thioxo-5,6,7,7a-tetrahydrothiazolo[4,5-d]pyrimidin-3(2H)-yl)-2-methylthieno[3,4-d]pyrimidin-4(3H)-one, and 2-methyl-5-(5-phenyl-thiazolo[5,4-d]isoxazol-6(5H)-yl)thieno[3,4-d]pyrimidin-4(3H)-one derivatives under microwave-assisted conditions. The new compounds were screened for their in vitro antimicrobial activity against two gram-positive bacteria (*Bacillus subtilis* NCIM-2063 and *Staphylococcus aureus* NCIM-2901), one Gram-negative bacteria (*Escherichia coli* NCIM-2256), and three fungal strains (*Candida albicans* NCIM-3471, *Aspergillus flavus* NCIM-539, and *Aspergillus niger* NCIM-1196) and showed promising biological activity.

Synthesis of 2-Methyl-5-(4-Oxo-2- (Substituted Phenyl)-Thiazolidin-3-yl)thieno[3,4-d]pyrimidin-4-one (IIIa-j) was done by reaction of an equimolar amounts of substituted aldehydes (IIa-j) (1 mmol), 5-amino-2-methyl-thieno[3,4-d]pyrimidin-4-one (I) (1 mmol), and thioglycolic acid (1 mmol) in 50 mL dimethylformamide containing 2–3 drops of glacial acetic acid and irradiating in MW (800 W) at 105°C for 6–9 min (Scheme 31a).

Synthesis of 5-(2,7-Diphenyl-5-Thioxo-5,6,7,7a-Tetrahydro- thiazolo[4,5-d]pyrimidin-3(2H)-yl)-2- Methylthieno[3,4-d]pyrimidin-4(3H)-one (IV) was done by reaction of an equimolar amounts of benzaldehyde (IIa) (1 mmol), 2-methyl- 5-(4-oxo-2-phenylthiazolidin-3-yl)thieno[3,4-d]pyrimidin-4-one (IIIa) (1 mmol), and thiourea (1 mmol) in 50 mL dimethylformamide in presence of 2–3 drops of glacial acetic acid and irradiating in MW (800 W) at 105°C for 9 min (Scheme 31b).

Synthesis of 2-Methyl-5-(5- Phenylthiazolo[5,4-d]isoxazol-6(5H)-yl)thieno[3,4-d]pyrimidin-4(3H)-one (V) was done by reaction of an equimolar amounts of dimethylformamide–dimethylacetal (DMF– DMA) (1 mmol), 2-methyl-5-(4-oxo-2-phenylthiazolidin-3-yl)thieno[3,4-d]pyrimidin-4-one (IIIa) (1 mmol), and hydroxylamine (1 mmol) in 50 mL dimethylformamide containing 2–3 drops of glacial acetic acid and irradiating in MW (800 W) at 105°C for 10 min (Scheme 31c).

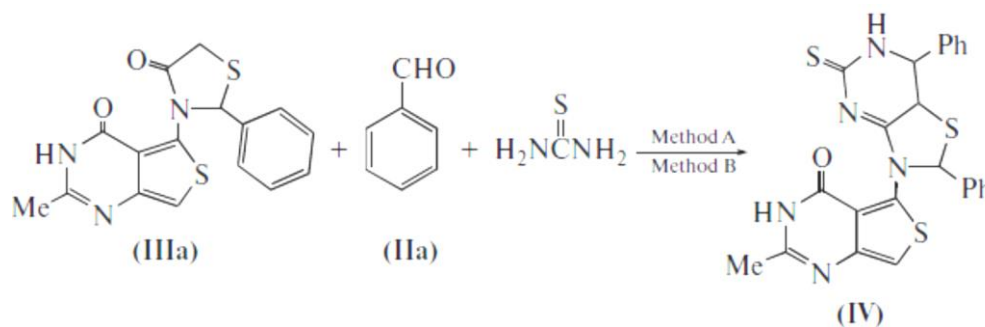


II, III	R	II, III	R
a	H	f	2,4-Methoxy
b	4-Hydroxy	g	2-Chloro
c	2-Nitro	h	4-Chloro
d	4-Nitro	i	4-CN
e	4-Methoxy	j	4-F

Scheme 31a: Synthesis of 2-methyl-5-(4-oxo-2-(substituted phenyl)thiazolidin-3-yl)thieno[3,4-d]pyrimidin-4-one (IIIa-j)

Method A. Microwave-assisted synthesis: AcOH, DMF as a solvent, 110°C, 8–10 min.

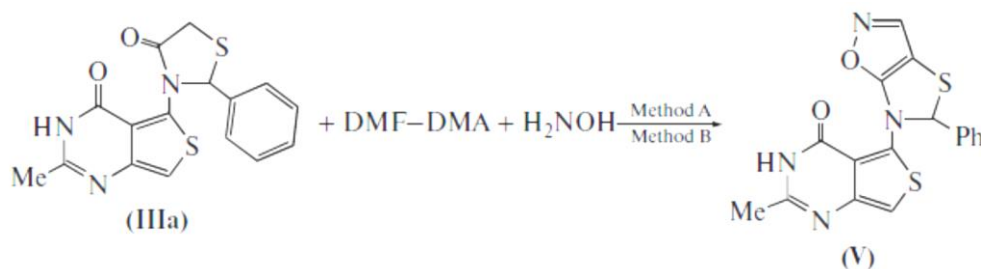
Method B. Conventional synthesis: AcOH, DMF as a solvent, reflux 4–6 h.



Scheme 31b: Synthesis of 5-(2,7-diphenyl-5-thioxo-5,6,7,7a-tetrahydrothiazolo[4,5-d]pyrimidin-3(2H)-yl)-2-methylthieno[3,4-d]pyrimidin-4(3H)-one (IV).

Method A. Microwave assisted synthesis: AcOH, DMF as a solvent, 110°C, 9 min.

Method B. Conventional synthesis: dry EtOH, conc. hydrochloric acid as a solvent, reflux 6 h.

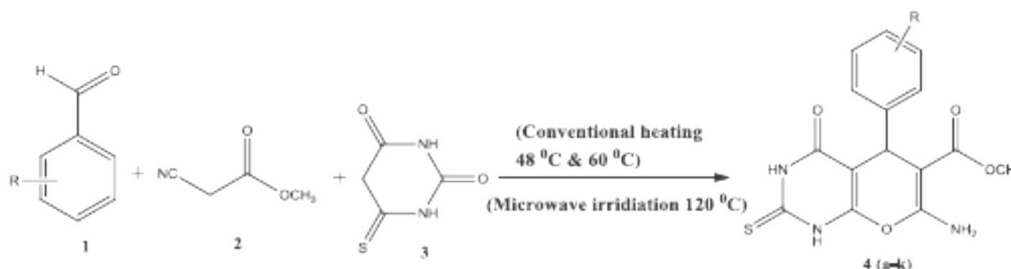


Scheme 31c: Synthesis of 2-methyl-5-(5-phenylthiazolo[5,4-d]isoxazol-6(5H)-yl)thieno[3,4-d]pyrimidin-4(3H)-one (V).

Method A. Microwave assisted synthesis: AcOH, DMF as a solvent, 110°C, 10 min.

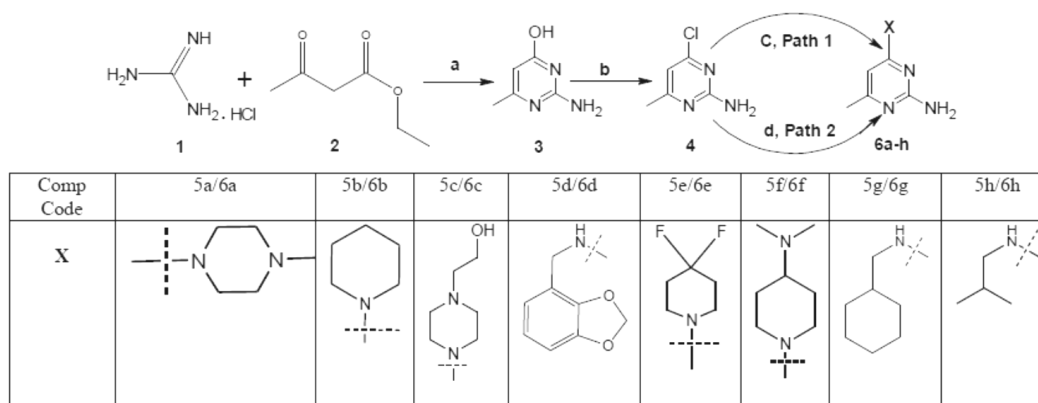
Method B. Conventional synthesis: dry xylene containing anhydrous K<sub>2</sub>CO<sub>3</sub> as a solvent, reflux 8 h.

BhatAjmal R., et.al [51] reported synthesis of new methyl 7-amino-4-oxo-5-phenyl-2-thioxo-2,3,4,5-tetrahydro-1H-pyrano[2,3-d] pyrimidine-6-carboxylate derivatives (4a–k) by reaction of a mixture of benzaldehyde derivatives 1 (1 mmol), methylcyanoacetate 2, (1.2 mmol), thio-barbituric acid 3, (1 mmol) and water (3.0 mL) and subjected to microwave irradiation under catalyst free conditions for a given time at power of 250W and 120°C. (Scheme 32). The synthesized compounds (4a–k) showed comparatively good in vitro antimicrobial and antifungal activities against different strains. The Compounds 4a, 4b, 4c, 4d, 4e and 4f showed maximum antimicrobial activity against *Staphylococcus aureus*, *Bacillus cereus* (gram-positive bacteria), *Escherichia coli*, *Klebsiella pneumoniae*, *Pseudomonas aeruginosa* (gram-negative bacteria). The synthesized compound antifungal activity against *Aspergillus Niger* and *Penicillium chrysogenum* strains. Streptomycin was used as a standard for bacterial studies and Mycostatin as standards for fungal studies.



Scheme 32: Microwave and conventional synthesis of methyl 7-amino-4-oxo-5-phenyl-2-thioxo-2,3,4,5-tetrahydro-1H-pyrano[2,3-d]pyrimidine-6-carboxylate derivatives (4a–k)

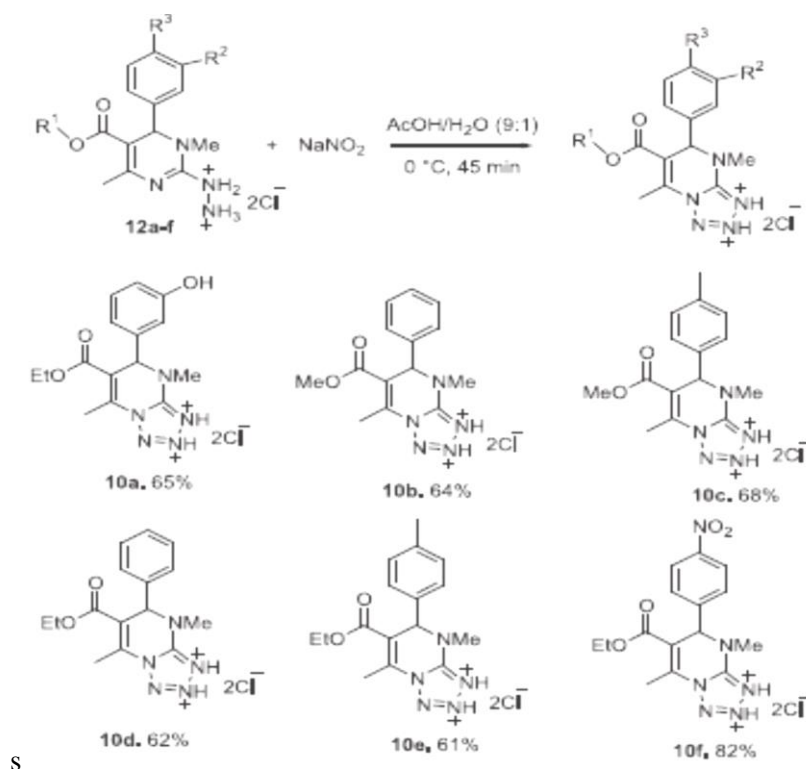
Kumar Manoj et.al [52] reported for the preparation of 2-Amino-6-substituted-pyrimidine derivatives (6a-h) by reaction of 2-amino-4-chloro-6-methylpyrimidine (4) (1.39 mmol), primary or secondary amines (5a-h) (1.66 mmol), Xantphos (0.027 mmol), Pd(dba)<sub>2</sub> (0.055 mmol), Cs<sub>2</sub>CO<sub>3</sub> (2.085 mmol) and freshly distilled dry acetonitrile (4 mL) dry microwave vial which was sealed and purged with a three cycles of vacuum / nitrogen and irradiated under microwave condition for 30 - 50 min for reaction to complete. (Scheme 33) The compounds were screened for their antimicrobial activity against pathogenic strains such as *S. aureus*, *E. coli*, *K. aerogenes*, *A. flavus* and *C. albicans* and anthelmintic activity conducted using *P. posthuma* (Indian Earthworm). Among the synthesized compounds 6a, 6c & 6f have shown significant antibacterial activity.



Scheme 33: Reagents and conditions: (a) Neat reaction (b) POCl<sub>3</sub>, Reflux (c) Path 1: Substituted primary and secondary amines (5a-h), Xantphos, Pd(dba)<sub>2</sub>, Cs<sub>2</sub>CO<sub>3</sub>, dry AcCN, Δ 80°C, 10-17 h; (d) Path 2: Substituted primary and secondary amines (5a-h), Xantphos, Pd(dba)<sub>2</sub>, Cs<sub>2</sub>CO<sub>3</sub>, dry AcCN, MW, 30-50 min.

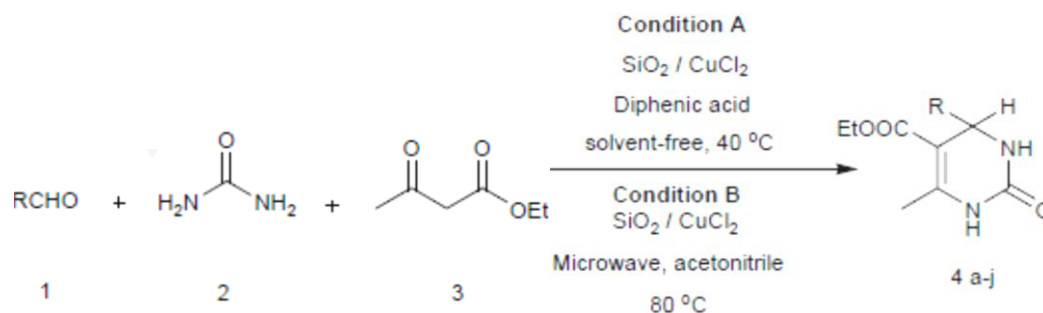
HashimJamshed et.al [53] reported the synthesis of tetrazolo dihydropyrimidines dihydrochloride (10a-f) by stirring reaction of 0.25 mmol of the corresponding hydrazinyl dihydropyrimidines dihydrochloride 12a-f and 69 mg (1.0 mmol, 4.0 equiv) of sodium

nitrite ( $\text{NaNO}_2$ ) in a microwave vial equipped with a magnetic stirrer and keeping it at  $0^\circ\text{C}$  in a  $\text{CaCl}_2$  ice bath followed by addition of 6 mL of 90% cold acetic acid in the dark and keeping the reaction mixture at  $0^\circ\text{C}$  for further 45 min under  $\text{N}_2$  atmosphere with continuous stirring to provide the desired 2-substituted dihydropyrimidines dihydrochloride (10a-f) as oils. (Scheme 34)



Scheme 34: Synthetic route for tetrazolo dihydropyrimidines dihydrochloride (10a-f)

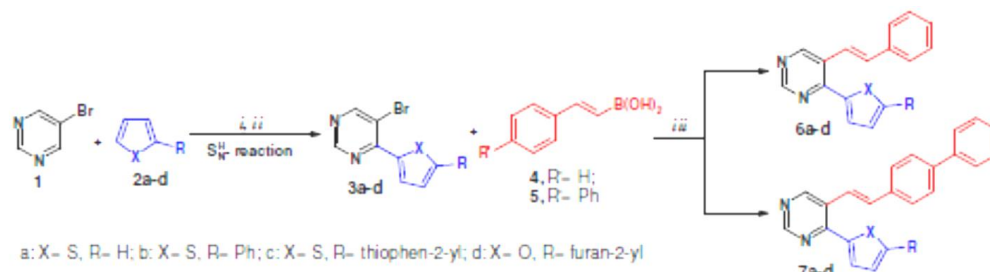
Kour Gurpreet, et al [54] reported the synthesis of 3,4-dihydropyrimidin-2(1H)-ones catalyzed by  $\text{SiO}_2 / \text{CuCl}_2$  by reaction of aldehyde (0.5 mmol), ethyl acetoacetate (0.5 mmol), urea (0.5 mmol) and  $\text{SiO}_2 / \text{CuCl}_2$  (0.05g) and stirring at  $80^\circ\text{C}$  in acetonitrile for appropriate time in a microwave. (Scheme 35)



Scheme 35: Synthetic route for 3,4-dihydropyrimidin-2(1H)-ones

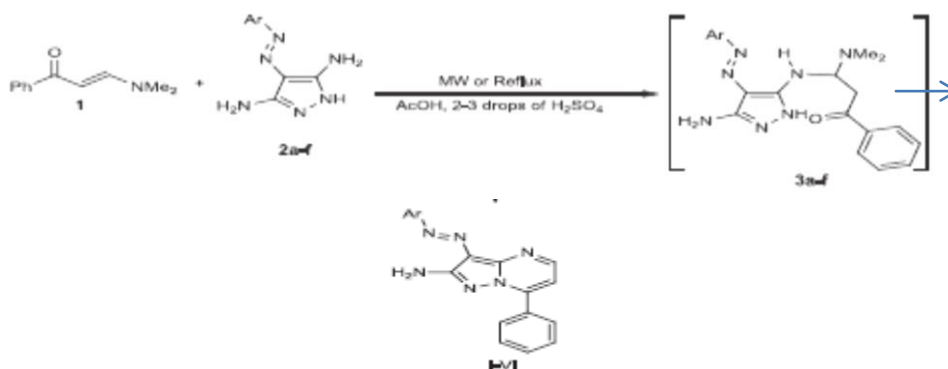
Kravchenko Marionella A. et al [55] reported the synthesis of 5-styryl-4-(hetero)arylpyrimidines 7a-d by the reaction of 5-bromopyrimidine (1) with thiophene (2a), bithiophene (2b) and 2-phenylthiophene (2c) in  $\text{CF}_3\text{COOH}$  to afford the corresponding  $\sigma\text{H}$ -adducts – 5-bromo-4-(hetero)aryl-3,4-dihydropyrimidinium salts. Further oxidation of these compounds with  $\text{K}_3\text{Fe}(\text{CN})_6$  in an aqueous solution of  $\text{KOH}$  gave 5-bromo-4-(hetero)arylpyrimidines 3a-c. The reaction 5-bromopyrimidine (1) with (E)-5-arylethenyl-4-(hetero)aryl-furan (2d) under the same conditions leads to the corresponding 5-bromo-4-(furyl-2'1)pyrimidine (3d) in a good yield (64%). Bromo derivatives 3a-d have been involved in the Suzuki reaction with trans-2-styrylboronic acids 4 and 5 under microwave

irradiation (155°C, 20 min). The corresponding cross-coupling products, pyrimidines 6a-d and 7a-d, have been obtained in high yields (Scheme 36). All intermediate 5-bromo-4-(hetero)aryl substituted pyrimidines and also the targeted 5-styryl-4-(hetero)arylpyrimidines were found to be active in micromolar concentrations in vitro against *Mycobacterium tuberculosis* H37Rv, avium, terrae, and multi-drug-resistant strain isolated from tuberculosis patients in Ural region (Russia). It has been found that some of these compounds possess a low toxicity and have a bacteriostatic effect, comparable and even higher with that of first-line antituberculosis drugs.



Scheme 36: Reagents and conditions: (i)  $\text{CF}_3\text{COOH}$ , r.t., 24 hours; (ii)  $\text{K}_3\text{Fe}(\text{CN})_6$ ,  $\text{KOH}$ ,  $\text{H}_2\text{O}$ , r.t., 6 hours; (iii)  $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{H}_2\text{O}$ ,  $\text{THF}$ , 155 °C, MW, 20 min.

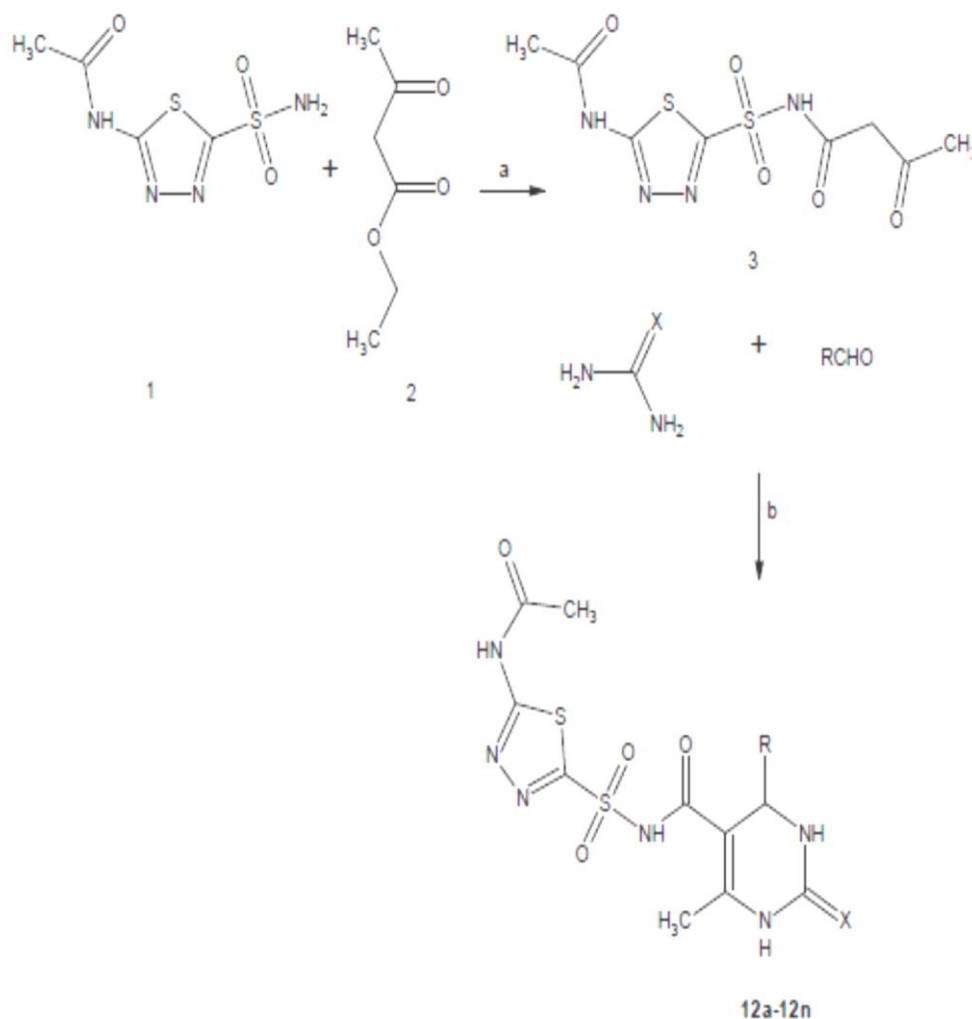
El-Gahami Mohamed A. et al [56] reported the synthesis of pyrazolo[1,5-a]pyrimidin-2-amine derivatives by reaction of E-3-dimethylamino-1-phenylprop-2-en-1-one (1) (2 mmol) and the appropriate 4-(Aryldiazenyl)-1H-pyrazole-3,5-diamine 2a-f (2 mmol) in glacial acetic acid (25 mL) and adding concentrated sulfuric acid is (0.5 ml) followed by microwave irradiation (300 W, 120°C) (Scheme 37). The UV-visible electronic spectra of some azo compounds of pyrazolo[1,5-a]pyrimidin-2-amine have been studied. The solvatochromic behavior of these compounds was investigated by studying their spectra in pure organic solvents of different polarities such as cyclohexane carbon tetrachloride, chloroform, ethanol and DMF. These exhibits a red shift in its  $\lambda_{\text{max}}$  with increase relative permittivity of medium changing from cyclohexane-carbon tetrachloride-chloroform-ethanol-DMF. The acid dissociation constants of these compounds were determined in aqueous-organic solvent mixtures such as acetone, methanol, ethanol and DMF. The ionization constants of the dyes in question depended largely on both the proportion and the nature of the organic solvent basicity contribute the major effects on the ionization process. In general,  $\text{pK}_a$  values in all compounds decrease with increase relative permittivity of the medium. The acidity of studied azo compounds increases in the following order:  $\text{p-NO}_2 < \text{m-CF}_3 < \text{p-F} < \text{p-Cl} < \text{p-H} < \text{p-CH}_3$ .



Scheme 37: Synthesis of pyrazolo[1,5-a]pyrimidin-2-amine derivatives.

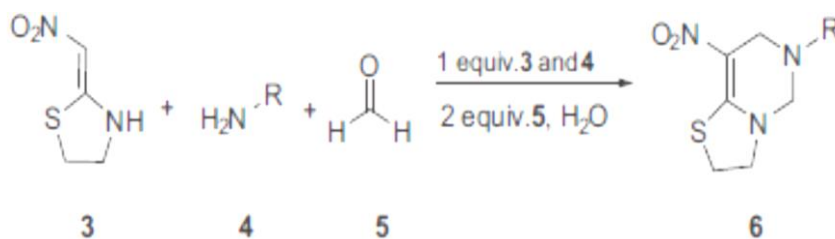
Elumalai Karthikeyan et al [57] reported the synthesis of 1,2,3,4-tetrahydropyrimidines (12a-n) by the reaction of mixture of N-[[5-(acetylamino)-1,3,4-thiadiazol-2-yl]sulfonyl]-3-oxobutanamide (0.005 M), urea/thiourea (0.0075 M), and appropriate aldehyde (0.005 M) with catalytic amount of p-toluenesulfonic acid in 10 ml of ethanol and subjecting to

microwave irradiation (300 W) for 10 min at the interval of 10s. (Scheme 38). The synthesized compounds were evaluated for in-vitro antimicrobial and cytotoxicity against *Bacillus subtilis*, *Escherichia coli* and Vero cells. The titled compounds exhibited weak, moderate, or high in-vitro antimicrobial and cytotoxicity. Compounds 12c, 12d, 12g and 12h, exhibited potential antimicrobial and in-vitro cytotoxicity.



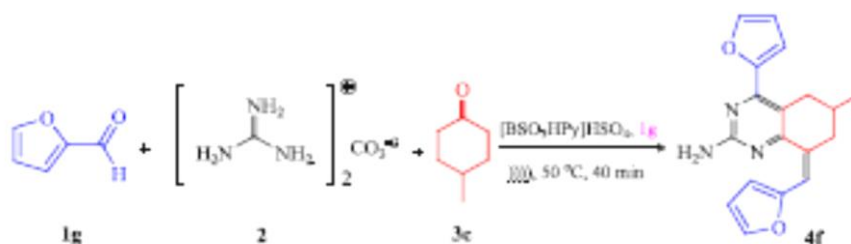
Scheme 38: Synthesis of compounds (12a-12n). R for compounds: 12a: (3-chlorophenyl); 12b: (3chlorophenyl); 12c: (4- chlorophenyl); 12d: (4-chlorophenyl); 12e:(3-nitrophenyl); 12f: (3-nitrophenyl); 12g: (4-fluorophenyl); 12h: (4-fluorophenyl); 12i: (2-nitrophenyl); 12j: (2-nitrophenyl); 12k: (2-chlorophenyl); 12l: (2-chlorophenyl); 12m: (4-pyridyl); 12n: (4-pyridyl). Reagents and conditions: (a) reflux 4.5 h,  $\text{CH}_3\text{COOH}$ ; (b)  $\text{C}_2\text{H}_5\text{OH}$ , p-toluenesulfonic acid, and microwave irradiation (300 W) for 10 min.

Yıldırım Muhammet et.al [58] reported the synthesis of thiazolo[3,2-c] pyrimidines (6) by reaction of 2-(Nitromethylene)thiazolidine 3 (0.5 or 0.25 mmol, 1.0 equiv), a primary amine 4 (0.5 or 0.25 mmol, 1.0 equiv) and formaldehyde 5 (37% v/v solution in water, 1.0 or 0.5 mmol, 2.0 equiv) were mixed in water (2 mL) and the resulting suspension was pre-stirred for 1 min in a microwave reactor prior to reaction. Then the reaction mixture was heated at exactly  $90^\circ\text{C}$  in the dynamic mode for 4-6 min. (Scheme 39)



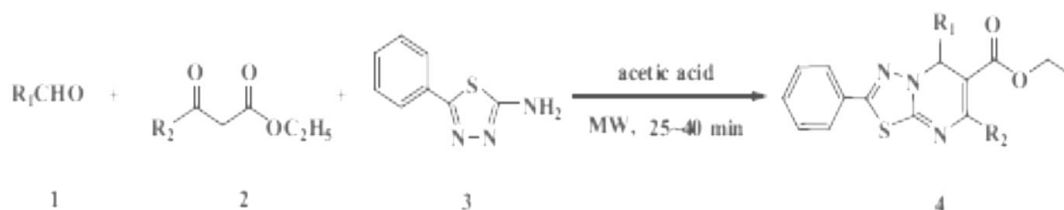
Scheme 39: Synthetic route for thiazolo[3,2-c] pyrimidines (6)

Zakeri Masoumeh et.al [59] reported the synthesis of quinazolin- and pyrimidine-2- amine 4f, by reaction of a mixture of 4-methyl cyclohexanon (2 mmol, 0.224 g), furfural (4 mmol, 0.384 g), guanidine carbonate (2 mmol, 0.180 g) and [BSO<sub>3</sub>HPy]HSO<sub>4</sub> (15 mol%) in 10 ml microwave reactor and heating to 120 °C. (Scheme 40)



Scheme 40: Synthetic route for quinazolin- and pyrimidine-2- amine 4f

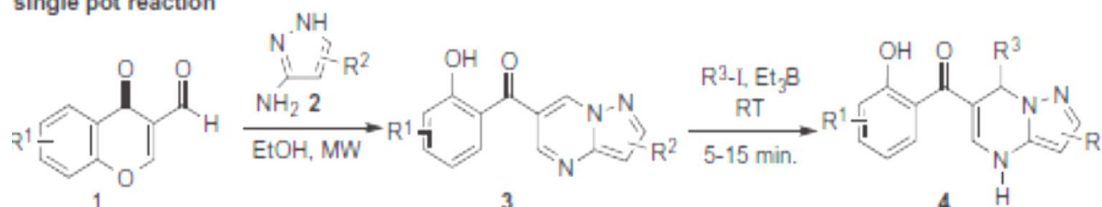
Zhao Bing et.al [60] reported the synthesis of novel 5H-[1,3,4] thiadiazolo[3,2-a]pyrimidine-6-carboxylate derivatives by reaction of mixture of 2-aminothiadiazole (1 mmol), appropriate aldehyde (1 mmol), ethyl acetoacetate (2 mmol) and 1.5 mL acetic acid in a pressurized microwave vial with snap on cap and subjected to microwave irradiation for appropriate time at 450W at 65°C. (Scheme 41)



Scheme 41: Synthetic route for 5H-[1,3,4] thiadiazolo[3,2-a]pyrimidine-6-carboxylate derivatives

Zimmerman Jake R. et.al [61] reported a two-step, single pot procedure for the synthesis of substituted dihydropyrazolo-pyrimidines starting with a microwave-assisted step to form pyrazolopyrimidine 3 followed by a tin-free radical addition to yield the substituted dihydropyrazolopyrimidine 4. (Scheme 42)

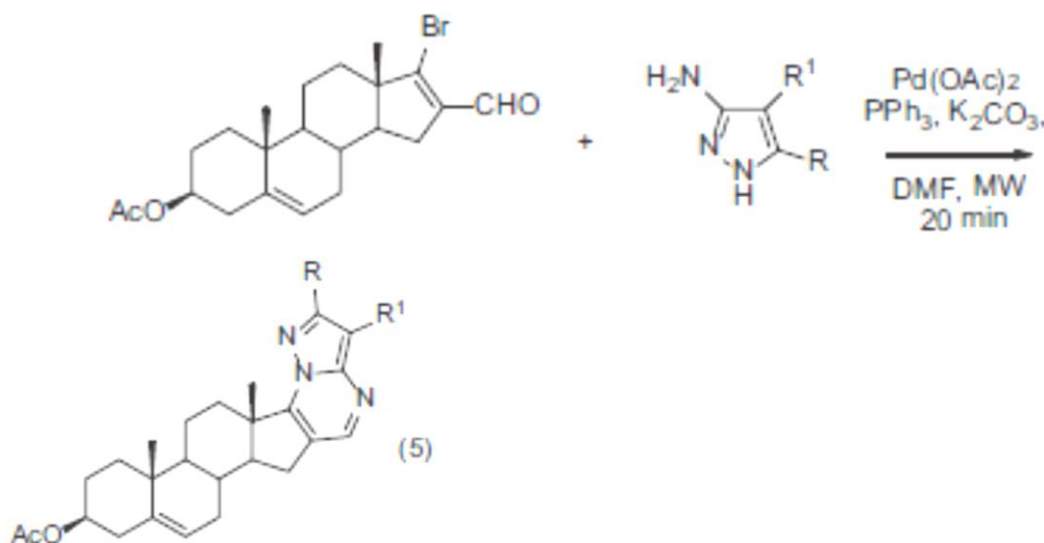
single pot reaction



Scheme 42: One-pot synthesis of substituted dihydropyrazolo-pyrimidines.

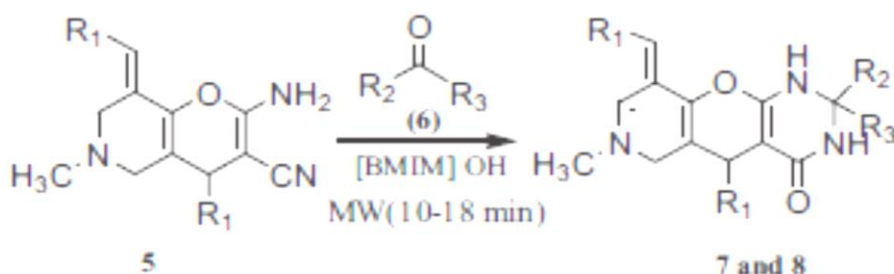
Shekarrao Kommuri et.al [62] reported the synthesis of steroidal D-ring fused pyrazolo[1,5-a]pyrimidines through irradiating a mixture of steroidal b-bromovinyl aldehydes (1.0 mmol)

with pyrazoloamines (1.0 mmol), palladium acetate (10 mol %) catalyst, triphenylphosphine (10 mol %), and  $K_2CO_3$  (1.2 mmol) in DMF and irradiating in microwave at 700W for 20 min. (Scheme 43)



Scheme 43: Synthesis of steroidal pyrazolo[1,5-a]pyrimidines and steroidal pyrimidines.

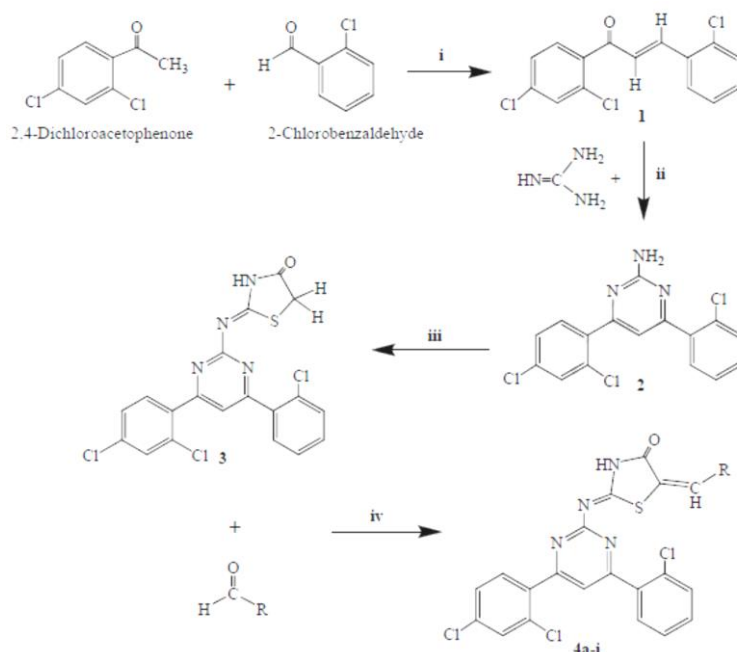
Siddiqui I.R. et.al [63] reported the synthesis of piperido[3',4':5,6] pyrano[2,3-d] pyrimidinone 7b catalyzed by basic ionic liquid [BMIM]OH by reaction of a solution of corresponding 2-amino-3-cyano-5,6,7,8-tetrahydro-4H-pyrano[3.2-c]piperidine 5a (1.0 mmol); cyclohexanone 6b (1.3 mmol) and [BMIM]OH (2 mL) in a reaction vial, sealed and irradiated at 90°C for 10 min (hold time) (Scheme 44).



Scheme 44: Synthesis of novel piperido[3';4':5,6]pyrano[2,3-d]pyrimidinone.

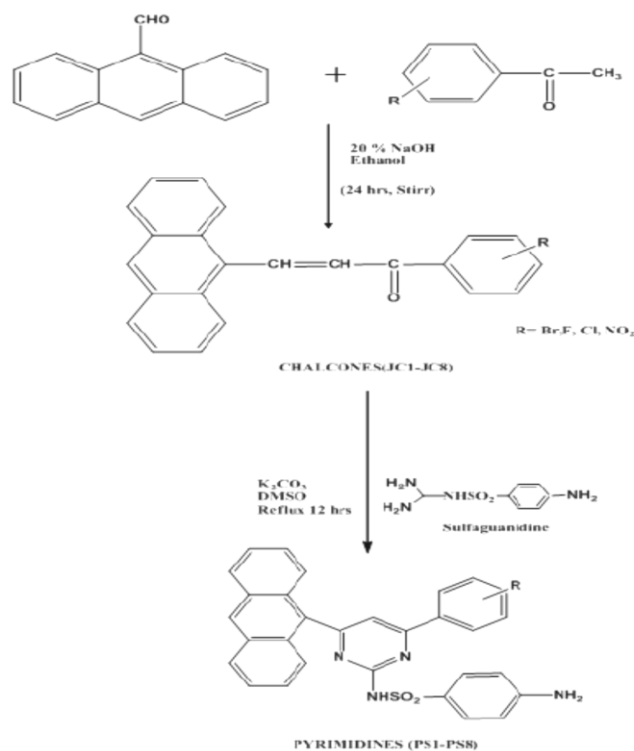
Rashid Mohd et.al [64] reported the synthesis of 2-(4-(2-chlorophenyl)-6-(2,4-dichlorophenyl) pyrimidin-2-yl-imino)-5-(substituted)thiazolidin-4-one (4a-i) by reacting compound 3 (0.01 mol, 4.49 mg) and different types of aromatic aldehydes (0.01 mol) in glacial acetic acid (20 mL) in presence of anhydrous sodium acetate (0.01 mol, 0.82 mg). The reaction mixture was placed into a microwave reaction vessel and kept into the scientific microwave synthesizer (model No. CATA-R, Catalyst systems, India) and then irradiated at a power level of 5 (50%, 350 W) for 10-15 min. The synthesized compounds were screened for their anticancer activities and these compounds were found to be new potential anticancer agents (Scheme 45).





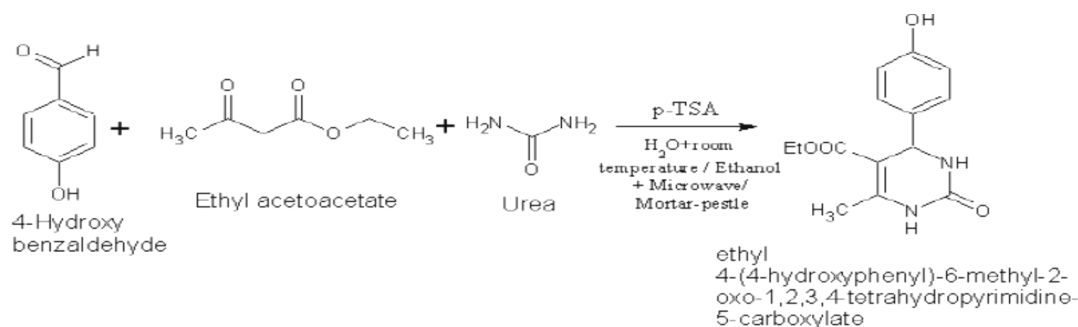
Scheme 45: Synthetic route for 2-(4-(2-chlorophenyl)-6-(2,4-dichlorophenyl) pyrimidin-2-yl-imino)-5-(substituted)thiazolidin-4-one (4a-i)

Jainey Pj et.al [65] reported the synthesis of 4-amino-N-(4-(anthracen-9-yl)-6-(aryl substituted) pyrimidin 2yl)benzene sulphonamide by reaction of chalcone (0.01 mol), sulfaguanidine (0.01 mol), and 20 ml dimethyl sulfoxide and anhydrous potassium carbonate and subjecting to microwave irradiation for 5-6 min. (Scheme 46). Biological activity studies showed that these compounds exhibited significant antitumor and antioxidant activities and showed good radical scavenging activity.



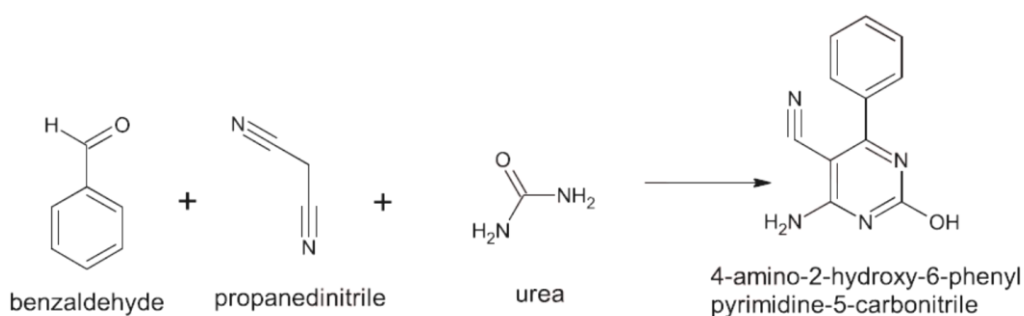
Scheme 46: Synthetic route for 4-amino-N-(4-(anthracen-9-yl)-6-(aryl substituted) pyrimidin 2yl)benzene sulphonamide

Jagwani D. et.al [66] reported the synthesis of 4-(4-Hydroxyphenyl)-6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylic acid ethyl ester by reaction of 4-Hydroxybenzaldehyde (0.122 gm, 0.001 mol), Ethylacetoacetate (0.127 ml, 0.001 mol) and Urea (0.132gm, 0.002 mol, ethanol (5 ml) as a solvent and p- TSA (0.034gm, 0.002mol) were added simultaneously to the above mixture. The mixture was irradiated for 3 minutes (30 second interval @ 20 power level). (Scheme 47)



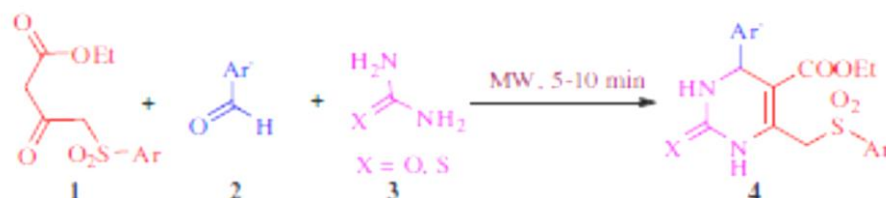
Scheme 47: Synthetic route for 4-(4-Hydroxyphenyl)-6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylic acid ethyl ester

Marabathuni V.Jhansipriya et.al [67] reported the synthesis of 4-Amino-2-Hydroxy-6-PhenylPyrimidine-5-Carbonitrile by reaction of substituted aldehyde (0.005 mol), malononitrile (0.005 mol), urea or thiourea (0.005mol) and conc.  $H_2SO_4$  (1-2 drops) in absolute ethanol (5 mL) and irradiating in the domestic type microwave oven 900 W with a frequency 2450 MHz (Kenstar OM-25 DCE, India) for 2-4 min (one pulse each of 30 sec) (Scheme 48). The synthesized compound was found to exhibit antibacterial activity.



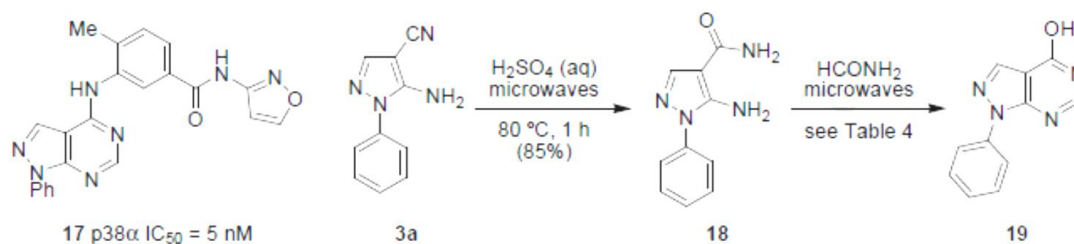
Scheme 48: Synthetic route for 4-Amino-2-Hydroxy-6-PhenylPyrimidine-5-Carbonitrile

Harikrishnan Palani Sokkan et.al [68] reported the synthesis of novel ethyl 2-oxo/thio-4-aryl-6-(arylsulfonylmethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylates by the Biginelli reaction of ethyl 3-oxo-4-(arylsulfonyl)butanoate 1, aromatic aldehyde 2 and diamide (urea/thiourea) 3 under microwave irradiation and solvent- and catalyst-free conditions (Scheme 1). The efficiency of this method has also been compared with reactions carried out in the presence of acid catalysts, viz. acetic acid, aluminium chloride, stannous chloride, cerium(III) chloride-7 $H_2O$ , ferric chloride and cupric acetate (Scheme 49).



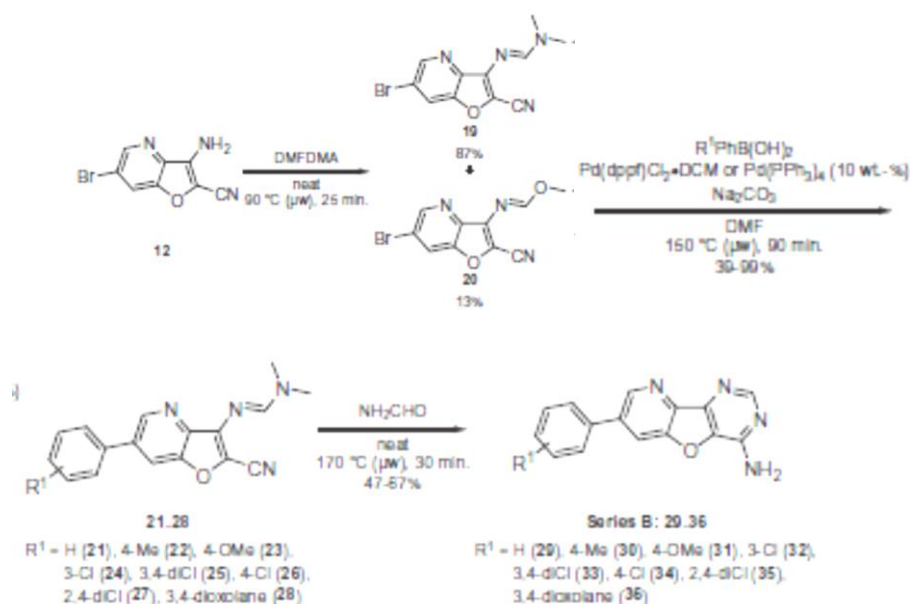
Scheme 49: Synthesis of ethyl 2-oxo/thio-4-aryl-6-(arylsulfonylmethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylates 4.

Bagley Mark C. et.al [69] reported the synthesis of pyrazolo[3,4-d]pyrimidines(19) by reaction of a mixture of 5-amino-1-phenyl-1H-pyrazole-4-carboxamide (18) (0.15 g, 0.74 mmol) in formamide (0.50 mL, 5.6 mmol) and irradiating at 150 °C for 45 min in a pressure-rated Pyrex tube (10 mL) using a CEM Discover microwave synthesizer by moderating the initial power (150 W). (Scheme 50) The synthesized compound was found to be inhibitor of mitogen activated protein kinase.



Scheme 50: Synthesis of pyrazolo[3,4-*d*]pyrimidine **18** from 5-aminopyrazole **3a**

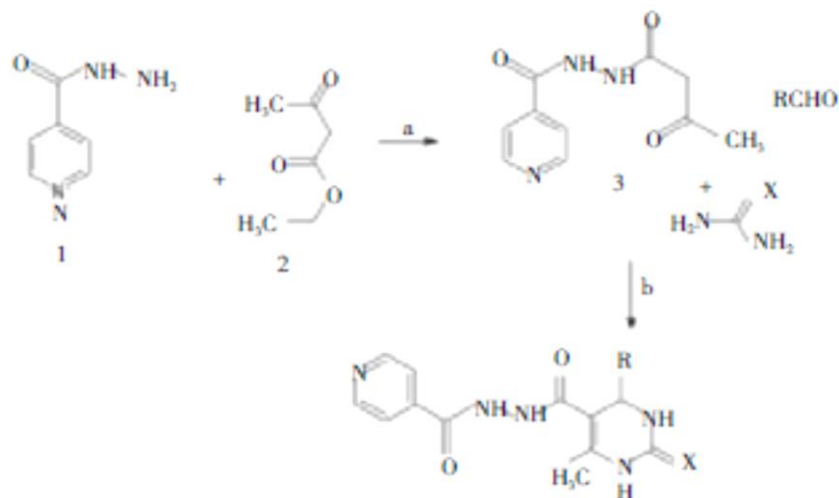
Deau Emmanuel et.al [70] reported the synthesis of 7-substituted pyrido[20,30:4,5]furo[3,2-*d*]pyrimidin-4-amines and their N-aryl analogues by reaction of precursor 12 with DMFDMA using microwave irradiation at atmospheric pressure, and converted into 6-bromodimethylformimidamide derivative 19 in 87% yield along with traces of the 6-bromoformimidate 20. Functionalization at position 7 of the intermediate 19 into the aryl adducts 21–28 was completed by a microwave-assisted Suzuki–Miyaura cross-coupling in the presence of a catalytic amount of [1,10- bis(diphenylphosphino)ferrocene]-dichloropalladium(II) dichloro methane complex (for cyanoamidines 21–27) or tetrakis(triphenylphosphine)palladium(0) (for cyanoamide 28), the appropriate phenylboronic acid, and sodium carbonate at 150°C (Scheme 51). Cyclisation of cyanoamide intermediates 21–28 gave the final 7-substituted pyrido[20,30:4,5]furo[3,2-*d*]pyrimidin-4-amines 29–36. The products were evaluated for their potent inhibition of a series of five Ser/Thr kinases (CDK5/p25, CK1 $\delta$ , CLK1, DYRK1A, GSK3 $\alpha/\beta$ ). and the compounds showed the inhibitory activity .



Scheme 51: Microwave-assisted synthesis of cyanoamidine 19 with DMFDMA, aryl adducts 21–28 by palladium-catalyzed Suzuki–Miyaura cross-coupling, and 7-substituted pyrido[20,30:4,5]furo[3,2-*d*]pyrimidin-4-amines 29–36 by formamide degradation.

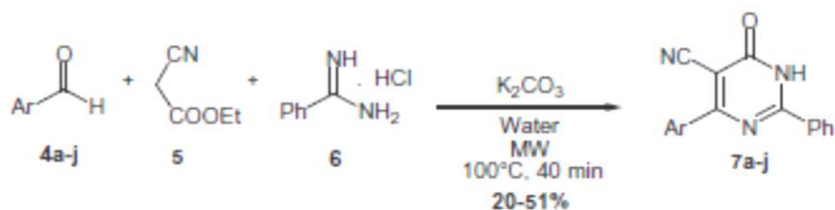
Elumalai Karthikeyan et.al [71] reported the synthesis of 1,2,3,4-tetrahydropyrimidines by microwave irradiation method (4a-n) by reaction of mixture of N' aceto acetyl isonicotino

hydrazide (0.005M), urea/thiourea (0.007 5 M), and appropriate aldehyde (0.005 M) with catalytic amount of benzenesulphonic acid in 10 ml of ethanol and subjected to microwave irradiation (300 W) for 8 min at the interval of 10 s (Scheme 52). The synthesized compounds were evaluated for in vitro antimicrobial and antimycobacterial activity against *Bacillus subtilis* (*B. subtilis*), *Escherichia coli* (*E. coli*), *Mycobacterium tuberculosis* (*M. tuberculosis*) CIP and H37Rv strain. Results: The titled compounds exhibited weak, moderate, or high antimicrobial and antimycobacterial activity.



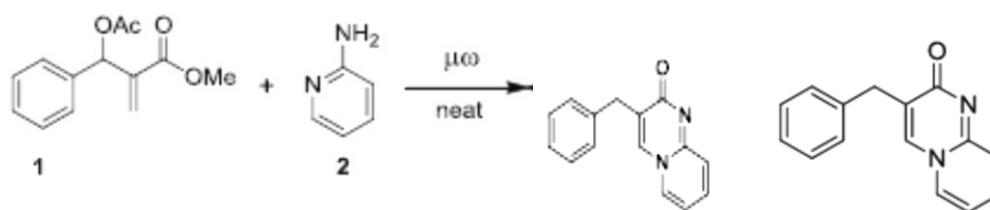
Scheme 52: Synthesis of 1,2,3,4-tetrahydropyrimidines (4a – 4n)

Xavier Augusto L. et.al [72] reported the synthesis of 6-oxo-2,4-diaryl-1,6-dihydropyrimidine-5-carbonitriles (7a–j) and 4-amino-2,6-diaryl-pyrimidine-5-carbonitriles (9a–j). by reaction of 1.13 mmol of benzamidine hydrochloride and 1.88 mmol of potassium carbonate dissolved in 10 mL of distilled water and stirring the mixture at room temperature until the neutralization of the benzamidine salt. To the clear basic solution, 0.94 mmol of the corresponding aromatic aldehyde (1.0 equiv) and 1.88 mmol (2.0 equiv) of ethyl cyanoacetate (for the synthesis of 7a–j) or malononitrile (for the synthesis of 9a–j) were added. The mixture was placed in a microwave reactor at temperature 100°C and 300W of initial power for 40 min. (Scheme 53)



Scheme 53: Three-component reaction of aromatic aldehydes, ethyl cyanoacetate, and benzamidine.

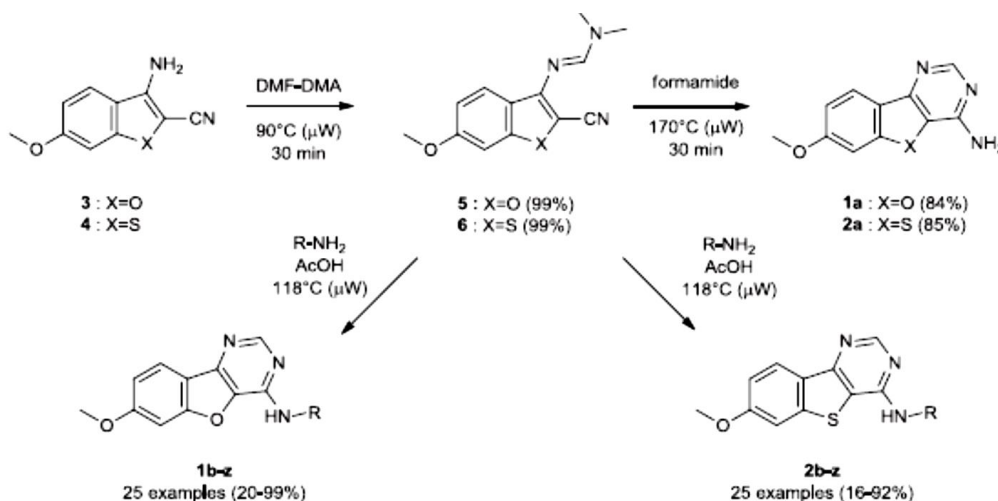
Satyanarayana S., et.al [73] reported the synthesis of 3-benzyl-2H-pyrido[1,2-a]pyrimidin-2-one derivatives by reaction of mixture of methyl 2-(acetoxymethyl)acrylate (1, 1 mmol), 2-heteroaryl amine (2, 1 mmol) in a 10 mL sealed glass tube followed by irradiation for a specified time as required to complete the reaction at 95 °C (temperature was monitored by inbuilt infrared sensor) at 150 W (Scheme 54).



Scheme 54: Synthetic route for 3-benzyl-2H-pyrido[1,2-a]pyrimidin-2-one derivatives

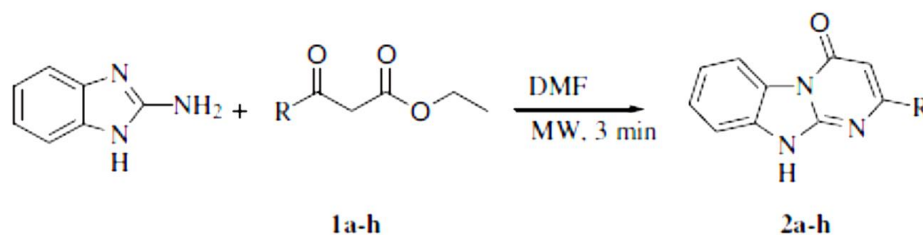
Loidreau Yvonnick et.al [74] reported the synthesis of N-aryl-7-methoxybenzo[b]furo[3,2-d]pyrimidin-4-amines and N-arylbenzo[b]thieno[3,2-d]pyrimidin-4-amine. Synthesis of N-aryl-7-methoxybenzofuro[3,2-d]pyrimidin-4-amines (1b-z) by reaction of a mixture of (E)-N0-(2-cyano-5-methoxybenzofuran-3-yl)-N,N-dimethylformimidamide(5) (0.1 g, 0.38 mmol) and appropriate aniline (1.0 equiv) in acetic acid (2 mL) and irradiating at 118 °C (400 W) (Scheme 55a).

Synthesis of N-aryl-7-methoxybenzothieno[3,2-d]pyrimidin-4-amines (2b-z) by reaction of a mixture of (E)-N0-(2-cyano-6-methoxybenzofuran-3-yl)-N,N-dimethylformimidamide (6) (0.1 g, 0.39 mmol) and appropriate aniline (1.0 equiv) in acetic acid (2 mL) and irradiating at 118°C (400 W) (Scheme 55b). The inhibitory potency of the final products against five protein kinases (CDK5/p25, CK1d/ε, GSK3a/b, DYRK1A and CLK1) was estimated. Compounds (2a-z) turned out to be particularly promising for the development of new pharmacological dual inhibitors of CLK1 and DYRK1A kinases.



(Scheme 55a) and (Scheme 55b): General synthesis of benzofuro[3,2-d]pyrimidines (1a-z) and benzothieno[3,2-d]pyrimidine (2a-z) derivatives.

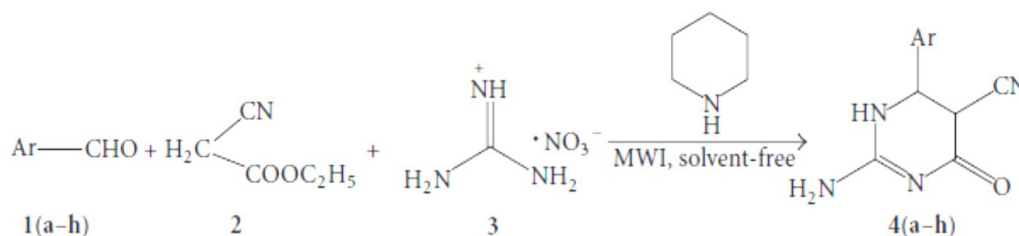
Puttaraju Kallimeledoddi B. et.al [75] reported the synthesis of 10H benzo[4,5]imidazo[1,2-a]pyrimidin-4-ones (2a-h) by reaction of an equimolar mixture of 2-aminobenzimidazole (0.5 g, 3.75 mmol) and b-ketoesters (1aeh) (3.75 mmol) in DMF (10 mL) in a microwave tube fitted with a reflux condenser and equipped with a magnetic stir bar and irradiating in a microwave reactor at a temperature of 130°C for 3 min at a maximum power of 320 W. The product was screened for their in vitro antibacterial activity against three Gram-positive bacteria viz., Staphylococcus aureus, Enterococcus faecalis, Streptococcus mutans and three Gram-negative bacteria viz., Escherichia coli, Klebsiella pneumonia, Pseudomonas aeruginosa and antifungal activity against Candida albicans, Aspergillus niger, Aspergillus fumigatus, Aspergillus flavus, Fusarium oxysporum, Penicillium chrysogenum and anti-cancer activity against Dalton's Ascitic Lymphoma (DAL) cell line. In general, all the compounds possessed better antifungal properties than antibacterial properties (Scheme 56).



R: a; *i*-Pr, b; 4-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, c; 3-FC<sub>6</sub>H<sub>4</sub>, d; CF<sub>3</sub>, e; C<sub>6</sub>H<sub>5</sub>, f; 4-FC<sub>6</sub>H<sub>4</sub>, g; 3-ClC<sub>6</sub>H<sub>4</sub>, h; 4-OCH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>

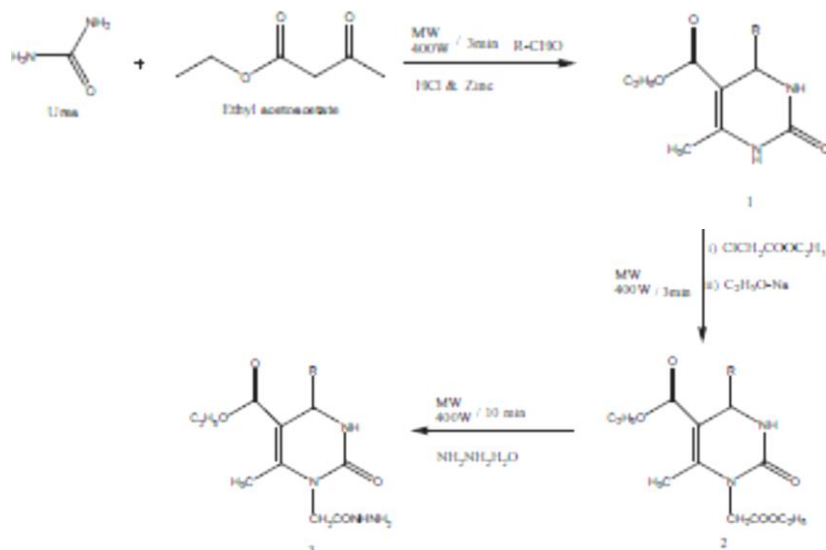
Scheme 56: Synthetic route for 10H benzo[4,5]imidazo[1,2-a]pyrimidin-4-ones (2a-h)

Bhatewara Anjna, et.al [76] reported the synthesis of 2-amino Dihydropyrimidinone Derivatives 4(a-h) by reaction of a mixture of aldehyde 1 (1mmol), ethyl cyanoacetate 2 (1.2mmol), guanidine nitrate 3 (1.5mmol), and 2-3 drops of piperidine subjecting to microwave irradiation at 60% power in 600W microwave oven for 5 min. (Scheme 57). The synthesized compounds showed a good anti-inflammatory, antibacterial, and antifungal activity.



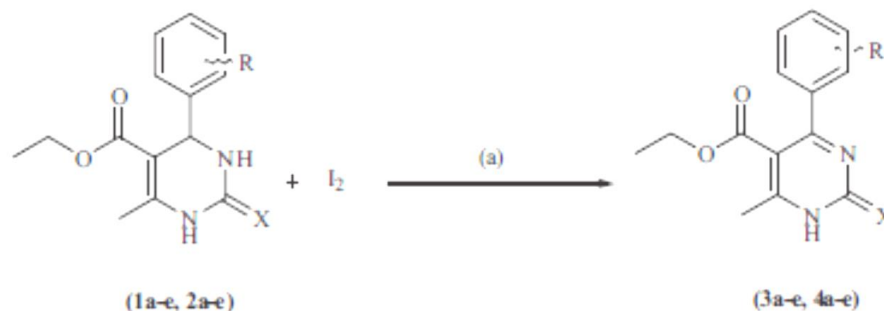
Scheme 57: Synthetic route for 2-amino Dihydropyrimidinone Derivatives 4(a-h)

Khamgaonkar Vishal D. [77] et.al reported the synthesis of Ethyl 1-[(ethoxycarbonyl) methyl]-1,2,3,4- tetrahydro-6-methyl-2-oxo-4-phenylpyrimidine-5- carboxylate (2a). The compound 1 (0.1 mol) was dissolved in a solution prepared by reacting Na (0.1 mol) with 200 ml of absolute ethanol. The solution was kept in microwave for 10 min at 200 W. Ethyl chloroacetate (0.1 mol) was then added in three portions over a period of 2 min. (Scheme 58). All compounds were evaluated for antibacterial activity by the broth microdilution assay method and found moderately active.



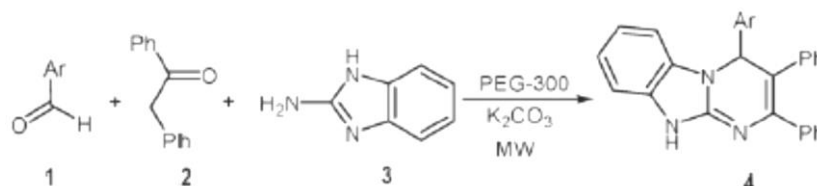
Scheme 58: Synthetic route for Ethyl 1-[(ethoxycarbonyl) methyl]-1,2,3,4- tetrahydro-6-methyl-2-oxo-4-phenylpyrimidine-5- carboxylate (2a).

Kodape Manisha M. et.al [78] reported facile aromatization of 3,4-dihydropyrimidin-2(1H)-ones using iodine in dimethyl sulfoxide under microwave irradiation at 120°C for 5 min(Scheme 59).



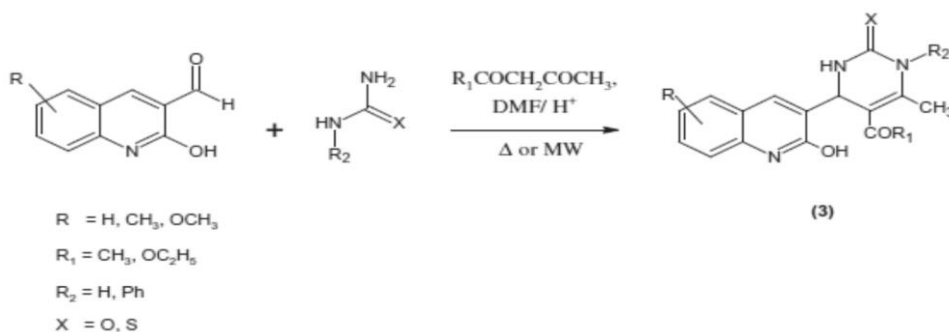
Scheme 59: Synthesis of 3,4-dihydropyrimidin-2(1H)-ones Reagent and condition: (a) I<sub>2</sub> (1:1), DMSO, MW, 120°C

WangShu-Liang, et.al [79] reported the one-pot synthesis of benzo[4,5]imidazo[1,2-a]pyrimidine derivatives under microwave irradiation conditions by reaction of aromatic aldehyde 1 (1 mmol), 1,2-diphenylethane-1-one 2 (1 mmol), 2-aminobenzimidazole 3 (1 mmol), potassium carbonate (0.4 mmol), and PEG-300 (2 mL) and irradiating at 120°C under microwave irradiation (initial power 100 W and maximum power 250 W) (Scheme 60).



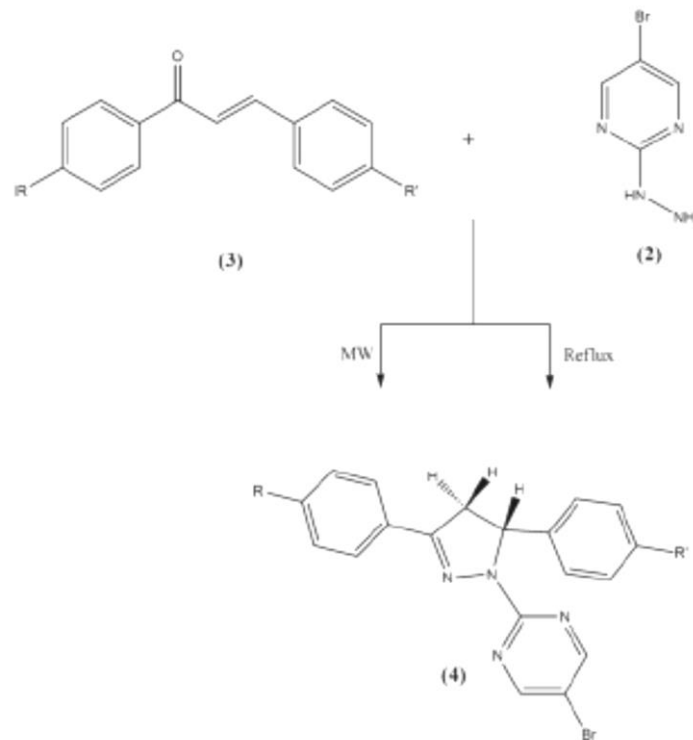
Scheme 60: Synthetic route for benzo[4,5]imidazo[1,2-a]pyrimidine derivatives

Adhikari Adithya, et.al [80] reported the synthesis of 1-substituted-4-(6-substituted-2-hydroxyquinoline-3-yl)-5-acetyl/carboxyethyl-6-methyl-pyrimidine-2-one/thiones (3a-l) by reaction of 6-Substituted-2-hydroxyquinoline-3-carbaldehyde (0.005 mol), ethyl acetoacetate/ acetylacetone (0.005 mol), urea/thiourea/ phenylthiourea (0.005 mol) and DMF (5 mL) and two drops of conc. H<sub>2</sub>SO<sub>4</sub> and subjecting to MW irradiation (160 W) (Scheme 61). Further these compounds were evaluated for their antioxidant, antifungal and antibacterial activities. Most of the compounds showed moderate to good activity when compared with standard.



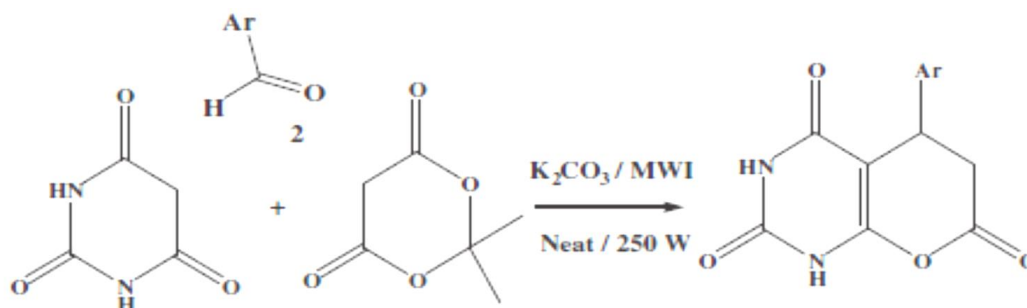
Scheme 61: Synthetic route for 1-substituted-4-(6-substituted-2-hydroxyquinoline-3-yl)-5-acetyl/carboxyethyl-6-methyl-pyrimidine-2-one/thiones

Adhikari Adithya et.al [81] reported the synthesis of 5-bromo-2-(3,5-diaryl-4,5-dihydro-1H-pyrazol-1-yl)pyrimidine by grounding substituted chalcones (3) (0.001 mol), 5-bromo-2-hydrazinylpyrimidine (2) (0.001 mol) and 5 drops of glacial acetic acid together in a mortar using a pestle for uniform mixing. This was taken in a 50 mL beaker and subjected to microwave irradiation (90 W) (Scheme 62). These new compounds were screened for their antioxidant, anti-inflammatory and analgesic activities. Some of these compounds exhibited potent biological activities compared to the standard drug.



Scheme 62: Synthetic route for 5-bromo-2-(3,5-diaryl-4,5-dihydro-1H-pyrazol-1-yl)pyrimidine

Azzam Sadeq Hamood Saleh et.al [82] reported the synthesis of novel pyrano[2,3-d]pyrimidine-2,4,7-triones in a pyrex cylindrical tube equipped with a magnetic stir bar, a mixture of aromatic aldehyde (2 mmol), Meldrums acid (2 mmol), barbituric acid (2 mmol), and  $K_2CO_3$  (10 mol %) then irradiating in a microwave reactor for (1 min) at  $100^\circ C/250W$  (Scheme 63).

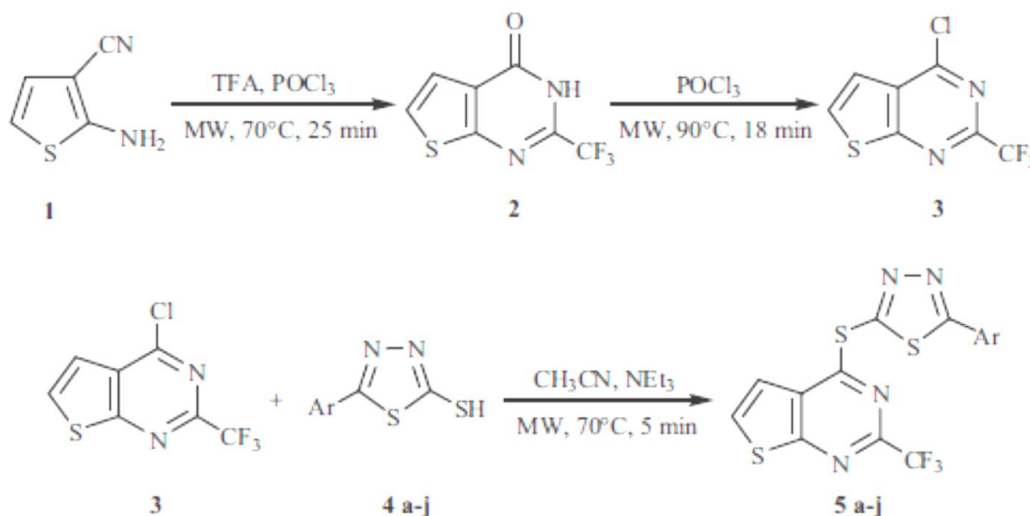


Scheme 63: Synthesis of 5-aryl-5,6-dihydro-1H-pyrano[2,3-d]pyrimidine-2,4,7-triones.

Song Xin Jian et.al [83] reported the synthesis of novel fluorinated thieno[2,3-d]pyrimidine derivatives containing 1,3,4-thiadiazole. Initially a mixture of 2-aminothiophene-3-carbonitrile 1 (10 mmol), trifluoroacetic acid (28 mL) and phosphorus oxychloride (2 mL) was irradiated at  $70^\circ C$  for 25 min by microwave to give 2-trifluoromethyl-3H-thieno[2,3-

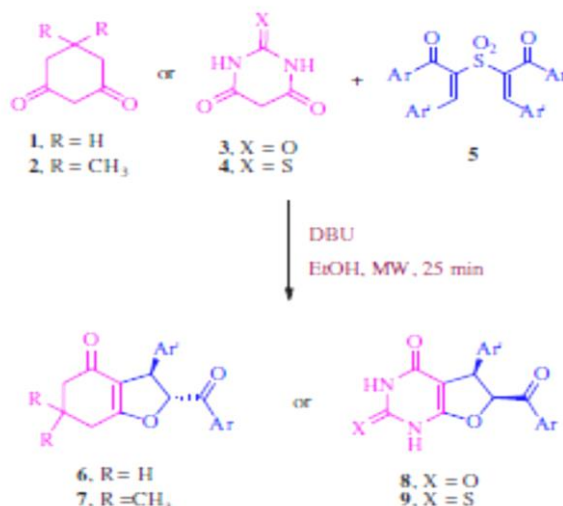


d]pyrimidin-4-one 2. Then a mixture of 2 (10 mmol) and phosphorus oxychloride (8 mL) was irradiated with microwave at 90°C for 18 min to afford 4-chloro-2-trifluoromethylthieno[2,3-d]-pyrimidine 3 followed by reaction of solution of 3 (5 mmol) and the appropriate 1,3,4-thiadiazole-2-thiol 4 (5 mmol), triethylamine (0.5 mL) in dried acetonitrile (15 mL) and irradiating to microwave irradiation for 5 min at 70°C to give compounds 5a-j (Scheme 64).



Scheme 64: Synthetic route of fluorinated thieno[2,3-d]pyrimidine derivatives 5a-j.

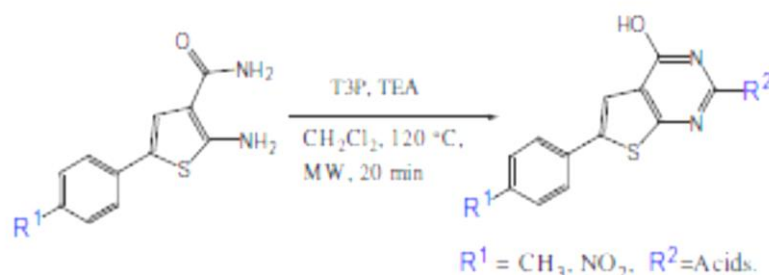
Vinosha Beermohamed et.al [84] reported the synthesis of trans-2,3-dihydrobenzofurans (6 and 7) and cis-5,6-dihydrofuro[2,3-d]pyrimidines (8 and 9) by reaction of a mixture of 2,20-sulfonylbis(1,3-diarylprop-2-en-1-ones) 5 (1 mmol), cyclic 1,3-diketones 1-4 (1 mmol) and DBU (0.5 mmol) in ethanol (5 ml) and subjecting to microwave irradiation microwave oven at 140°C, 50 W, 5 bar pressure and maximum absorption level (Scheme 65). This transformation presumably occurred via domino Michael addition-proton exchange-annulation via intramolecular displacement sequence.



Scheme 65: Synthesis of trans-2,3-dihydrobenzofurans (6 and 7) and cis-5,6-dihydrofuro[2,3-d]pyrimidines (8 and 9)

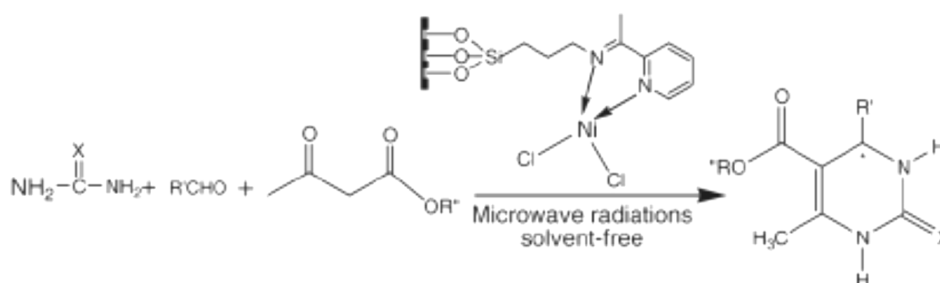
PoojariSubba, et.al [85] reported the synthesis of thieno [2,3-d] pyrimidin-4-ol derivatives mediated by polyphosphonic anhydride by reaction of a mixture of 2-amino-5-(4-methylphenyl) thiophene-3-carboxamide 1a (0.25 g, 0.107 mol) and 4-benzyloxy benzoic acid 4a (0.319 g, 0.140 mol) in anhydrous chloroform (4 ml) followed by addition of triethylamine (0.32 g, 0.323 mol) and phosphonic acid cyclic anhydride (1.02 g, 0.323 mol).

The reaction mixture was irradiated at 120°C in a microwave initiator for 30 min (Scheme 66).



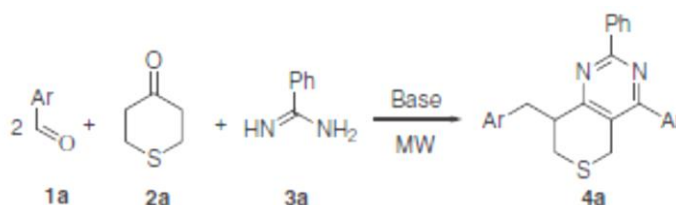
Scheme 66: Synthetic route for thieno [2,3-d] pyrimidin-4-ol derivatives

Sharma R.K. et.al [86] reported the one-pot synthesis of dihydropyrimidinones in presence of covalently anchored nickel complex on silica as catalyst under solvent-free conditions by reaction of a mixture of aldehyde (15 mmol), urea (20 mmol), ethyl acetoacetate (15 mmol), and catalyst (5 mol%) and irradiating in the microwave reactor (100 W) at 80°C for 5 minutes (Scheme 67).



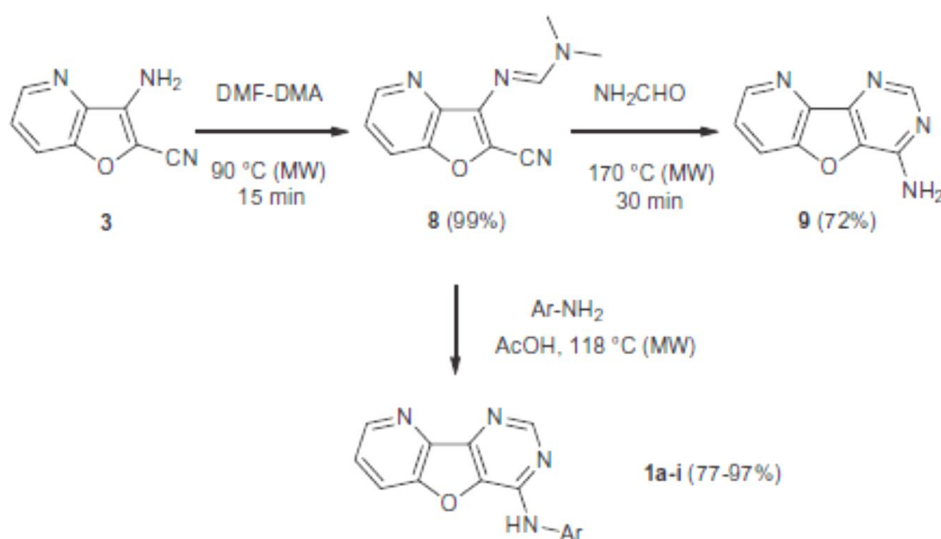
Scheme 67: One-pot synthesis of 3,4-dihydropyrimidinone catalyzed by covalently anchored nickel catalyst under microwave radiation.

Jiang Bo et.al [87] reported the synthesis of a series of new functionalized thiopyrano-, pyrano[4,3-d]pyrimidine derivatives with benzyl group residing in 8-position of fused pyrimidine nucleus were synthesized through multicomponent reactions of aromatic aldehydes, tetrahydrothiopyran-4-one (tetrahydropyran-4-one), and aryl amidines using *t*-BuOK as a base under microwave heating by stirring mixture of aromatic aldehyde 1 (2 mmol), tetrahydrothiopyran-4-one or tetrahydropyran-4-one 2 (1 mmol), *t*-BuOK (2.0 mmol), *t*-BuOH (2.0 mL) at room temperature for 10 min. Subsequently aryl amidine 3 (1.1 mmol) was added into the mixture and then capped. The mixture was irradiated for a given time at 140°C. (Scheme 68).



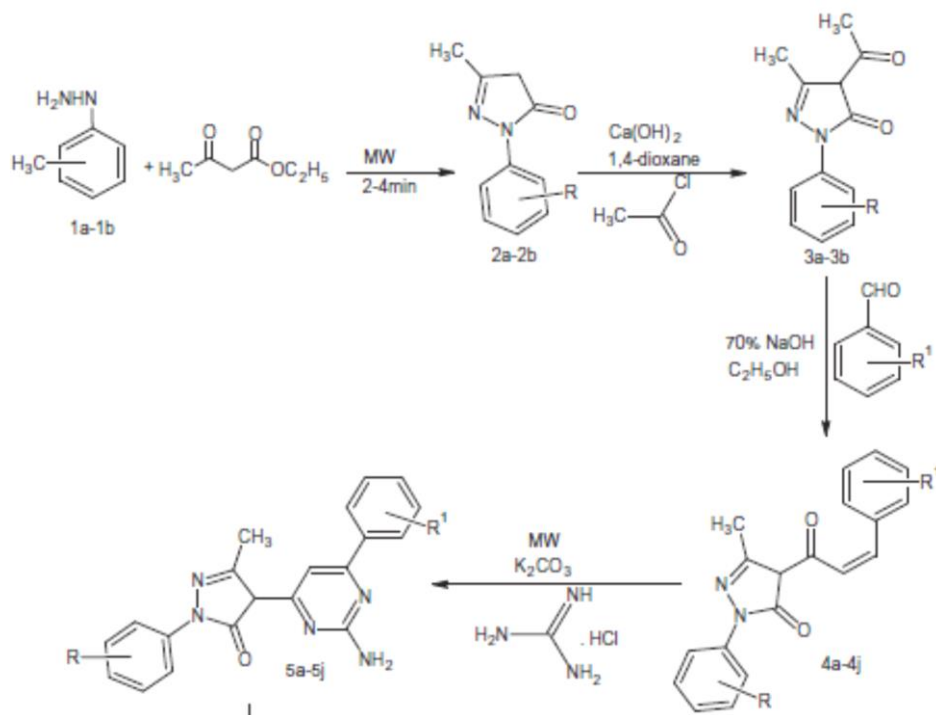
Scheme 68: Synthetic route for thiopyrano-, pyrano[4,3-d]pyrimidine derivatives

Loidreau Yvonnick et.al [88] reported the synthesis of *N*-arylpyrido[2',3':4,5]furo[3,2-d]pyrimidin-4-amines (1a-i) by reaction of a mixture of *N*'-(2-cyanofuro[3,2-b]pyridin-3-yl)-*N,N*-dimethyl formimidamide 8 (0.1 g, 0.47 mmol) and appropriate aniline (1.0 equiv) in acetic acid (2 mL) and irradiating at 118°C (400 W, 3 min ramp) (Scheme 69).



Scheme 69: Synthetic route for N-arylpyrido[2',3':4,5]furo[3,2-d]pyrimidin-4-amines (1a-i)

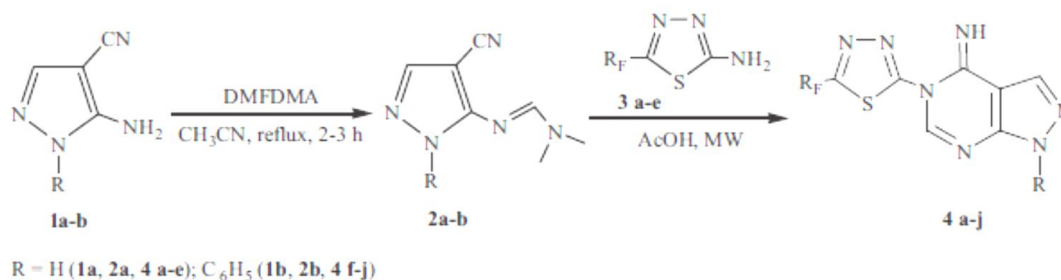
Antre Rishikesh V. et.al [89] reported the synthesis of 4-(2-amino-6-(substituted)pyrimidin-4-yl)-3-methyl-1-(substituted)-1H-pyrazol-5(4H)-one derivatives (5a-5j) by reaction of mixture of the chalcone derivatives (4a-4j) (0.001 mol) and guanidine hydrochloride (0.002 mol) in acetone (5 mL) and ethanol (5 mL), K<sub>2</sub>CO<sub>3</sub> and stirring vigorously. After 5 min, solvent was evaporated and the dry powder was irradiated in microwave oven for the appropriate time in between 6-9 min at 50% power output (Scheme 70). The compounds were found to exhibit anti-inflammatory, analgesic and antipyretic activities.



Scheme 70: Synthetic route for 4-(2-amino-6-(substituted)pyrimidin-4-yl)-3-methyl-1-(substituted)-1H-pyrazol-5(4H)-one derivatives (5a-5j)

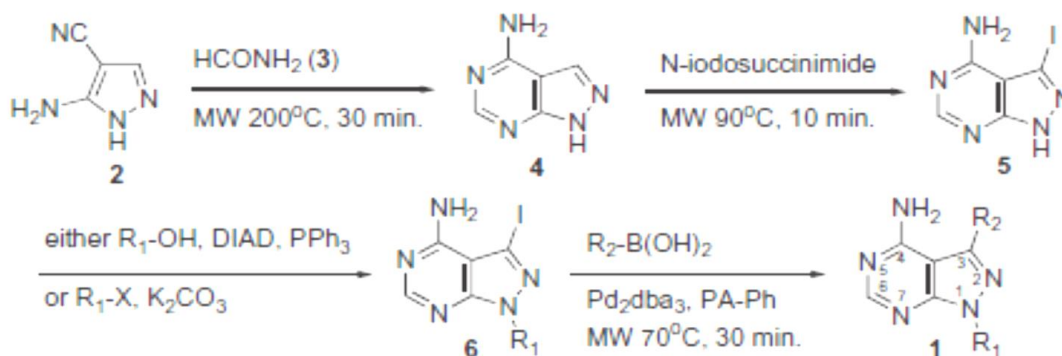
Song Xin Jian et.al [90] reported the synthesis of novel fluorinated pyrazolo[3,4-d]pyrimidine derivatives containing 1,3,4-thiadiazole. The synthesis of 5-(5-substituted-1,3,4-thiadiazol-2-yl)-1H-pyrazolo[3,4-d]pyrimidin-4(5H)-imine (4a-e) was done by reaction of the solution of 1H-4-cyano-5-[(N,N-dimethylaminomethylene)amino]pyrazole 2a (0.33 g, 2 mmol) and the appropriate aminothiadiazoles 3 (2 mmol) in 6 mL glacial acetic

acid and irradiating at 140°C for 18 min by microwave (Scheme 71). Their antitumor activities were evaluated against HL-60 by an MTT assay. The preliminary results indicated that some title compounds exhibit more potent antitumor inhibitory activity than doxorubicin (DOX).

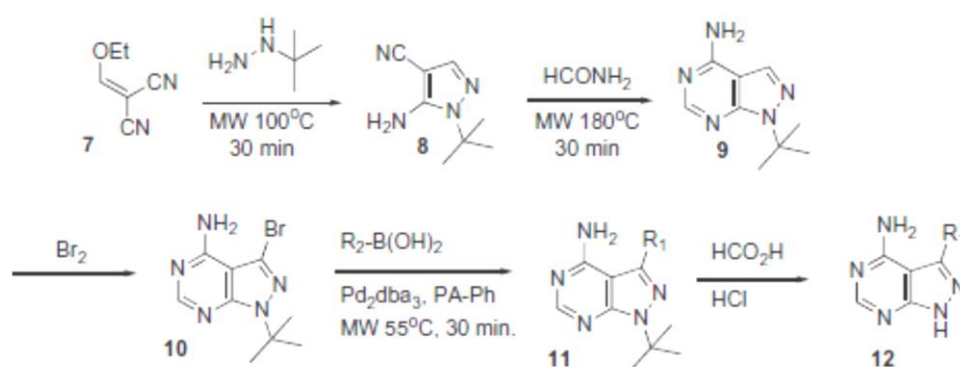


Scheme 71: Synthetic route for fluorinated pyrazolo[3,4-d]pyrimidine derivatives containing 1,3,4-thiadiazole

Todorovic Nick et.al [91] reported synthesis of a library N1- and C3-substituted-pyrazolo[3,4-d]pyrimidines. The microwave-assisted approach involved the de novo generation of the heterocyclic scaffold, facile alkylation at N1 via either a Mitsunobu or a direct alkylation reaction and arylation at C<sub>3</sub> via a Suzuki reaction (Scheme 72a) (Scheme 72b).



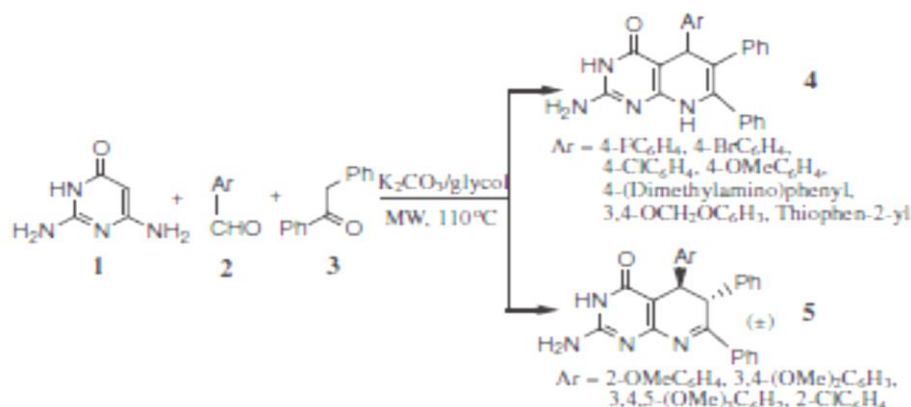
Scheme 72a: Synthetic route to 1-alkyl-3-aryl-1H-pyrazolo[3,4-d]pyrimidin-4-amines



Scheme 72b: Synthetic route to 3-aryl-1H-pyrazolo[3,4-d]pyrimidin-4-amines(12)

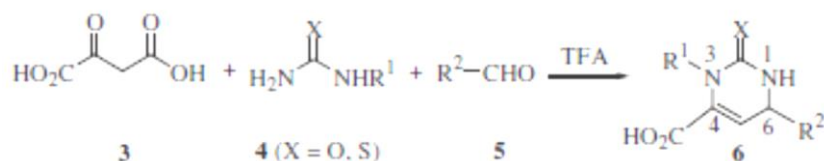
Shi Feng et.al[92] reported the synthesis of novel 5,6,7-triarylpyrido[2,3-d]pyrimidin-4-one derivatives by reaction of 2,6-diaminopyrimidin-4(3H)-one 1 (1 mmol), aromatic aldehydes 2 (1 mmol) and 1,2-diphenylethanone 3 (1 mmol) in 2mL of glycol with the presence of K<sub>2</sub>CO<sub>3</sub> (0.1 mmol) at 110°C under microwave irradiation (MW) at the initial/maximum power of 100 W/250 W (Scheme 73). The novel compounds were subjected to the test of in

vitro cytotoxicity to carcinoma SW1116 and SGC7901 cells. Most of the tested compounds showed significant cytotoxicity to SW1116 cells.



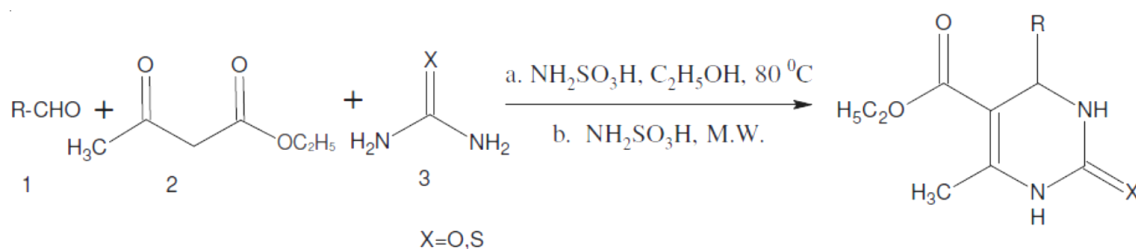
Scheme 73: Chemoselective synthesis of 5,6,7-triarylpyrido[2,3-d]pyrimidin-4-one derivatives 4 and 5.

Fang Zhanxiong et.al [93] reported the synthesis of 5-unsubstituted 3,4-dihydropyrimidin-2-ones and thiones by reaction of a mixture of oxalacetic acid (2.6 mmol), aldehyde (2.0 mmol), and urea/thiourea (2.6 mmol) in THF (3 mL) was added TFA (0.10 mL). The mixture was heated at 95°C for 15 min using microwave irradiation in a sealed tube. (Scheme 74)



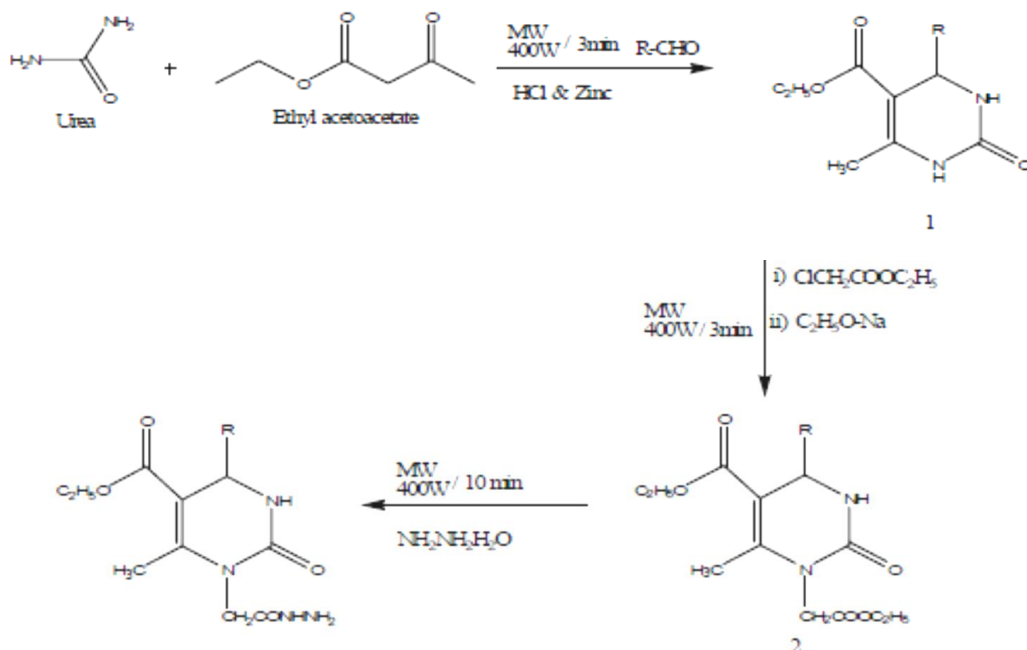
Scheme 74: Synthetic route for 5-unsubstituted 3,4-dihydropyrimidin-2-ones and thiones

Pai Nandini et.al [94] reported the synthesis of 3,4-dihydropyrimidinones (DHPMS) using sulfamic acid as catalyst from an aldehyde (1mmol), ethyl acetoacetate (1 mmol), urea or thiourea (1.5 mmol) and sulfamic acid (20 mol %) in a beaker covered with watch glass and irradiating at 300 watt. (Scheme 75)



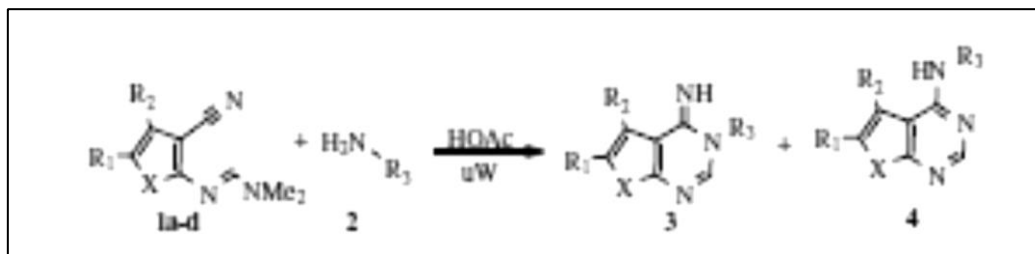
Scheme 75: Synthetic route for 3,4-dihydropyrimidinones (DHPMS)

Singh Niharika I et.al [95] reported the synthesis of Ethyl 1-(2-Hydrazino-2-Oxoethyl)-6-Methyl-2-oxo-4-Phenyl-1, 2, 3, 4-Tetrahydropyrimidine-5-Carboxylate by reaction of a mixture of the appropriate compound (1) (0.1mol), hydrazine hydrate (8 drops), and 10ml of 95% ethanol and refluxing under microwave for 3min at 840 power (Scheme 76).



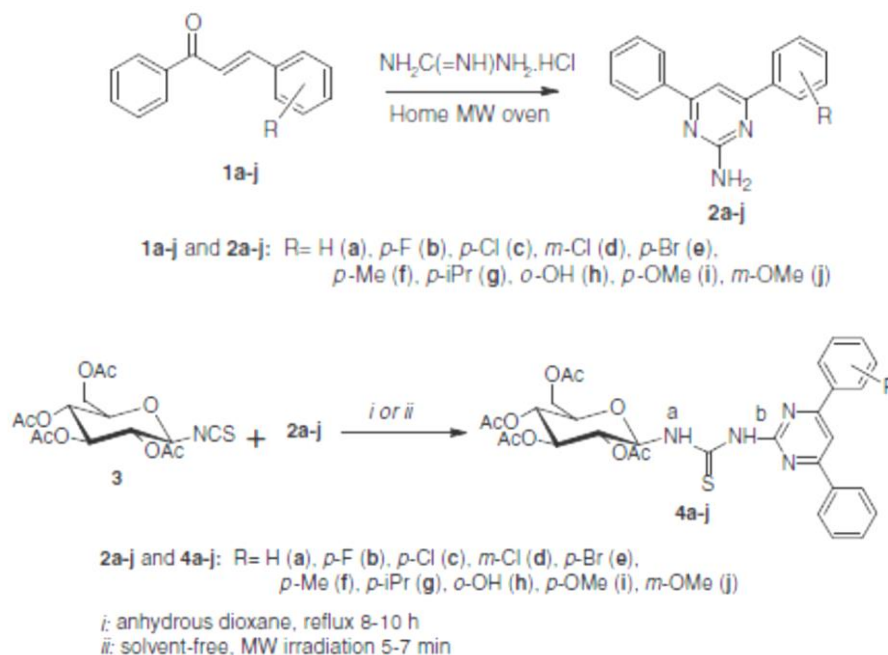
Scheme 76: Synthetic route for Ethyl 1-(2-Hydrazino-2-Oxoethyl)-6-Methyl-2-oxo-4-Phenyl-1, 2, 3, 4-Tetrahydropyrimidine-5-Carboxylate

Han Ying et.al [96] reported the synthesis of furo- and thieno[2,3-d]pyrimidin-4-amine derivatives by microwave irradiation from readily available amines and substituted 2-aminofuran-3-carbonitrile or 2-aminothiophene-3-carbonitrile (6.2 mmol) and converting into corresponding formamidines (1a) in DMF using benzenesulfonyl chloride (12 mmol). Then in the next step acetic acid (0.8 mL), formimidamide 1a (0.2 mmol), and benzylamine (0.3 mmol) 2a was irradiated in microwave at 180°C for 35 min. (Scheme 77).



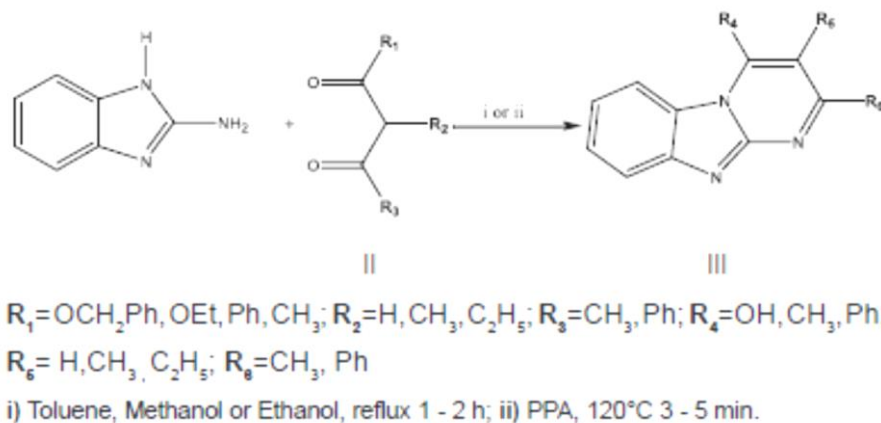
Scheme 77: Synthetic route for furo- and thieno[2,3-d]pyrimidin-4-amine derivatives

Thanh Nguyen Dinh et.al [97] reported the synthesis of N-(2,3,4,6-tetra-O-acetyl-b-D-glucopyranosyl)-N0-(40,60-diarylpyrimidin-20-yl)thioureas (4a-j) by grinding of a mixture of 2-amino-4,6-diarylpyrimidine 2 (2 mmol) and 2,3,4,6-tetra-O-acetyl-b-D-glucopyranosyl isothiocyanate 3 (2 mmol) in a porcelain beaker. Then the mixture was put into a domestic microwave oven (the power output is 750 W). The adjustor of the microwave oven was set to the proper temperature (about 50°C) (Scheme 78). The compounds were found to exhibit antibacterial and antifungal activities against *Staphylococcus epidermidis*, *Enterobacter aerogenes* and *Candida albicans*.



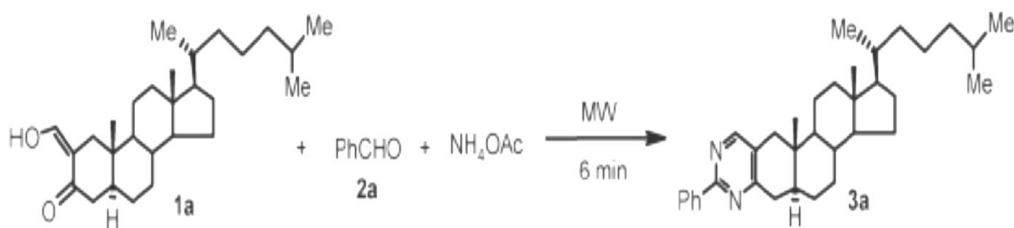
Scheme 78: Synthetic pathway for N-(2,3,4,6-tetra-O-acetyl-b-D-glucopyranosyl)-N0-(4,6-diarylpyrimidin-2-yl)thioureas (4a-j)

Algul Oztekin et.al [98] reported the synthesis of 2,4-di- and 2,3,4-trisubstituted benzimidazo[1,2-a]pyrimidine compounds (IIIa-g) by reaction of 2-Aminobenzimidazole (2.5 mmol), dicarbonyl compounds (2.5 mmol) and 5 g PPA in a microwave oven and irradiating at low power for 3 - 5 min. The compounds were found to exhibit antimicrobial activities. (Scheme 79)



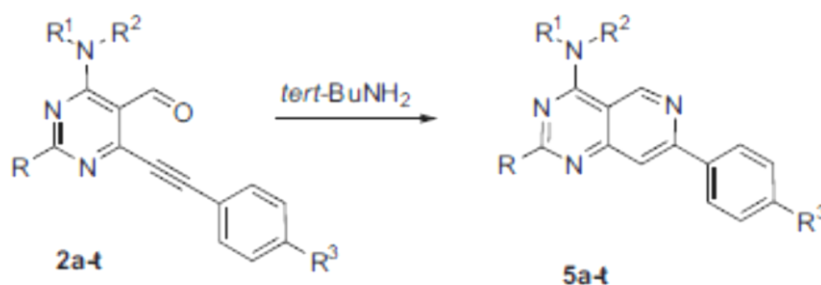
Scheme 79: Synthetic route for 2,4-di- and 2,3,4-trisubstituted benzimidazo[1,2-a]pyrimidine compounds (III)

Barthakur Madan G. et.al [99] reported the synthesis of 2'-Aryl-steroidal[3,2-d]pyrimidine by intimately mixing of 2-Hydroxymethylene-3-ketosteroid (1a, 1mmol), aromatic aldehyde (2a, 2mmol) and ammonium acetate (2mmol) with silica gel (60–120mesh, 2.0 g) in a mortar and irradiating the mixture in microwave reactor for 6min after setting reaction temperature at 120°C and power at 60% (maximum output 300W) (Scheme 80).



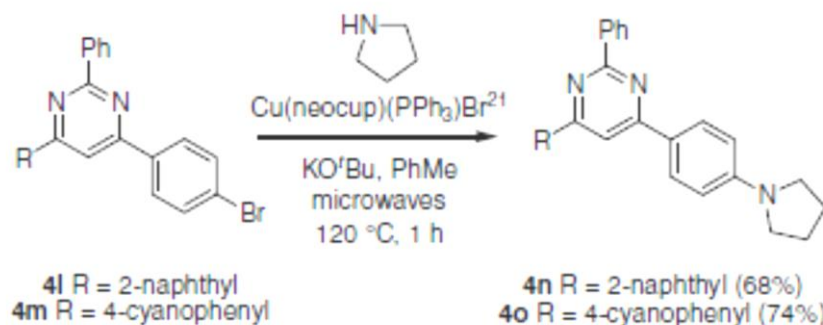
Scheme 80: Synthesis of steroidal pyrimidines 3a from 2-hydroxymethylene-3-ketocholestan 1a.

Cikotiene Inga et.al [100] reported the synthesis of 2,4- disubstituted 7-arylpyrido[4,3-d]pyrimidines 5a-t by reaction of a solution of the corresponding 2,4-disubstituted 6-arylethynylpyrimidine-5-carbaldehydes 2a-t (0.2 mmol) and tert-butylamine (0.15 g, 2 mmol) in DMF (3 mL) and irradiating in a microwave oven at 500W for 10 min to one hour. (Scheme 81)



Scheme 81: Synthetic route for 2,4- disubstituted 7-arylpyrido[4,3-d]pyrimidines 5a-t

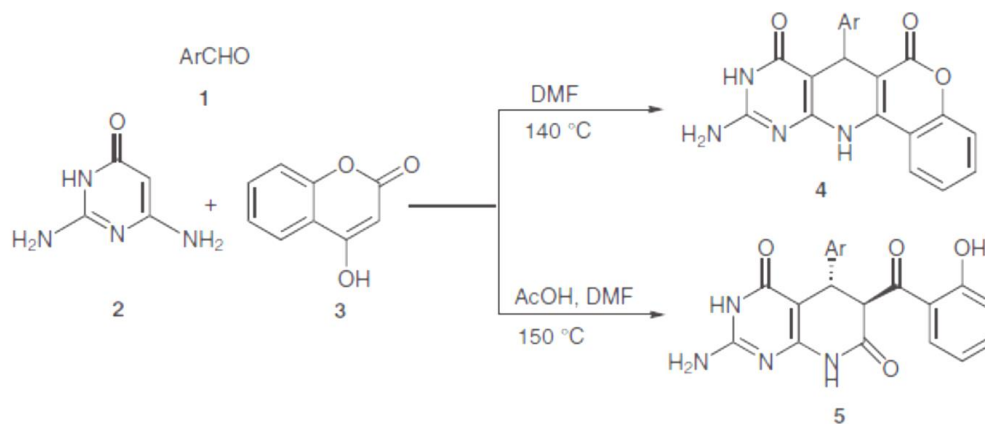
Bagley Mark C. et.al [101] reported the synthesis of 4-{2-phenyl-6-[4-(pyrrolidin-1-yl)phenyl]pyrimidin-4-yl}benzonitril by adding a solution of benzonitrile 4m (0.57 mmol, 1 equiv) in toluene (3 mL) to a stirred solution of pyrrolidine (1.13 mmol, 2 equiv), Cu(neocup)(PPh<sub>3</sub>)Br<sup>21</sup> (10 mol %) and potassium tert-butoxide (0.85 mmol, 1.5 equiv) in toluene (3 mL) and irradiating at 120°C, at an initial power of 150 W for 1 h (Scheme 82).



Scheme 82: Synthetic route for 4-{2-phenyl-6-[4-(pyrrolidin-1-yl)phenyl]pyrimidin-4-yl}benzonitril

Tus, et.al [102] reported the synthesis of 2-Amino-5-aryl-6-(2-hydroxybenzoyl)-5,8-dihydropyrido[2,3-d]pyrimidine 4,7 (3H,6H)-dione 5; by reaction of aldehyde 1 (1 mmol), 2,6-diaminopyrimidin-4(3H)-one (2, 126 mg, 1 mmol), 4-hydroxy-2H-chromen-2-one (3, 162 mg, 1 mmol), AcOH (0.4 mL), and DMF (1.6 mL) and irradiating for 5 min at 100 W (initial power) and 200 W (maximum power) and 150 °C. (Scheme 83)





Scheme 83: Synthetic route for 2-Amino-5-aryl-6-(2-hydroxybenzoyl)-5,8-dihydropyrido[2,3-d]pyrimidine 4,7 (3H,6H)-dione 5

### Conclusion:

As described in this paper, pyrimidine derivatives are very important, effective and valuable compounds in medicine and pharmaceutical industries. Therefore, synthesis of these compounds has received much attention among the organic researchers. During recent years, to synthesize pyrimidine derivatives, prior attention is given to non conventional methods as compared to traditional protocol since in traditional protocols in organic synthesis, unfortunately, most of them have main disadvantage such as unsatisfactory yields, formation of side product, harsh reaction conditions, use of strong bases, and environmentally unfriendly due to use of an hazardous and toxic organic solvent as reaction medium. Green chemistry promotes a modern idea for accomplishing chemical research and productions. With enhancing public concerns on environmental protection, chemistry science researchers are asked to maximize the advantages of chemical products and minimize the side effects that could be harmful to the environment and humans. Organic reactions conducting under nontraditional conditions (green synthetic strategies) are obtaining in popularity, primarily to decrease environmental concerns. The advantages of green synthetic strategies facilitated (assisted) organic transformations, namely the selectivity, simplicity of experimental manipulation, and increased reaction rates, were highlighted. The advantage of these strategies is obvious; it is not only useful for the world economy, but also acts toward making the world more environmentally friendly.

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